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Selective Sequential Intramolecular Cyclization of Ethenetricarboxylates with Arylpropenamines

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Preface

This thesis deals with the studies conducted during April 2016 to March 2019 under the guidance of Professor Akiya Ogawa at the Department of Applied Chemistry, Graduate School of Engineering, Osaka Prefecture University.

This thesis is concerned with the studies on the highly selective intramolecular cyclization reaction of ethenetricarboxylate and other electron-deficient carboxylic acids with arylpropenamines. One of the topics is the control of the chemo- and stereoselectivity of the intramolecular cyclization reaction by phenylpropenamines bearing substituents on the benzene ring. The other is stereoselective sequential intramolecular cyclization reactions of ethenetricarboxylates with heteroarylpropenamines. In the course of the research, novel chemoselective intramolecular cyclization reactions of β -substituted cinnamylamines are also studied.

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Contents

Chapter 1	General Introduction	p. 1
Chapter 2	Sequential Intramolecular Diels-Alder Reaction of Furyl Ethenetricarboxylate	amines with p. 7
Chapter 3	Intramolecular [2 + 2] and [4 + 2] Cycloaddition Reactions of Cir of Ethenetricarboxylate in Sequential Processes	mamylamides
Chapter 4	Sequential Intramolecular Cyclization of Diarylpropenamines v Deficient Carboxylic Acids	vith Electron-
Chapter 5	Sequential Intramolecular Diels-Alder Reaction of 3- propenylamides of Ethenetricarboxylate	-Heteroaryl-2- p. 91
Chapter 6	Intramolecular Cyclization Reactions of β -Substituted Cinnar Ethenetricarboxylate	nylamides of p. 115
Chapter 7	Conclusion	p. 130
List of Publicat	ions	p. 132
Acknowledgem	ent	р. 133

Chapter 1 General Introduction

Multicyclic compounds are present in a large number of physiologically active substances and functional materials. Among them, multicyclic heterocycles containing nitrogen and oxygen often have high functionality, and it is important to develop efficient synthesis methods of the heterocycles.¹ Their preparation usually requires lengthy steps. It is desirable to construct such multicyclic compounds rapidly with high efficiency.

Cycloaddition reactions form two bonds in cyclic systems. Intramolecular reactions access fused and bridged ring systems.² Intramolecular cycloaddition reactions of styrenes may form $[2+2]^3$ and $[4+2]^4$ cycloadducts.

However, [4 + 2] cycloaddition (Diels-Alder) reaction of the styrene as a diene, requires relatively high temperature, because it involves dearomatization of the benzene ring.⁵ Intramolecular Diels-Alder (IMDA) reaction of vinylfuran as a diene has been reported. But it generally requires higher temperatures than that of a furan ring as a diene⁶, and there are fewer examples.

Yamazaki's group reported that Lewis acid (MX_n) -promoted cyclization/halogenation of alkenyl ethenetricarboxylates gives 3,4-*trans* five-membred rings stereoselectively with high generality ((**a**) in Scheme 1-1).⁷ 2-Alkenyl amides of ethenetricarboxylates also undergo facile intramolecular ene reactions ((**b**) in Scheme 1-1).^{7c} Ethenetricarboxylate derivative has high electrophilicity at the alkene site by three carbonyl groups.⁸ The utility of ethentricarboxylates has been shown for various inter- and intramolecular reactions, for example, leading to cyclic compounds. It is of interest to examine the reaction of highly electrophilic ethenetricarboxylates bearing arylvinyl groups.

In this thesis, sequential amide formation/cyclization reactions of ethenetricarboxylate and other electron-deficient alkenic carboxylate, such as fumarate have been investigated. The selectivity of the reactions has been discussed.



Scheme 1-1. Cyclization reaction of ethenetricarboxylate derivatives.⁷

This thesis is divided into seven chapters. The introduction is presented in chapter 1.

In chapter 2, IMDA reaction of 2-furylmethylamides of ethenetricarboxylate in sequential process is described. Reaction of 1,1-diethyl 2-hydrogen ethenetricarboxylate and 2-furylmethylamines in the presence of EDCI/HOBt/Et₃N at room temperature led directly to IMDA adducts.



Scheme 1-2. Chapter 2.

In chapter 3, intramolecular [2 + 2] and [4 + 2] cycloaddition reactions of cinnamylamides of ethenetricarboxylate in sequential processes are described. Reaction of 1,1-diethyl 2-hydrogen ethenetricarboxylate and *trans*-cinnamylamines in the presence of

EDCI/HOBt/Et₃N led to pyrrolidine products in one pot via intramolecular [2 + 2] and [4 + 2] cycloaddition reactions. The types of the products depend on the substituents on the benzene ring and the reaction conditions. Reaction of cinnamylamines without substituents on the benzene ring and with halogens and OMe on the *para* position at room temperature gave cyclobutane-fused pyrrolidines as the major products via [2 + 2] cycloaddition. On the other hand, reaction of 1,1-diethyl 2-hydrogen ethenetricarboxylate and cinnamylamines bearing electron-withdrawing groups such as NO₂, CN, CO₂Me, CO₂Et, and CF₃ on *ortho* and *para* positions in the presence of EDCI/HOBt/Et₃N at room temperature or at 60–80 °C gave tetrahydrobenz[*f*]isoindolines via [4 + 2] cycloaddition (IMDA) as the major products.



Scheme 1-3. Chapter 3.

In chapter 4, IMDA reactions of various 3,3-diarylpropenylamides of electron-deficient alkenes to give hexahydrobenzo[f]isoindoles were investigated. Reaction of ethenetricarboxylate with 3,3-diaryl-2-propen-1-amines under the amide formation conditions gave the tricyclic compounds in sequential processes involving IMDA reaction. The reaction gave cis- and trans-fused tricyclic compounds selectively, depending on the substituents on the benzene ring, reaction temperature and solvent. the reaction with 1,1-diethyl 2-hydrogen In ethenetricarboxylate substituted by 3,3-diaryl-2-propen-1-amines, trans-substituted aryl group reacted mainly as a styrene component. Amides of electron-deficient alkenic carboxylic acids such as fumarate do not undergo cyclization at room temperature sequentially and the reaction on heating gave *trans*-fused hexahydrobenzo[*f*]isoindoles.



Scheme 1-4. Chapter 4.

In chapter 5, the stereoselectivity in the reaction of ethenetricarboxylate with heteroarylpropenylamines was investigated. Reaction of 1,1-diethyl 2-hydrogen ethenetricarboxylate with (E)-3-(2-furyl)-2-propenylamines in the presence of EDCl/HOBt/Et₃N at 80-110 °C gave *cis*-fused tricyclic compounds as the major products. On the other hand, reaction with (E)-3-(3-furyl)-2-propenylamines at 80-110 °C gave *trans*-fused tricyclic compounds as the major products. The reaction with *E*-3- and 4-pyridinyl-2-propenylamines was also carried out. The reaction with 3-pyridinyl propenylamine gave HOBt-incroporated pyrrolidine diastereoselectively, and the reaction with 4-pyridinyl-2-propenylamines gave a complex mixture.



Scheme 1-5. Chapter 5.

In chapter 6, reaction of β -substituted cinnnamylamines of ethenetricarboxylates was examined. Reaction of ethenetricarboxylate with (*E*)-3-aryl-2-buten-1-amine and (*E*)-3-aryl-3-bromo-2-propen-1-amine under the amide formation conditions gave cyclized products with chemo- and stereoselectivites, similar to that with cinnamylamines in chapter 3 (Scheme 1-6).

Finally, the summary of this thesis is presented in chapter 7.



Scheme 1-6. Chapter 6.

References

¹ K. Speck, T. Magauer, Beilstein J. Org. Chem. 2013, 9, 2048.

² (a)W. Oppolzer, Angew. Chem., Int. Ed. Engl. 1997 16, 10. (b) S. E. Wolkenberg, D. L. Boger, J.

Org. Chem. 2002, 67, 7361. (c) Q. Wang, A. Padwa, Org. Lett. 2004, 6, 2189. (d) P. S. Sarang, A.

A. Yadav, P. S. Patil, U. M. Krishna, G. K. Trivedi, M. M. Salunkhe, Synthesis 2007, 1091. (e) L.

B. Nielsen, R. Slamet, D. Wege, Tetrahedron 2009, 65, 4569.

³ (a) T. Bach, C. Pelkmann, K. Harms, *Tetrahedron Lett.* **1999**, 40, 2103. (b) S. Nakano, K. Kakugawa, T. Nemoto, Y. Hamada, *Adv. Synth. Catal.* **2014**, 356, 2088.

⁴ (a) E. Ciganek, Org. React. 1984, 32, 1. (b) W. R. Roush, In Comprehensive Organic Synthesis;
B. M. Trost, I. Fleming, Pergamon. Eds, 1991, 5, 513. (c) C. O. Kappe, S. S. Murphree, A. Padwa, Tetrahedron 1997, 53, 14179.

⁵ (a) L. H. Klemm, T. M. McGuire, K. W. Gopinath, *J. Org. Chem.* **1976**, 41, 2571. (b) L. S. Kocsis, E. Benedetti, K. M. Brummond, *Org. Lett.* **2012**, 14, 4430. (c) E. Benedetti, A. B. E. Veliz, M. Charpenay, L. S. Kocsis, K. M. Brummond, *Org. Lett.* **2013**, 15, 2578. (d) T. Ozawa, T.

Kurahashi, S. Matsubara, Org. Lett. 2011, 13, 5390. (e) S. Sun, I. J. Turchi, D. Xu, W. V. Murray, J. Org. Chem. 2000, 65, 2555. (f) R. Pedrosa, C. Andrés, J. Nieto, J. Org. Chem. 2002, 67, 782.
⁶ (a) I. H. Yuriy, Z. L. Roman, V. H. Yuriy, P. Z. Vladimir, F. M. Dmitriy, N. B. Maria, V. N. Eugenia, L. Tadeusz, K. Vasyl, S. M. Vasyl, I. Z. Fedor, V. V. Alexey, D. O. Mykola, Tetrahedron Lett. 2015, 56, 4499 (b) H. Kotsuki, A. Kawamura, M. Ochi, T. Tokoroyama, Chem. Lett. 1981, 917. (c) K. Fischer, S. Hünig, Chem. Ber. 1987, 120, 325, (d) J. A. Cooper, P. Cornwall, C. P. Dell, D. W. Knight, Tetrahedron Lett. 1988, 29, 2107. (e) P. Cornwall, C. P. Dell, D. W. Knight, J. Chem. Soc., Perkin Trans. 1. 1993, 2395. (f) M. G. B. Drew, A. Jahans, L. M. Harwood, S. A. B. H. Apoux, Eur. J. Org. Chem. 2002, 3589. (g) R. E. Patre, S. Gawas, S. Sen, P. S. Parameswaran, S. G. Tilve, Tetrahedron Lett. 2007, 48, 3517.

- ⁷ (a) S. Yamazaki, K. Fujinami, Y. Maitoko, K. Ueda, K. Kakiuchi, J. Org. Chem. 2013, 78, 8405.
- (b) Y. Fukushima, S. Yamazaki, A. Ogawa, Org. Biomol. Chem. 2014, 12, 3964. (c) S. Yamazaki,
- J. Wada, K. Kakiuchi, Can. J. Chem. 2015, 93, 1122.
- ⁸ S.Yamazaki, J. Synth. Org. Chem. Jpn. 2014, 72, 6, 666.

Chapter 2

Sequential Intramolecular Diels-Alder Reaction of Furylamines with Ethenetricarboxylate

2-1 Introduction

Intramolecular Diels-Alder (IMDA) reaction is one of the most widely used synthetic tools for natural products.¹

The IMDA reaction of furans as diene is used for facile formation of multicyclic skeletons. ² In normal electron-demand Diels-Alder reaction, the alkene component (dienophile) is usually electron-deficient. In general, the greater number of electron-withdrawing substituents on the double bond, results in higher reactivity of the dienophile, owing to the lowering of the energy of the LUMO of the dienophile by the substituents. Ethenetricarboxylate derivatives bearing three carbonyl groups have been employed as highly electrophilic C=C components in various bond-forming reactions.³ Although an intramolecular inverse electron demand hetero Diels-Alder reaction of 1-allylic 2,2-dimethyl esters of ethene-1,2,2-tricarboxylate has been studied,⁴ only a few normal electron demand Diels-Alder reactions of ethenetricarboxylates allow for the facile derivatization at 2-carboxyl group. The electron-deficient alkene moiety is expected to work as a reactive dienophile in the IMDA reaction.

In this chapter, sequential IMDA reaction of ethenetricarboxylate derivatives with furan as diene has been studied.

2-2 Results and Discussion

IMDA reaction of ethenetricarboxylates has been examined. Reaction of *N*-allyl- or *N*-benzyl-2-furylmethylamine **2a**,**c** and 1,1-diethyl 2-hydrogen ethenetricarboxylate **1** in the presence of EDCI/HOBt/Et₃N at room temperature led directly to an IMDA adducts **3a**,**c** in 65-82% yields (Table 2-1).⁶ The possible intermediate **4** could not be observed under the reaction conditions of amide formation (Scheme 2-1). The reaction of **1** and **2a** in the absence of condensation reagents only gave the mixture of **1** and **2a**, probably forming a salt. Treatment of the mixture with condensation reagents led to the Diels-Alder adduct **3a** in 75% yield.

		EtO ₂ C CO ₂ E	t + R ²	0 N-F H 2	$R^{1} \xrightarrow{\text{HOBt}^{c} \text{EtO}_{2C}}_{\text{EtO}_{3}N} \xrightarrow{\text{EtO}_{2C}}_{O_{4}}$	R^{2} R^{2} R^{2} R^{3} N_{3} R^{2} R^{3} R	
Entry	2	\mathbb{R}^1	\mathbb{R}^2	Temp, Time	Solvent	Product	Yield (%)
1 ^a	2a	CH ₂ CH=CH ₂	Η	r.t., 17 h	THF	3a	82
2 ^a	2b	CHMePh	Н	r.t., 20 h	THF	3 b	43(dr=1:1)
3 ^a	2c	CH ₂ Ph	Н	r.t., 21 h	THF	3c	65
4	2c	CH ₂ Ph	Н	60 °C, 20 h	THF	3c	46
5	2c	CH ₂ Ph	Н	80 °C, 20 h	$CH_2ClCH_2Cl^b\\$	3c	50
6	2d	CH ₂ Ph	Br	r.t., 20 h	THF	3d	48
7	2d	CH ₂ Ph	Br	r.t., 1 h	THF	3d	40
8	2d	CH ₂ Ph	Br	60 °C, 20 h	THF	3d	55
9	2d	CH ₂ Ph	Br	80 °C, 20 h	$CH_2ClCH_2Cl^b$	3d	75

 Table 2-1. Reactions of 1,1-diethyl 2-hydrogen ethenetricarboxylate 1 and 2-furylmethylamines 2.

^a Results of entries 1-3 are cited from reference 6. ^b The byproducts were removed by column chromatography.⁷ ^c HOBt: 1-hydroxybenzotriazole. ^d EDCI: 1-[3-(dimethyl-amino)propyl]-3-ethylcarbodiimide hydrochlorid.



Scheme 2-1. Formation of IMDA product 3.

The structure of 10-oxa-3-aza-tricyclo[$5.2.1.0^{1.5}$]dec-8-ene **3a** was determined by X-ray analysis (Figure 2-1). The exo stereochemistry of Diels-Alder adducts **3a-d** with respect to the amide group was also determined by NOEs. NOEs between C5-*H* and C2-H*H* and/or between C5-*H* and C9-*H* were observed (atom numbering is shown in Table 2-1).



Figure 2-1. ORTEP drawing of **3a** (thermal ellipsoids are drawn at 50% probability). The atom numbering is different from that in Table 2-1.

In order to explain the stereoselectivity of the IMDA reaction was examined by B3LYP/6-31G* calculations including the PCM solvent effect (solvent=THF).

The endo and exo IMDA reactions from a model compound **4m** as a possible intermediate were calculated (Figure 2-2). The activation energy ΔG^{\ddagger} of endo TS (31.17 kcal/mol) is much higher than that of exo TS (21.41 kcal/mol).



Figure 2-2. RB3LYP/6-31G* SCRF = (PCM, solvent=THF) optimized structures for endo:exo IMDA reaction paths of **4m**.

The acid-catalyzed IMDA reactions of **4m** were also calculated (Figure 2-3). The protonated six-membered ring intermediates with hydrogen bonding were assumed.⁸ The acid *in situ*, possibly generating from EDCI (1-[3-(dimethylamino)propyl]-3-ethylcarbodiimide

hydrochloride) or starting material **1** may catalyze the cycloaddition reactions. Stepwise mechanism with zwitter-ionic intermediates was obtained for the H⁺-catalyzed reaction. The acid-catalyzed reaction lowers the activation energies compared to the uncatalyzed reaction. The activation energy ΔG^{\ddagger} of the second bond formation TS (H⁺IM-exo-TS2) (10.10 kcal/mol) is higher than that of the first TS (H⁺IM-exo-TS1) (1.51 kcal/mol) for exo addition. The H⁺-catalyzed process accelerates the formation of the exo adduct **3**.

The first bond formation TS (H⁺IM-endo-TS1) for endo addition was obtained; however the second bond formation TS could not be obtained. Optimization of the initial structure of endo**4m-H**⁺ led to the intermediate (H⁺IM- endo-Int). The endo TS and product are highly unstable probably because of the steric constraint. Therefore, the exo adducts **3** are produced stereoselectively.

Reaction of 1-phenylethyl-2-furylmethylamine or **2b** gave Diels-Alder adducts **3b** with diastereomer ratios of 1:1 in 43% yield. Reaction of *N*-benzyl-(5-bromofuran-2-yl)methylamine **2d** gave Diels-Alder adduct **3d** at room temperature in THF for 1 h in 40%, for 20 h in 48% and at 80 °C in CH₂ClCH₂Cl⁷ for 20 h in 75%. A small amount of byproducts formed at room temperature possibly contain amine adducts at C=C bond. The reaction of **1** and **2c** with DCC gave a complex mixture containing a small amount of DCU-incorporated byproducts.⁹



Figure 2-3. RB3LYP/6-31G* SCRF = (PCM, solvent=THF) optimized structures for endo:exo acid-catalyzed IMDA reaction paths of **4m**. Values in parentheses are Mulliken charges with hydrogens summed into heavy atoms.

2-3 Conclusion

In summary, IMDA reaction of ethenetricarboxylate and furylamines has been studied. Reaction of benzyl- or allyl-2-furylmethylamine and 1,1-diethyl 2-hydrogen ethenetricarboxylate in the presence of EDCI/HOBt/Et₃N at room temperature led directly to IMDA adducts stereoselectively. The highly functionalized cyclic compounds obtained in this chapter should be useful synthetic intermediates.

2-4 Experimental Section

General Procedures

¹H Chemical shifts are reported in ppm relative to Me₄Si. ¹³C Chemical shifts are reported in ppm relative to CDCl₃ (77.1 ppm). ¹³C mutiplicities were determined by DEPT and HSQC. Peak assignments are made by 2D COSY, HSQC, NOESY, and HMBC spectra. Mass analyzer type used for EI is double-focusing in the HRMS measurements. Column chromatography was performed on silica gel (75-150 μm).

N-Benzyl-furfurylamine (2c): (8.9 mmol scale, 1.62 g, 97%). 2c was also prepared by reaction of furfurylamine (2 equiv) with benzyl bromide in Et₂O (5.0 mmol scale, 449 mg, 47%) according to the literature procedure.¹⁰

2c: R_f = 0.2 (hexane-Et₂O = 1 : 1); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.85 (bs, 1H), 3.78 (s, 4H), 6.18 (bd, *J* = 3.3 Hz, 1H), 6.31 (dd, *J* = 3.3, 1.8 Hz, 1H), 7.22-7.28 (m, 1H), 7.29-7.34 (m, 4H), 7.36 (dd, *J* = 1.8, 0.7 Hz, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 45.37 (CH₂), 52.81 (CH2), 107.12 (CH), 110.15 (CH), 127.08 (CH), 128.32 (CH), 128.46 (CH), 139.86 (C), 141.88 (CH), 153.80 (C).

N-Benzyl-(5-bromofuran-2-yl)methylamine (2d): (9.0 mmol scale, 2.14 g, 89%): $R_f = 0.3$ (hexane-Et₂O = 1 : 1); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.75 (bs, 1H), 3.71 (s, 2H), 3.75 (s, 2H), 6.13 (d, J = 3.2 Hz, 1H), 6.20 (d, J = 3.2 Hz, 1H), 7.21-7.26 (m, 1H),

7.28-7.33 (m, 4H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 45.28 (CH₂), 52.60 (CH₂), 109.86 (CH), 111.72 (CH), 120.57 (C), 127.06 (CH), 128.20 (CH), 128.41 (CH), 139.62 (C), 155.94 (C); IR (neat) 3329, 3027, 2917, 2830, 1602, 1505, 1453, 1200, 1125, 1010 cm⁻¹; MS (EI) *m/z* 267 (M⁺, 12), 265 (M⁺, 12), 186 (11), 161 (21), 159 (22), 106 (23), 91 (100%); HRMS (EI) M⁺ 265.0099, 267.0045 (calcd for C₁₂H₁₂BrNO 265.0102, 267.0082).

Typical experimental procedure preparation of 3 in Scheme 1 (Table 1, entry 4). To a solution of 1,1-diethyl 2-hydrogen ethenetricarboxylate (1) (272 mg, 1.00 mmol) (prepared from 1,1-diethyl 2-*tert*-butyl ethenetricarboxylate¹¹ upon treatment with CF₃CO₂H)¹² in THF (0.8 mL) were added *N*-benzyl-furfurylamine (**2c**) (385 mg, 1.00 mmol) in THF (1.5 mL), Et₃N (0.14 mL, 101 mg, 1.00 mmol), HOBt (1-hydroxybenzotriazole) (270 mg, 2.00 mmol), and EDCI (1-[3-(dimethylamino)propyl]-3-ethylcarbodiimide hydrochloride) (199 mg, 1.04 mmol) at 0 °C. The reaction mixture was stirred for 1 h at 0 °C, and was allowed to warm to 60 °C and then stirred for 20 h. The reaction mixture was concentrated under reduced pressure and the residue was diluted with CH₂Cl₂. The organic phase was washed with saturated aqueous NaHCO₃ solution, 2 M aqueous citric acid, saturated aqueous NaHCO₃ and water, dried (Na₂SO₄), and evaporated *in vacuo*. The residue was purified by column chromatography over silica gel eluting with hexane-Et₂O to give **3c** (175 mg, 46%).

3c: (2.35 mmol scale, 590 mg, 65%): $R_f = 0.8$ (CH₂Cl₂-Et₂O = 1 : 1); colorless crystals; mp 138-140 °C (EtOAc); ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.32 (t, J = 7.1 Hz, 3H), 1.33 (t, J = 7.1 Hz, 3H), 3.47 (s, 1H), 3.59 (d, J = 11.9 Hz, 1H), 3.65 (d, J = 11.9 Hz, 1H), 4.11-4.40 (m, 5H), 4.68 (d, J = 15.0 Hz, 1H), 5.37 (d, J = 1.6 Hz, 1H), 6.32 (dd, J = 5.7, 1.6 Hz, 1H), 6.54 (d, J = 5.7 Hz, 1H), 7.23-7.36 (m, 5H). Selected NOEs are between δ 3.47 (C5-*H*) and δ 3.65 (C2-H*H*), 6.54 (C9-*H*), between δ 3.65 (C2-H*H*) and δ 3.47 (C5-*H*), and between δ 6.32 (C8-*H*) and δ 5.37 (C7-*H*), 6.54 (C9-*H*).; ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 14.07 (CH₃), 14.09 (CH₃), 46.76

(CH₂), 48.13 (CH₂), 54.69 (CH), 61.99 (CH₂), 62.23 (CH₂), 62.38 (CH₂), 83.58 (CH), 89.49 (C), 127.74 (CH), 128.00 (CH), 128.88 (CH), 135.73 (CH), 135.89 (CH), 136.96 (CH), 168.20 (C), 168.86 (C), 170.01 (C). Selected HMBC correlations are between δ 3.59 (C2-*H*H), 3.65 (C2-H*H*), 5.37 (C7-*H*) and δ 136.96 (C9), between δ 3.59 (C2-*H*H), 6.54 (C9-*H*), 5.37 (C7-*H*), 6.32 (C8-*H*) and δ 89.49 (C1) and between δ 3.47 (C5-*H*), 6.54 (C9-*H*), 6.32 (C8-*H*) and δ 83.58 (C7).; IR (KBr) 2984, 1756, 1734, 1690, 1478, 1368, 1266, 1192, 1117, 1046 cm⁻¹; MS (EI) *m/z* 385 (M⁺, 1.9), 340 (9.4), 295 (11), 248 (19), 221 (25), 200 (40), 186 (100%); HRMS (EI) 385.1521 (calcd for C₂₁H₂₃NO₆ M⁺ 385.1525); Anal. Calcd for C₂₁H₂₃NO₆: C, 65.44; H, 6.02; N, 3.63. Found: C, 65.50; H, 6.06; N, 3.76.

3d: (1 mmol scale, 227 mg, 48%): $R_f = 0.1$ (hexane-Et₂O = 1 : 4); pale yellow crystals; mp 127-127.5 °C (EtOAc-Et₂O = 1 : 1); ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.32 (t, J = 7.1 Hz, 3H), 1.33 (t, *J* = 7.1 Hz, 3H), 3.54 (s, 1H), 3.66 (d, *J* = 11.9 Hz, 1H), 3.78 (d, *J* = 11.9 Hz, 1H), 4.24-4.41 (m, 4H), 4.43 (d, J = 15.0 Hz, 1H), 4.61 (d, J = 15.0 Hz, 1H), 6.41 (d, J = 5.5 Hz, 1H), 6.54 (d, J = 5.5 Hz, 1H), 7.25-7.30 (m, 3H), 7.33-7.37 (m, 2H). Selected NOEs are between δ 3.54 (C5-H) and δ 6.54 (C9-H), between δ 3.78 (C2-HH) and δ 6.54 (C9-H), and between δ 6.41 (C8-H) and δ 6.54 (C9-H).; ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 13.96 (CH₃), 14.05 (CH₃), 46.77 (CH₂), 48.09 (CH₂), 57.80 (CH), 62.42 (CH₂), 62.56 (CH₂), 66.86 (C), 87.69 (C), 91.14 (C), 127.86 (CH), 128.02 (CH), 128.94 (CH), 135.50 (C), 137.23 (CH), 140.81 (CH), 166.06 (C), 167.52 (C), 169.23 (C). Selected HMBC correlations are between δ 3.78 (C2-HH) and δ 137.23 (C9), between δ 3.66 (C2-HH), 6.54 (C9-H), 6.41 (C8-H) and δ 87.69 (C1) and between δ 3.54 (C5-H), 6.54 (C9-H), 6.41 (C8-H) and δ 91.14 (C7).; IR (KBr) 2980, 1745, 1718, 1687, 1475, 1440, 1359, 1311, 1288, 1264, 1240, 1213, 1199, 1083, 1037 cm⁻¹; MS (EI) *m/z* 465 (M⁺, 0.6), 463 (M⁺, 0.6), 438 (4.5), 436 (4.6), 266 (92), 264 (100%); HRMS (EI) 463.0648, 465.0623 (calcd for C₂₁H₂₂BrNO₆ M⁺ 463.0631, 465.0610); Anal. Calcd for C₂₁H₂₂BrNO₆: C, 54.32; H, 4.78; N, 3.02. Found: C, 54.22; H, 4.68; N, 3.05.

References

¹ W. Oppolzer, Angew. Chem., Int. Ed. Engl. 1997 16, 10.

² (a) D. D. Sternbach, D. M. Rossana, Tetrahedron Lett. 1982, 23, 303. (b) M. E. Jung, L. J. Street, J. Am. Chem. Soc. 1984, 106, 8327. (c) D. D. Sternbach, D. M. Rossana, and K. D. Onan, Tetrahedron Lett. 1985, 26, 591. (d) M. E. Jung, Synlett 1990, 186. (e) M. E. Jung, J. Gervay, Tetrahedron Lett. 1990, 31, 4685. (f) M. E. Jung, J. Gervay, J. Am. Chem. Soc. 1991, 113, 224. (g) T. Hudlicky, G. Butora, S. P. Fearnley, A. G. Gum, P. J. Persichini, III, M. R. Stabile, J. S. Merola, J. Chem. Soc. Perkin Trans. 1 1995, 2393. (h) C. Giessner-Prettre, S. Hückel, J. Maddaluno, M. E. Jung, J. Org. Chem. 1997, 62, 1439. (i) M. E. Jung, M. Kiankarimi, J. Org. Chem. 1998, 63, 2968. (j) A. L. Schwan, J. L. Snelgrove, M. L. Kalin, R. D. J. Froese, K. Morokuma, Org. Lett. 1999, 1, 487. (k) M. E. Jung, S.-J. Min, J. Am. Chem. Soc. 2005, 127, 10834. (1) A. Padwa, Q. Wang, J. Org. Chem. 2006, 71, 3210. (m) K. Paulvannan, J. W. Jacobs, Tetrahedron 1999, 55, 7433. (n) F. I. Zubkov, I. K. Airiyan, J. D. Ershova, T. R. Galeev, V. P. Zaytsev, E. V. Nikitina, A. V. Varlamov, RSC Adv. 2012, 2, 4103. (o) C.-C. Wang, W.-D. Z. Li, J. Org. Chem. 2012, 77, 4217. (p) C.-H. Chen, G. S. Yellol, C.-H. Tsai, P. B. Dalvi, C.-M. Sun, J. Org. Chem. 2013, 78, 9738. (g) K. A. Parker, M. R. Adamchuk, Tetrahedron Lett. 1978, 1689 ³ (a) S. Yamazaki, K. Fujinami, Y. Maitoko, K. Ueda, K. Kakiuchi, J. Org. Chem. 2013, 78, 8405. (b) S. Yamazaki, Y. Maenaka, K. Fujinami, Y. Mikata, RSC Adv. 2012, 2, 8095. (c) S. Yamazaki, Y. Yamamoto, Y. Fukushima, M. Takebayashi, T. Ukai, Y. Mikata, J. Org. Chem. 2010, 75, 5216. (d) S. Yamazaki, M. Takebayashi, K. Miyazaki, J. Org. Chem. 2010, 75, 1188. (e) S. Yamazaki, Y. Iwata, Y. Fukushima, Org. Biomol. Chem. 2009, 7, 655. (f) S. Morikawa, S. Yamazaki, M. Tsukada, S. Izuhara, T. Morimoto, K. Kakiuchi, J. Org. Chem. 2007, 72, 6459. (g) S. Yamazaki and Y. Iwata, J. Org. Chem. 2006, 71, 739.

⁴ B. B. Snider, D. M. Roush, T. A. Killinger, J. Am. Chem. Soc. 1979, 101, 6023.

⁵ (a) H. K. Hall, P. Nogues, J. W. Rhoades, R. C. Sentman, M. Detar, *J. Org. Chem.* 1982, 47, 1451. (b) W. Zestawski, J. Barafiska, M. Jamrozik, J. Jamrozik, *Monatsh. Chem.* 1997, 128, 389.
(c) E. Winterfeldt, H.-J. Dillinger, *Chem. Ber.* 1966, 99, 1558.

⁶ S.Yamazaki, H. Sugiura, M. Niina, Y. Mikata, A. Ogawa, *Heterocycles* 2016, 92, 485.

⁷ The byproducts by the reaction of CH₂ClCH₂Cl and HOBt in the presence of Et₃N were formed and removed by column chromatography. (a) J.-G. Ji, D.-Y. Zhang, Y.-H. Ye, Q.-Y. Xing, *Tetrahedron Lett.* **1998**, 39, 6515. (b) W. A. Feld, D. G. Evans, *J. Chem. Eng. Data* **1983**, 28, 138.

⁸ S. Duan, R. Jana, J. A. Tunge, *J. Org. Chem.* **2009**, 74, 4612.

⁹ A. Volonterio, C. R. de Arellano, M. Zanda, J. Org. Chem. 2005, 70, 2161.

- ¹⁰ H.-P. Wu, R. Aumann, R. Fröhlich, E. Wegelius, *Organometallics* **2001**, 20, 2183.
- ¹¹ S. Yamazaki, K. Ohnitsu, K. Ohi, T. Otsubo, K. Moriyama, Org. Lett. 2005, 7, 759.
- ¹² (a) S. Yamazaki, S. Morikawa, Y. Iwata, M. Yamamoto, K. Kuramoto, Org. Biomol. Chem.

2004, 2, 3134. (b) S. Yamazaki, M. Yamamoto, S. Morikawa, Heterocycles 2006, 67, 269.

Chapter 3

Intramolecular [2 + 2] and [4 + 2] Cycloaddition Reactions of Cinnamylamides of Ethenetricarboxylate in Sequential Processes

3-1 Introduction

Multiple bond formations in one pot are efficient to synthesize cyclic compounds. Intramolecular cycloaddition reactions in cinnamyl group (styrenes) give [2 + 2] and [4 + 2] cycloadducts.^{1, 3b}

Intramoramolecuar Diels-Alder (IMDA) reaction between styryl group as dienes and alkynes were reported can be to give tricyclic compounds.² One the other hand, alkenes bearing styryl group to undergoing [2 + 2] cycloaddition give cyclobutane-fused compounds by photochemical¹ or metal-catalyzed reaction.³ Snider et al. reported that heating cinnamyl ester of ethentricarboxylate led to an equilibrium mixture of the ester and a hetero Diels-Alder adduct.⁴ Thus, styrene works as an alkene or diene component in intramolecular [2 + 2] or [4 + 2] cycloadditions with electron-deficient alkenes. The both reactions may be useful for the construction of multicyclic skeletons and the question is how to control the selectivity.

Ethenetricarboxylate derivatives have been employed as highly electrophilic C=C components in various bond-forming reactions.⁵ In chapter 2, reaction of 1,1-diethyl 2-hydrogen ethenetricarboxylate 1 and 2-furylmethylamines in the presence of EDCI/HOBt/Et₃N at room temperature led directly to IMDA adducts in one pot.⁶

It is of interest to investigate the reaction of the highly electrophilic ethenetricarboxylates bearing styryl group as an extension of an alkenyl group. In this chapter, the reaction of cinnamyl amides bearing electron-donating groups and electron withdrawing-groups on the benzene ring has been examined. Development of selective intramolecular cyclization of styrenes and elucidation the factor to control the selectivity.

3-2 Result and Discussion

First, reactions of 1,1-diethyl 2-hydrogen ethenetricarboxylate 1 and *E*-cinnamylamines (X = H) 2a in the presence of EDCI/HOBt/Et₃N have been examined. It was found that the reaction gave cyclobutane-fused pyrrolidines 3a in 43% yield as the isolable major product (Table 3-1). The products may be formed via amide formation/intramolecular [2 + 2] cycloaddition. Reaction of 1 and 2b-d (R=CH₂-cyclohexyl, CH₂C₆H₄-4-CF₃, CH₂CH=CH₂) gave the cyclobutane-fused pyrrolidines in 41-51% yields similarly.⁷ Reaction of RHNCH₂-CH=CH-C₆H₄-X (X = 4-halogen, 4-OCH₃) 2e-h also gave cyclobutane-fused pyrrolidines 3e-h in 39–48% yields as the isolable major products.

Et(D ₂ Et R +	Ň H	HOBt EDCI Et ₃ N r.t. THF	R=N A
	1		2		3 X
	Entry ^a	2	R	Х	Product (Yield)
	1	2a	CH ₂ Ph	Н	3a (43%)
	2	2b	CH ₂ -cyclohexyl	Н	3b (51%)
	3	2c	$CH_2C_6H_4$ -4- CF_3	Н	3c (41%)
	4	2d	CH ₂ CH=CH ₂	Н	3d (42%)
	5	2e	CH ₂ Ph	F	3e (39%)
	6	2f	CH ₂ Ph	Cl	3f (40%)
	7	2g	CH ₂ Ph	Br	3g (40%)
	8	2h	CH ₂ Ph	OCH ₃	3h (48%)

Table 3-1. Reactions of 1,1-dietyl 2-hydrogene ethenetricarboxylate 1 and *E*-cinnamylamines 2a-h.

^a Results of entries 2-8 are cited from reference 7

The intermediate amide **A** was not observed under the reaction conditions of amide formation (Scheme 3-1). The amide undergoes the first C–C bond formation to give a zwitter-ionic intermediate **B**, which is stabilized by the phenyl group. The second C–C bond formation proceeds, affording a highly strained cyclobutane-fused bicyclic compound **3**.



Scheme 3-1. Proposed mechanism for formation of cyclobutane-fused bicyclic compounds 3.

When the reaction of **1** and **2a** was carried out at 80 °C in 1,2-dichloroethane (ClCH₂CH₂Cl) or in α, α, α -trifluorotoluene, δ -lactone-fused pyrrolidine **4a** was obtained as the major product in 69% and 50% yields, respectively (Scheme 3-2). The relative configuration of **4** was determined as shown in Scheme 3-2 by NOEs.



Scheme 3-2. Formation of δ -lactone-fused pyrrolidine 4a.

Formation of **4** from **3** under the reaction conditions is likely. The reaction conditions may produce a small amount of HCl from EDCI along with formation of the byproducts BtOCH₂CH₂Cl and BtOCH₂CH₂OBt.⁸ Reaction of cyclobutane products **3** with HCl was next examined. After examining various ring-opening conditions, the reaction of cyclobutane **3a** with 1 equiv of HCl/ether and 1 equiv of H₂O in ClCH₂CH₂Cl at 80 °C for 20 h was found to give **4a** efficiently in 70% yield (Table 3-2, entry 1). The reaction of **3a** with 1 equiv of HCl/H₂O in THF at room temperature gave the mixture of alcohol **5a** and **4a** (entry 2). Treatment of alcohol **5a** with 1 equiv of HCl/ether in CH₂Cl₂ at room temperature overnight gave **4a** quantitatively. On the other hand, in entry 3, the reaction of **3a** with 1 equiv of HCl/ether in CH₂Cl₂ at room temperature gave Cl-adduct **6a** as a single diastereomer along with **4a**. The stereochemistries of **5a** and **6a** could be deduced as follows.

O N Ph	H CO ₂ Et CO ₂ Et conditions H $3a$ CO_2Et conditions Ph H $4a$ CO_2Et CO_2Et H O H O H O H O H O H O H O CO_2Et H O H O O H O O H O O H O O D D D D D D D D	EtO_2C CO_2Et H H H H Cl Cl H
Entry	Conditions	Product (Yield)
1	1 equiv of 1 M HCl/ether, 1 equiv of H ₂ O, ClCH ₂ H ₂ Cl 80 °C	4a (70%)
2	1 equiv of 1 M HCl/H ₂ O, THF, r.t.	5a (42%), 4a (47%)
3	1 equiv of 1 M HCl/ether, CH ₂ Cl ₂ , r.t.	6a (60%), 4a (27%)

Table 3-2. Ring-opening reactions of cyclobutane-fused pyrrolidine 3a.

The 3,4-*cis* stereochemistries of **5a** and **6a** were determined by NOEs. Preferred conformations of **5a** and **6a** may be as depicted in Figure 3-1 from the coupling constants and consideration of steric effects, respectively. The coupling constant between CH(OH)Ph and C4-H of **5a** (J = 10.9 Hz) and that between CHClPh and C4-H of **6a** (J = 4.3Hz) suggest the configurations of the side-chains, as shown in figure 3-1. The similarity in the coupling constant

between CHOHPh and C4–H of **5a** and that between C4–H and C3a–H of **4a** (J = 11.3 Hz) supports the assignment of the configuration of **5a**.



Figure 3-1. Conformations of 5a and 6a.

Thus, δ -lactone 4 may form from cyclobutane 3 via intermiediate **B-H**⁺ and alcohol 5, followed by transesterification (Scheme 3-3). Formation of 5 may proceed in two steps and formation of Cl-adducts 6 may proceed in one step ring opening based on their suggested stereochemistries.



Scheme 3-3. Reaction mechanism of formation of 4, 5, and 6.

Next, the reaction of 1,1-diethyl 2-hydrogen ethenetricarboxylate **1** and cinnamylamines bearing electron-withdrawing groups on *ortho* and *para* positions in the presence of the amide

condensation reagents was examined. Interestingly, reaction of **1** and PhCH₂HNCH₂– CH=CH–C₆H₄–X (X = 2- or 4-NO₂, CN, CO₂Me, CO₂Et, or CF₃) **2i–n** with EDCI/HOBt/Et₃N at room temperature, 60 °C, and 80 °C gave tetrahydrobenz[*f*]isoindolines **7** as the major products via [4 + 2] cycloaddition (Table 3-3). The *trans*-fused pyrrolidine stereochemistry of **7** was determined by NOEs (in C₆D₆, CD₃CN, or (CD₃)₂CO, for some products).

 Table 3-3. [4 + 2] Cycloaddition reaction of 1 and electron-withdrawing group substituted

 cinnamylamines 2.

EtO ₂ C	С	CO ₂ Et +	HOBt EDCI Et ₃ N	C 1 Ph 2	$\begin{array}{c c} & CO_2Et \\ H & CO_2E \\ \hline 9a 9 & 8 \\ \hline 3a & 4 \\ \hline 3 & H & 5 \end{array}$	t $\sqrt{-1}^{7}X$ Ph	H CO ₂ Et CO ₂ Et
	1	Ph	2		7		3
Entry	2	Х	Solvent	Temp.	7 Yield(%)	Х	3 Yield(%)
1 ^a	2i	2-NO ₂	THF	r.t.	75%	5-NO ₂	
2 ^a	2i	2-NO ₂	Benzene	80 °C	54%	5-NO ₂	
3	2j	$4-NO_2$	THF	r.t.	46%	7-NO ₂	
4	2j	$4-NO_2$	THF	60 °C	68%	7-NO ₂	
5	2j	$4-NO_2$	ClCH ₂ CH ₂ Cl	80 °C	66%	7-NO ₂	
6	2k	4-CN	THF	r.t.	71%	7-CN	
7	2k	4-CN	THF	60 °C	75%	7-CN	
8	21	4-CO ₂ Me	THF	r.t.	71%	7- CO ₂ Me	b
9	21	4-CO ₂ Me	THF	60 °C	66%	7- CO ₂ Me	b
10	2m	4-CO ₂ Et	THF	r.t.	57%	7- CO ₂ Et	b
11	2m	4-CO ₂ Et	THF	60 °C	56%	7- CO ₂ Et	b
12	2n	4-CF ₃	THF	r.t.	49%	7-CF ₃	3%
13	2n	4-CF ₃	THF	60 °C	48%	7-CF ₃	6%
14	2n	4-CF ₃	Benzene	80 °C	51%	7-CF ₃	3%

^a Results of entries 1 and 2 are cited from reference 7. ^b A small amount of cyclobutanefused pyrrolidine **3** was detected but could not be isolated.

Formation of the zwitter-ionic intermediate **B** corresponding to that in Scheme 3-1 may be strongly destabilized by the resonance and inductive effects of *ortho* and *para* electron-withdrawing group on the benzene ring (Scheme 3-4). Instead, the interaction between a styrene moiety and an alkene moiety of ethenetricarboxylate may lead to the IMDA adduct **C**. The 1,3-H transfer isomerization of **C** to the products **7** may proceed by a stepwise process via intermediate **D-H**⁺.



Scheme 3-4. Reaction mechanism of formation 7.

In order to examine the effects of electron-withdrawing group in [4 + 2] cycloaddition of a styrene moiety and the generality of the reaction, the reactions of other electron-deficient olefins 8-10 with the carboxyl group and cinnamylamines without substituents 2a and with o-NO₂ group 2i were carried out (Table 3-4). Reaction of monomethyl maleate 8 and 2a or 2i with EDCI/HOBt/Et₃N at room temperature gave amides 11a and 11i as isolable products along with the corresponding *trans* isomers 12 (Table 3-4, entries 1, 2). Formation of byproducts 12 may arise from partial isomerization of 8 to 9 under the reaction conditions. Reaction of monomethyl fumarate 9 and 2a or 2i gave amides 12a and 12i respectively. Reaction of 4,4,4-trifluoro-3-(trifluoromethyl)-crotonic acid 10 and 2i with EDCI/HOBt/Et₃N at room temperature gave amide 13i in 57% yield.



Table 3-4. Reactions of electron-deficient olefins 8-10 with carboxyl group and cinnamylamines.

^a Results of entries 1-4 are cited from reference 7. ^b A small amount of impurity could not be removed. ^c 12a could be formed but not confirmed. ^d 12i was formed in 11% yield as byproduct.

Compound 11i gradually changes to 14i at room temperature. Heating 11i at 80 °C in ClCH₂CH₂Cl for 18 h gave 14i via [4 + 2] cycloaddition/H-transfer.⁷ On the other hand, heating 11a at 80 °C in ClCH₂CH₂Cl for 18 h gave complex mixtures. The reaction of 12i at 110 °C in toluene for 18 h gave 15i as isolable products (Table 3-5). Reaction of 12a at 80 °C in ClCH₂CH₂Cl for 18 h remained starting materials, and the reaction at 110 °C in toluene for 18 h gave complex mixtures. The stereochemistries of 14i and 15i were determined by NOEs. The pyrrolidine ring junction is *trans*. Thermal [4 + 2] cycloaddition reaction of 11i and 12i proceeded

stereospecifically, and the product retained the original *cis* and *trans* stereochemistries of C=C double bonds. Thermal reaction of **13i** at 80 °C in ClCH₂CH₂Cl for 22 h gave ca. 1:1 mixture of **13i** and **16i**. Heating **13i** at 110 °C in toluene for 20 h completed the conversion, and **16i** was obtained in 89% yield.

$\begin{array}{cccccccccccccccccccccccccccccccccccc$						
Entry ^a	11-13	\mathbf{R}_1	R_2	Х	Temp.	Product (Yield)
1	11a	CO ₂ Me	Н	Н	80 °C	14a (0) ^b
2	11i	CO ₂ Me	Н	2-NO ₂	80° C	14i (33)
3	12a	Н	CO ₂ Me	Н	110 °C	15a (0) ^b
4	12i	Н	CO ₂ Me	2-NO ₂	110 °C	15i (31)
5	13i	CF ₃	CF ₃	2-NO ₂	110 °C	16i (89)

Table 3-5. Thermal reaction of amide derivatives.

^a Results of entries1-4 are cited from reference 7. ^b Complex mixtures.

Higher reactivity of **11** than that of **13** may arise from preferable steric overlap on the transition states of [4 + 2] cycloaddition (Scheme 3-5). Much higher reactivity of ethenetricarboxylate intermediates **A** compared to **11** and **13** may arise from activation of C=C double bond by three electron-withdrawing carbonyl groups. Lower reactivity of **13** than that of **A** could be due to the steric effect of CF₃ groups.



Scheme 3-5. Steric efficients in [4+2] cycloaddition.

3-3 Conclusion

In summary, intramolecular [2 + 2] and [4 + 2] (IMDA) cycloaddition reactions of cinnamylamides and ethenetricarboxylate in sequential processes have been studied. Reaction of cinnamylamines without substituents on the benzene ring and with halogens and OMe on *para* positions at room temperature gave cyclobutane-fused pyrrolidines as the major products via [2 + 2] cycloaddition. The reaction at 80 °C in 1.2-dichloroethane gave δ -lactone-fused pyrrolidines as the major products, possibly via ring-opening of the cyclobutanes. Interestingly, reaction of 1,1-diethyl 2-hydrogen ethenetricarboxylate and cinnamylamines bearing electronwithdrawing groups such as NO₂, CN, CO₂Me, CO₂Et, or CF₃ on *ortho* and *para* positions in the presence of EDCI/HOBt/ Et₃N at room temperature or at 60–80 °C gave tetrahydrobenz[*f*]isoindolines via IMDA reaction as the major products. Diversity of the reaction pattern depending on the substituents of the benzene ring was found. The synthesized highly functionalized heterocyclic products and promising as useful precursors to synthesize functionalmaterials and physiologically active substances.

3-4 Experimental Section

General Procedures

¹H Chemical shifts are reported in ppm relative to Me₄Si. ¹³C Chemical shifts are reported in ppm relative to CDCl₃ (77.1 ppm). ¹⁹F Chemical shifts are reported in ppm relative to CFCl₃. ¹³C Mutiplicities were determined by DEPT and HSQC. Mass spectra were recorded at an ionizing voltage of 70 eV by EI, FAB, or ESI. Mass analyzer type used for EI, FAB, and is double-focusing and that for ESI is TOF in the HRMS measurements. All reactions were carried out under a nitrogen atmosphere. Column chromatography was performed on silica gel (75–150 μm).

Ethenetricarboxylate 1 was prepared according to the literature.⁹ Cinnamylamines 2a–n were prepared from the corresponding cinnamaldehydes and amines by reductive amination in methanol (for 2a–l, 2n) or ethanol (for 2m) according to the literature procedure.¹⁰ ¹H NMR of 2a was in accord with the reported data.¹¹

4-cyanocinnamaldehyde (86%) was prepared from the corresponding benzaldehydes and acetoaldehyde according to the literature procedure.¹² ¹H NMR spectra of 4-cyanocinnamaldehydewas in accord with reported data. ¹³ 4-(Methoxycarbonyl)cinnamaldehyde (59%) was prepared by the the palladium-catalyzed reaction of the corresponding aryl iodides with acrolein diethyl acetal.¹³ ¹H NMR spectra of 4-(methoxycarbonyl)cinnamaldehyde were in accord with the reported data.¹⁴ 4-(Ethoxycarbonyl)cinnamaldehyde was prepared according to the literature.¹³ 4-(Trifluoromethyl)cinnamaldehyde (58%) was prepared from the corresponding benzaldehydes and formylmethylenetriphenylphosphorane according to the literature procedure.¹⁵

4-(Trifluoromethyl)cinnamaldehyde: (8.2 mmol scale, 0.951 g, 58%); $R_f = 0.6$ (hexane–ether = 1:1); pale yellow crystals; mp 60 °C; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 6.78 (dd, J = 16.0, 7.6 Hz, 1H), 7.52 (d, J = 16.0 Hz, 1H), 7.69 (s, 4H), 9.76 (d, J = 7.6 Hz, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 123.7 (C, q, $J_{CF} = 272$ Hz), 126.0 (CH, q, $J_{CF} = 3.8$ Hz), 128.6 (CH), 130.5 (CH), 132.4 (C, q, $J_{CF} = 33$ Hz), 137.3 (C, q, $J_{CF} = 1.5$ Hz), 150.3 (CH), 193.2 (CH); ¹⁹F NMR (376 MHz, CDCl₃) δ

(ppm) -63.05; IR (KBr) 2817, 2733, 1680, 1324, 1172, 1122, 1066 cm⁻¹; MS (EI) *m/z* 200 (M⁺, 38), 199 (32), 151 (47), 131 (100%); HRMS (EI) *m/z* M⁺ 200.0448 (calcd for C₁₀H₇F₃O 200.0449).

Typical experimental procedure for preparation of cinnamylamines 2. A solution of 4-nitrocinnamaldehyde (1.771 g, 10 mmol) and benzylamine (0.954 g, 8.9 mmol) in methanol (6.8 mL) was heated under reflux for 30 min, followed by the portionwise addition of NaBH₄ (567 mg, 15 mmol) in ice-cooled bath. The mixture was stirred overnight at room temperature. Excess sodium borohydride was quenched by the addition of acetone (3.7 mL). The mixture was concentrated, and the residue was dissloved in CH_2Cl_2 and water. The organic layer was washed with water, dried over Na₂SO₄, and concentrated. The residue was purified by column chromatography over silica gel eluting with hexane–Et₂O to give **2j** (1.08 g, 45%).

Benzyl 4-nitrocinnamylamine (2j): (8.9 mmol scale, 1.08 g, 45%); $R_f = 0.2$ (hexane–ether = 1:4); yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.68 (bs, 1H), 3.49 (dd, J = 5.9, 1.4 Hz, 2H), 3.85 (s, 2H), 6.50 (dt, J = 16.0, 5.9 Hz, 1H), 6.62 (d, J = 16.0 Hz, 1H), 7.24–7.30 (m, 1H), 7.31–7.35 (m, 4H), 7.47 (d-like, J = 8.9 Hz, 2H), 8.15 (d-like, J = 8.9 Hz, 2H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 50.9 (CH₂), 53.5 (CH₂), 124.0 (CH), 126.7 (CH), 127.2 (CH), 128.2 (CH), 128.5 (CH), 129.1 (CH), 133.8 (CH), 134.0 (C), 143.7 (C), 146.8 (C); IR (neat) 3328, 3027, 2833, 1651, 1595, 1520, 1494, 1454, 1346, 1110, 971 cm⁻¹; MS (EI) *m/z* 268 (M⁺, 6.9), 196 (16), 132 (23), 91 (100%); HRMS (EI) *m/z* M⁺ 268.1207 (calcd for C₁₆H₁₆N₂O₂ 268.1212).

Benzyl 4-cyanocinnamylamine (2k): (6.4 mmol scale, 0.837 g, 53%); $R_f = 0.2$ (hexane–ether = 1:4); yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.55 (bs, 1H), 3.47 (dd, J = 5.9, 1.4 Hz, 2H), 3.84 (s, 2H), 6.44 (dt, J = 15.9, 5.9 Hz, 1H), 6.56 (d, J = 15.9 Hz, 1H), 7.24– 7.30 (m, 1H), 7.32–7.35 (m, 4H), 7.42 (d-like, J = 8.4 Hz, 2H), 7.57 (d-like, J = 8.4 Hz, 2H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 50.9 (CH₂), 53.5 (CH₂), 110.5 (C), 119.0 (C), 126.7 (CH), 127.2 (CH), 128.2 (CH), 128.5 (CH), 129.5 (CH), 132.4 (CH), 132.8 (CH), 140.0 (C), 141.7 (C); IR (neat) 3315, 3028, 2821, 2224, 1651,

1604, 1495, 1453, 1412, 1360, 1175, 1118, 971 cm⁻¹; MS (EI) *m/z* 248 (M⁺, 13), 196 (10), 146 (32), 106 (34), 91 (100%); HRMS(EI) *m/z* M⁺ 248.1317 (calcd for C₁₇H₁₆N₂ 248.1313).

Benzyl 4-(methoxycarbonyl)cinnamylamine (2l): (5 mmol scale, 0.625 g, 44%); $R_f = 0.2$ (hexane–ether = 1:4); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.52 (bs, 1H), 3.46 (dd, J = 6.1, 1.4 Hz, 2H), 3.85 (s, 2H), 3.90 (s, 3H), 6.44 (dt, J = 15.9, 6.1 Hz, 1H), 6.58 (d, J = 15.9 Hz, 1H), 7.22–7.29 (m, 1H), 7.31–7.35 (m, 4H), 7.41 (d, J = 8.3 Hz, 2H), 7.97 (d-like, J = 8.3 Hz, 2H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 51.1 (CH₂), 52.1 (CH₃), 53.5 (CH₂), 126.2 (CH), 127.1 (CH), 128.2 (CH), 128.5 (CH), 128.8 (C), 130.0 (CH), 130.4 (CH), 131.5 (CH), 140.2 (C), 141.7 (C), 167.0 (C); IR (neat) 3326, 3028, 2950, 1721, 1606, 1454, 1435, 1281, 1178, 1109, 1017, 971 cm⁻¹; MS (EI) *m*/*z* 281 (M⁺, 14), 132 (35), 106 (25), 91 (100%); HRMS (EI) *m*/*z* M⁺ 281.1417 (calcd for C₁₈H₁₉NO₂ 281.1416).

Benzyl 4-(ethoxycarbonyl)cinnamylamine (2m): (6 mmol scale, 0.832 g, 47%); $R_f = 0.2$ (hexane–ether = 1:4); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.39 (t, J = 7.1 Hz, 3H), 1.56 (bs, 1H), 3.46 (dd, J = 6.1, 1.4 Hz, 2H), 3.85 (s, 2H), 4.36 (q, J = 7.1 Hz, 2H), 6.43 (dt, J = 15.9, 6.1 Hz, 1H), 6.58 (d, J = 15.9 Hz, 1H), 7.24–7.29 (m, 1H), 7.31–7.36 (m, 4H), 7.41 (d-like, J = 8.4 Hz, 2H), 7.98 (d-like, J = 8.4 Hz, 2H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 14.4 (CH₃), 51.2 (CH₂), 53.5 (CH₂), 60.9 (CH₂), 126.1 (CH), 127.1 (CH), 128.2 (CH), 128.5 (CH), 129.2 (C), 129.9 (CH), 130.4 (CH), 131.3 (CH), 140.2 (C), 141.6 (C), 166.5 (C); IR (neat) 3316, 2980, 1713, 1607, 1495, 1453, 1413, 1366, 1275, 1178, 1105, 1020, 972 cm⁻¹; MS (EI) m/z 295 (M⁺, 31), 204 (20), 132 (71), 91 (100%); HRMS (EI) m/z M⁺ 295.1581 (calcd for C₁₉H₂₁NO₂ 295.1572).

Benzyl 4-(trifluoromethyl)cinnamylamine (2n): (3.6 mmol scale, 0.996 g, 96%); $R_f = 0.5$ (hexane–ether = 1:1); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.55 (bs, 1H), 3.46 (dd, J = 6.1, 1.4 Hz, 2H), 3.84 (s, 2H), 6.41 (dt, J = 15.8, 6.1 Hz, 1H), 6.58 (d, J = 15.8 Hz, 1H), 7.25–7.30 (m, 1H), 7.32–7.35 (m, 4H), 7.44 (d, J = 8.2 Hz, 2H), 7.55 (d, J = 8.2 Hz, 2H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 51.1 (CH₂), 53.5 (CH₂), 124.1 (C, q, $J_{CF} = 272$ Hz), 125.6

(CH, q, $J_{CF} = 3.8$ Hz), 126.5 (CH), 127.2 (CH), 128.3 (CH), 128.6 (CH), 129.2 (C, q, J = 32 Hz), 130.0 (CH), 131.4 (CH), 140.2 (C), 140.7 (C); ¹⁹F NMR (376 MHz, CDCl₃) δ (ppm) –62.50; IR (neat) 3310, 3029, 2823, 1652, 1615, 1495, 1455, 1415, 1327, 1163, 1120, 1067, 1016, 970 cm⁻¹;MS (EI) *m/z* 291 (M⁺, 100), 200 (11), 185 (35), 132 (67), 91 (100%); HRMS (EI) *m/z* M⁺ 291.1235 (calcd for C₁₇H₁₆F₃N 291.1235).

Typical experimental procedure for preparation of 3, 7, 11-13 (Table 3-1, entry 1). To a solution of 1,1-diethyl 2-hydrogen ethenetricarboxylate (1) (prepared from 1,1-diethyl 2-*tert*-butyl ethenetricarboxylate (272 mg, 1 mmol) upon treatment with CF₃CO₂H (4 mL))⁹ in THF (0.7 mL) were added benzyl einnamylamine (**2a**) (223 mg, 1 mmol) in THF (0.7 mL), Et₃N (0.14 mL, 102 mg, 1 mmol), HOBt (1-hydroxybenzotriazole) (270 mg, 2 mmol), and EDCI (1-[3-(dimethylamino)-propyl]-3-ethylcarbodiimide hydrochloride) (199 mg, 1.04 mmol) at 0 °C. The reaction mixture was stirred for 1 h at 0 °C, and was allowed to warm to room temperature and stirred for 20 h. The reaction mixture was concentrated under reduced pressure and the residue was diluted with CH₂Cl₂. The organic phase was washed with saturated aqueous NaHCO₃ solution, 2 M aqueous citric acid, saturated aqueous NaHCO₃ and water, dried (Na₂SO₄), and evaporated *in vacuo*. The residue was purified by column chromatography over silica gel eluting with hexane-Et₂O to give **3a** (180 mg, 43%).

3a: $R_f = 0.1$ (hexane-ether = 1 : 8); colorless crystals; mp 137– 138.5 °C; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.29 (t, J = 7.0 Hz, 3H), 1.34 (t, J = 7.1 Hz, 3H), 2.39 (ddd, J = 10.7, 7.0, 5.9 Hz, 1H), 2.67 (d, J = 10.7 Hz, 1H), 3.31 (dd, J = 10.7, 5.9 Hz, 1H), 3.89 (d, J = 7.0 Hz, 1H), 3.89 (d, J = 14.3 Hz, 1H), 4.03-4.17 (m, 2H), 4.22-4.36 (m, 3H), 4.89 (d, J = 14.3 Hz, 1H), 6.75 (d-like, J = 7.6 Hz, 2H), 7.22-7.42 (m, 8H). Selected NOEs are between δ 2.39 (C5-*H*) and δ 3.31 (C4-H*H*), 6.75 (Ar-*H*), 3.89 (C1-*H*). Atom numbering is shown in Table 3-1.; ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 14.5 (CH₃), 15.0 (CH₃), 36.1 (CH), 41.1 (CH), 44.8 (CH₂), 46.4 (CH₂), 59.9 (CH₂), 64.8 (CH₂), 79.2 (C), 79.8 (CH), 127.4 (CH), 127.9 (CH), 128.7 (CH), 128.9 (CH), 129.05

(CH), 129.11 (CH), 136.6 (C), 136.8 (C), 163.0 (C), 167.3 (C), 173.1 (C). Selected HMBC correlations are between δ 2.39 (C5-*H*), 2.67 (C4-H*H*), 3.89 (C1-*H*) and δ 173.1 (*C*2), between δ 2.39 (C5-*H*), 2.67 (C4-*H*H), 3.31 (C4-H*H*), 3.89 (C1-*H*) and δ 79.8 (C6), between δ 2.67 (C4-*H*H) and δ 41.1 (C1) and between δ 2.67 (C4-*H*H), 3.31 (C4- H*H*), 3.89 (C1-*H*) and δ 36.1 (C5).; IR (KBr) 2981, 1699, 1634, 1285, 1079 cm⁻¹; MS (EI) *m/z* 421 (M⁺, 14), 222 (42), 199 (58), 132 (63), 91 (100%); HRMS *m/z* M⁺ 421.1886 (calcd for C₂₅H₂₇NO₅ 421.1889).

Typical experimental procedure for preparation of 4a. To a solution of 1,1-diethyl 2-hydrogen ethenetricarboxylate (1) (prepared from 1,1-diethyl 2-tert-butyl ethenetricarboxylate (272 mg, 1 mmol) upon treatment with CF₃CO₂H (4 mL))⁹ in 1,2-dichloroethane (0.7 mL) were added benzyl cinnamylamine (**2a**) (201 mg, 0.90 mmol) in 1,2-dichloroethane (0.7 mL), Et₃N (0.14 mL, 102 mg, 1 mmol), HOBt (1-hydroxybenzotriazole) (270 mg, 2 mmol), and EDCI (1-[3-(dimethylamino)propyl]- 3-ethylcarbodiimide hydrochloride) (199 mg, 1.04 mmol) at 0 °C. The reaction mixture was stirred for 1 h at 0 °C, and was allowed to warm to 80 °C and then stirred for 20 h. The reaction mixture was diluted with CH₂Cl₂. The organic phase was washed with saturated aqueous NaHCO₃ solution, 2 M aqueous citric acid, saturated aqueous NaHCO₃, and water, dried (Na₂SO₄), and evaporated *in vacuo*. The residue was purified by column chromatography over silica gel eluting with hexane–Et₂O to give **4a** (246 mg, 69%).

4a: $R_f = 0.1$ (hexane–ether = 1:4); colorless crystals; mp 107.5– 108 °C; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.39 (t, J = 7.1 Hz, 3H), 2.78–2.86 (m, 2H), 3.28 (dd, J = 11.2, 7.9 Hz, 1H), 3.26–3.78 (m, 2H), 4.33 (d, J = 14.4 Hz, 1H), 4.40 (q, J = 7.1 Hz, 2H), 4.55 (d, J = 14.4 Hz, 1H), 4.78 (d, J = 11.3 Hz, 1H), 7.07 (d-like, J = 8.0 Hz, 2H), 7.21–7.24 (m, 2H), 7.29–7.39 (m, 6H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 14.2 (CH₃), 37.2 (CH), 41.1 (CH), 45.9 (CH₂), 46.8 (CH₂), 47.1 (CH), 62.5 (CH₂), 81.4 (CH), 127.6 (CH), 128.2 (CH), 128.5 (CH), 129.0 (CH), 129.1 (CH), 129.7 (CH), 135.2 (C), 135.7 (C), 167.5 (C), 167.6 (C), 172.2 (C); ¹H NMR (400 MHz, CD₃CN) δ
(ppm) 1.33 (t, J = 7.1 Hz, 3H), 2.70 (dd, J = 10.8, 1.9 Hz, 1H), 3.00 (dddd, J = 11.7, 10.1, 8.2, 1.9 Hz, 1H), 3.26 (dd, J = 10.8, 8.2 Hz, 1H), 3.65 (dd, J = 10.6, 10.1 Hz, 1H), 3.83 (d, J = 10.6 Hz, 1H), 4.30 (d, J = 14.8 Hz, 1H), 4.317 (q, J = 7.1 Hz, 1H), 4.320 (q, J = 7.1 Hz, 1H), 4.48 (d, J = 14.8 Hz, 1H), 5.10 (d, J = 11.7 Hz, 1H), 7.22–7.26 (m, 4H), 7.30–7.40 (m, 6H). Selected NOEs are between δ 3.00 (C3a–*H*) and δ 3.26 (C3–H*H*), 3.65 (C7a–*H*) and between δ 2.70 (C3–*H*H) and δ 5.10 (C4–*H*). Atom numbering is shown in Scheme 3-2. ¹³C NMR (100.6 MHz, CD₃CN) δ (ppm) 14.5 (CH₃), 36.8 (CH), 41.9 (CH), 46.8 (CH₂), 46.9 (CH₂), 48.2 (CH), 62.7 (CH₂), 82.1 (CH), 128.6 (CH), 128.8 (CH), 129.0 (CH), 129.7 (CH), 129.8 (CH), 130.4 (CH), 137.0 (C), 137.4 (C), 169.0 (C), 169.3 (C), 173.2 (C). Selected HMBC correlations are between δ 2.70 (C3–*H*H), 3.26 (C3–*H*H), 3.00 (C3a–*H*), 5.10 (C4–*H*) and δ 41.9 (C7a). IR (KBr) 3448, 2929, 1752, 1740, 1691, 1449, 1375, 1266, 1156, 1045, 1021 cm⁻¹; MS (EI) *m/z* 393 (M⁺, 16), 186 (30), 91 (61), 57 (100%); HRMS (EI) *m/z* M⁺ 393.1574 (calcd for C₂₃H₂₃NO₅ 393.1576).

Transformation of 3a to 4a (Table 3-3, entry 1). To a solution of **3a** (210 mg, 0.5 mmol) in $ClCH_2CH_2Cl(0.7 \text{ mL})$ were added 1 M HCl/ether (0.5 mL, 0.5 mmol) and H₂O (9 mg, 0.5 mmol). The mixture was stirred at 80 °C for 20 h. The reaction mixture was concentrated under reduced pressure. The residue was purified by column chromatography over silica gel eluting with hexane–Et₂O to give **4a** (139 mg, 70%). Transformation of 3a to **5a** and **4a** (Table 3-2, entry 2). To a solution of **3a** (245 mg, 0.58 mmol) in THF (0.8 mL) was added 1 M HCl/H₂O (0.58 mL, 0.58 mmol). The mixture was stirred at room temperature for 20 h. The reaction mixture was concentrated under reduced pressure. The residue was diluted with CH₂Cl₂. The organic phase was washed with water, dried (Na₂SO₄), and evaporated *in vacuo*. The residue was purified by column chromatography over silica gel eluting with hexane–Et₂O to give **5a** (111 mg, 42%) and **4a** (107 mg, 47%).

5a: Rf = 0.6 (ether); colorless oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.27 (t, J = 7.1 Hz, 3H), 1.35 (t, J = 7.1 Hz, 3H), 2.15 (bs, 1H), 2.54 (dd, J = 10.3, 2.6 Hz, 1H), 2.87 (dddd, J = 10.9, 7.4, 6.6, 2.6 Hz, 1H), 2.97 (dd, J = 10.2, 6.6 Hz, 1H), 3.67 (dd, J = 10.2, 7.4 Hz, 1H), 4.06 (d, J = 14.5 Hz, 1H), 4.08 (d, J = 10.2 Hz, 1H), 4.19–4.39 (m, 5H), 4.58 (d, J = 14.5 Hz, 1H), 6.87–6.89 (m, 2H), 7.16–7.23 (m, 5H), 7.29–7.35 (m, 3H). Selected NOEs are between δ 3.67 (C3–*H*) and δ 2.87 (C4–*H*), 2.97 (C5–H*H*) and between δ 2.54 (C5–*H*H), 2.97 (C5– H*H*), and δ 6.87–6.89 (Ph–*H*). Atom numbering is shown in Table 3-2. ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 14.2 (CH₃), 42.2 (CH), 46.0 (CH), 46.6 (CH₂), 47.5 (CH₂), 51.1 (CH), 61.6 (CH₂), 61.8 (CH₂), 4.0 (CH), 126.7 (CH), 127.8 (CH), 128.4 (CH), 128.69 (CH), 128.76 (CH), 128.77 (CH), 136.4 (C), 142.5 (C), 168.6 (C), 169.4 (C), 172.8 (C). Selected HMBC correlations are between δ 2.54 (C5–*H*H), 2.87 (C4–*H*), 2.87 (C4–*H*), 3.67 (C3–*H*), and δ 172.8 (C2), between δ 2.54 (C5–*H*H), 2.97 (C5–H*H*), 3.67 (C3–*H*), and δ 172.8 (C2), between δ 2.54 (C5–*H*H), 2.87 (C4–*H*), 3.67 (C3–*H*), and δ 172.8 (C2), between δ 2.54 (C5–*H*H), 2.97 (C5–H*H*), 3.67 (C3–*H*), and δ 142.2 (C4). IR (neat) 3419, 2981, 1747, 1732, 1684, 1494, 1455, 1376, 1301, 1032 cm⁻¹; MS (EI) *m/z* 439 (M⁺, 15), 393 (13), 332 (33), 174 (70), 84 (100%); HRMS (EI) *m/z* M⁺ 439.2003 (calcd for C₂₅H₂₉NO₆ 439.1995).

Transformation of 3a to 6a (Table 3-2, entry 3). To a solution of **3a** (178 mg, 0.42 mmol) in CH_2Cl_2 (0.6 mL) was added 1 M HCl/ ether (0.42 mL, 0.42 mmol). The mixture was stirred at room temperature or 20 h. The reaction mixture was concentrated under reduced pressure. The residue was purified by column chromatography over silica gel eluting with hexane–Et₂O to give **6a** (117 mg, 60%) and **4a** (45 mg, 27%).

6a: $R_f = 0.7$ (hexane–ether = 1:8); pale yellow oil; ¹H NMR (400MHz, CDCl₃) δ (ppm) 1.28 (t, J = 7.1 Hz, 3H), 1.36 (t, J = 7.1 Hz, 3H), 3.19 (dd, J = 10.4, 2.7 Hz, 1H), 3.24 (dd, J = 10.4, 7.1 Hz, 1H), 3.32 (dddd, J = 8.7, 7.1, 4.3, 2.7 Hz, 1H), 3.62 (dd, J = 10.5, 8.7 Hz, 1H), 3.85 (d, J = 10.5 Hz, 1H), 4.15–4.41 (m, 5H), 4.67 (d, J = 14.7 Hz, 1H), 4.95 (d, J = 4.3 Hz, 1H), 7.26–7.36 (m,

10H). Selected NOEs are between δ 3.62 (C3–*H*) and δ 3.32 (C4–*H*), 3.24 (C5–H*H*) and between δ 3.85 (CH(CO₂Et)₂), 3.19 (C5–*H*H), and δ 4.95 (C*H*ClPh). Atom numbering is shown in Table 3-2. ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 14.1 (CH₃), 14.2 (CH₃), 41.4 (CH), 44.7 (CH), 46.6 (CH₂), 46.9 (CH₂), 49.8 (CH), 61.95 (CH), 61.98 (CH₂), 62.2 (CH₂), 127.0 (CH), 127.8 (CH), 128.5 (CH), 128.66 (CH), 128.72 (CH), 128.8 (CH), 135.8 (C), 138.9 (C), 168.4 (C), 168.5 (C), 171.9 (C). Selected HMBC correlations are between δ 3.19 (C5–*H*H), 3.24 (C5–H*H*), 3.24 (C5–H*H*), 3.62 (C3–*H*), and δ 171.9 (*C*2), between δ 3.19 (C5–*H*H), 3.24 (C5–H*H*), 3.62 (C3–*H*), and δ 61.95 (*C*HClPh), and between δ 3.19 (C5–*H*H), 3.24 (C5–H*H*), 3.62 (C3–*H*), and δ 61.95 (*C*HClPh), and between δ 3.19 (C5–*H*H), 3.24 (C5–H*H*), 3.62 (C3–*H*), and δ 61.95 (*C*HClPh), and between δ 3.19 (C5–*H*H), 3.24 (C5–H*H*), 3.62 (C3–*H*), and δ 61.95 (*C*HClPh), and between δ 3.19 (C5–*H*H), 3.24 (C5–H*H*), 3.62 (C3–*H*), and δ 61.95 (*C*HClPh), and between δ 3.19 (C5–*H*H), 3.24 (C5–H*H*), 3.62 (C3–*H*), and δ 61.95 (*C*HClPh), and between δ 3.19 (C5–*H*H), 3.24 (C5–H*H*), 3.62 (C3–*H*), and δ 61.95 (*C*HClPh), and between δ 3.19 (C5–*H*H), 3.24 (C5–H*H*), and δ 41.4 (*C*4). IR (neat) 2981, 1747, 1732, 1689, 1604, 1495, 1447, 1371, 1028 cm⁻¹; MS (EI) *m/z* 459 (M⁺, 6.3), 457 (M⁺, 17), 332 (33), 198 (52), 72 (100%); HRMS (EI) *m/z* M⁺ 457.1655, 459.1647 (calcd for C₂₅H₂₈CINO₅ 457.1656, 459.1627).

7j: (1 mmol scale, 317 mg, 68%); $R_f = 0.3$ (hexane–ether = 1:8); colorless crystals; mp 133–134.5 °C (AcOEt); ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.25 (t, J = 7.1 Hz, 3H), 1.39 (t, J = 7.1 Hz, 3H), 2.55 (ddddd, J = 12.9, 12.1, 9.6, 7.5, 5.1 Hz, 1H), 2.90 (dd, J = 17.0, 12.1 Hz, 1H), 3.04 (d, J = 12.9 Hz, 1H), 3.07 (dd, J = 9.6, 9.3 Hz, 1H), 3.14 (dd, J = 17.0, 5.1 Hz, 1H), 3.48 (dd, J = 9.3, 7.5 Hz, 1H), 4.15 (dq, J = 10.7, 7.1 Hz, 1H), 4.28–4.51 (m, 4H), 4.68 (d, J = 14.8 Hz, 1H), 7.27–7.37 (m, 6H), 8.08 (dd, J = 8.6, 2.3 Hz, 1H), 8.30 (d, J = 2.3 Hz, 1H); Atom numbering is shown in Table 3-3. ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 13.9 (CH₃), 14.1 (CH₃), 32.3 (CH), 34.4 (CH₂), 46.6 (CH₂), 50.1 (CH), 50.2 (CH₂), 60.4 (C), 62.5 (CH₂), 63.2 (CH₂), 122.8 (CH), 126.2 (CH), 127.7 (CH), 128.2 (CH), 128.8 (CH), 130.7 (CH), 135.9 (C), 136.6 (C), 143.1 (C), 146.6 (C), 167.6 (C), 169.9 (C), 171.0 (C); ¹H NMR (400 MHz, (CD₃)₂CO) δ (ppm) 1.19 (t, J = 7.1 Hz, 3H), 1.31 (t, J = 7.1 Hz, 3H), 2.54 (ddddd, J = 12.9, 11.9, 9.6, 7.6, 5.3 Hz, 1H), 3.06 (dd, J = 17.8, 11.9 Hz, 1H), 3.11 (d, J = 12.9 Hz, 1H), 3.22 (dd, J = 9.6, 9.0 Hz, 1H), 3.26 (dd, J = 17.8, 5.3 Hz, 1H), 3.54 (dd, J = 9.0, 7.6 Hz, 1H), 4.08–4.42 (m, 5H), 4.69 (d, J = 15.0 Hz, 1H), 7.27–7.32 (m, 1H), 7.33–7.37 (m, 4H), 7.50 (d, J = 8.6 Hz, 1H), 8.11 (dd, J = 8.6, 2.3 Hz, 1H), 8.23 (d, J = 2.3 Hz, 1H). Selected NOEs are between δ 2.54

(C3a–H) and δ 3.54 (C3–H*H*), 3.26 (C4–H*H*). ¹³C NMR (100.6 MHz, (CD₃)₂CO) δ (ppm) 14.1 (CH₃), 14.3 (CH₃), 33.1 (CH), 34.7 (CH₂), 46.6 (CH₂), 50.3 (CH), 50.6 (CH₂), 61.4 (C), 62.6 (CH₂), 63.0 (CH₂), 123.2 (CH), 126.2 (CH), 128.1 (CH), 128.7 (CH), 129.3 (CH), 132.0 (CH), 137.0 (C), 138.4 (C), 145.3 (C), 147.1 (C), 168.2 (C), 170.4 (C), 171.3 (C). Selected HMBC correlations are between δ 3.06 (C4–*H*H), 3.11 (C9a–*H*), and δ 50.6 (C3), between δ 3.22 (C3–*H*H), 3.54 (C3–H*H*), and δ 50.3 (C9a), between δ 3.06 (C4–*H*H), 3.26 (C4–*HH*), 3.11 (C9a–*H*), 3.54 (C3–H*H*), and δ 33.1 (C3a), and between δ 3.11 (C9a–*H*) and δ 61.4 (C9). IR (KBr) 2982, 2936, 1747, 1732, 1699, 1520, 1347, 1255, 1190, 1098, 1029 cm⁻¹; MS (EI) *m*/*z* 466 (M⁺, 96), 363 (53), 91 (100%); HRMS (EI) *m*/*z* M⁺ 466.1734 (calcd for C₂₅H₂₆N₂O₇ 466.1740). Anal. Calcd for C₂₅H₂₆N₂O₇: C, 64.37; H, 5.62; N, 6.01. Found: C, 64.68; H, 5.34; N, 5.97.

7k: (1 mmol scale, 334 mg, 75%); $R_f = 0.2$ (hexane-ether = 1:4); colorless crystals; mp 118.5–119.5 °C (AcOEt); ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.25 (t, J = 7.1 Hz, 3H), 1.38 (t, J = 7.1 Hz, 3H), 2.52 (ddddd, J = 13.3, 12.1, 9.7, 7.4, 5.3 Hz, 1H), 2.87 (dd, J = 16.6, 12.1 Hz, 1H), 3.01 (d, J = 13.3 Hz, 1H), 3.05 (dd, J = 9.7, 9.4 Hz, 1H), 3.08 (dd, J = 16.6, 5.3 Hz, 1H), 3.47 (dd, J = 9.4, 7.4 Hz, 1H), 4.14 (dq, J = 10.7, 7.1 Hz, 1H), 4.28–4.49 (m, 4H), 4.67 (d, J = 14.8 Hz, 1H), 7.23 (d, J = 8.0 Hz, 1H), 7.27–7.37 (m, 5H), 7.50 (dd, J = 8.0, 1.7 Hz, 1H), 7.70 (d, J = 1.7 Hz, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 14.0 (CH₃), 14.1 (CH₃), 32.2 (CH), 34.5 (CH₂), 46.5 (CH₂), 50.1 (CH), 50.2 (CH₂), 60.3 (C), 62.5 (CH₂), 63.1 (CH₂), 110.7 (C), 118.6 (C), 127.7 (CH), 128.2 (CH), 128.8 (CH), 130.8 (CH), 131.1 (CH), 134.9 (CH), 135.7 (C), 136.6 (C), 141.3 (C), 167.6 (C), 170.0 (C), 171.2 (C); ¹H NMR (400 MHz, CD₃CN) δ (ppm) 1.17 (t, J = 7.0 Hz, 3H), 1.28 (t, J = 7.1 Hz, 3H), 2.44 (ddddd, J = 12.9, 11.9, 9.4, 7.6, 5.3 Hz, 1H), 2.91 (dd, J = 17.2, 51.9 Hz, 1H), 3.02 (d, J = 12.9 Hz, 1H), 3.11 (dd, J = 9.4, 9.2 Hz, 1H), 3.11 (dd, J = 17.2, 5.3 Hz, 1H), 3.44 (CH₃) (CH₃) (CH₃) (CH₃) (CH₃) (CH₃) (CH₃) (CH₃) (C), 141.3 (C), 14.6 (C), 170.0 (C), 171.2 (C); ¹H NMR (400 MHz, CD₃CN) δ (ppm) 1.17 (t, J = 7.0 Hz, 3H), 1.28 (t, J = 7.1 Hz, 3H), 2.44 (ddddd, J = 12.9, 11.9, 9.4, 7.6, 5.3 Hz, 1H), 2.91 (dd, J = 17.2, 5.3 Hz, 1H), 3.02 (d, J = 12.9 Hz, 1H), 3.11 (dd, J = 9.4, 9.2 Hz, 1H), 3.11 (dd, J = 17.2, 5.3 Hz, 1H), 3.46 (dd, J = 9.2, 7.6 Hz, 1H), 4.07 (dq, J = 10.7, 7.0 Hz, 1H), 4.19–4.38 (m, 4H), 4.62 (d, J = 15.2 Hz, 1H), 7.28–7.39 (m, 6H), 7.60 (dd, J = 8.0, 1.6 Hz, 1H), 7.66 (d, J = 1.6 Hz, 1H). Selected NOEs are between δ 2.44 (C3a–H) and δ 3.46 (C3–HH), and between δ 2.91 (C4–H) and δ 3.02

(C9a–*H*). ¹³C NMR (100.6 MHz, CD₃CN) δ (ppm) 14.2 (CH₃), 14.3 (CH₃), 32.9 (CH), 34.7 (CH₂), 46.7 (CH₂), 50.4 (CH), 51.0 (CH₂), 61.4 (C), 62.9 (CH₂), 63.4 (CH₂), 110.8 (C), 119.3 (C), 128.3 (CH), 128.8 (CH), 129.5 (CH), 132.0 (CH), 132.1 (CH), 135.3 (CH), 136.7 (C), 138.4 (C), 143.3 (C), 168.6 (C), 170.9 (C), 171.9 (C). Selected HMBC correlations are between δ 2.91 (C4–*H*H) and δ 51.0 (C3), between δ 3.46 (C3–H*H*) and δ 50.4 (C9a), δ 2.91 (C4–*H*H), 3.02 (C9a–*H*), 3.46 (C3–H*H*), and δ 32.9 (C3a), and between δ 3.02 (C9a–*H*) and δ 61.4 (C9). IR (KBr) 2981, 2937, 2229, 1742, 1730, 1696, 1496, 1442, 1366, 1252, 1190, 1029 cm⁻¹; MS (EI) *m/z* 446 (M⁺, 100), 343 (58), 149 (60), 91 (92%); HRMS (EI) *m/z* M⁺ 446.1846 (calcd for C₂₆H₂₆N₂O₅ 446.1842). Anal. Calcd for C₂₆H₂₆N₂O₅; C, 69.94; H, 5.87; N, 6.27. Found: C, 69.59; H, 5.96; N, 6.15.

71: (1 mmol scale, 342 mg, 71%); $R_f = 0.3$ (hexane-ether = 1:4); colorless crystals; mp 145–146 °C (AcOEt); ¹H NMR (400 MHz, CDCl3) δ (ppm) 1.23 (t, J = 7.1 Hz, 3H), 1.37 (t, J = 7.1 Hz, 3H), 2.53 (ddddd, J = 13.2, 12.1, 9.6, 7.4, 5.1 Hz, 1H), 2.86 (dd, J = 16.6, 12.1 Hz, 1H), 3.04 (d, J = 13.2 Hz, 1H), 3.05 (dd, J = 9.6, 9.2 Hz, 1H), 3.08 (dd, J = 16.6, 5.1 Hz, 1H), 3.46 (dd, J = 9.2, 7.4 Hz, 1H), 3.90 (s, 3H), 4.14 (dq, J = 10.7, 7.1 Hz, 1H), 4.26–4.50 (m, 4H), 4.68 (d, J = 10.7, 7.1 Hz, 1H), 4.26–4.50 (m, 4H), 4.68 (d, J = 10.7, 7.1 Hz, 1H), 4.26–4.50 (m, 4H), 4.68 (d, J = 10.7, 7.1 Hz, 1H), 4.26–4.50 (m, 4H), 4.68 (d, J = 10.7, 7.1 Hz, 1H), 4.26–4.50 (m, 4H), 4.68 (d, J = 10.7, 7.1 Hz, 1H), 4.26–4.50 (m, 4H), 4.68 (d, J = 10.7, 7.1 Hz, 1H), 4.26–4.50 (m, 4H), 4.68 (d, J = 10.7, 7.1 Hz, 1H), 4.26–4.50 (m, 4H), 4.68 (d, J = 10.7, 7.1 Hz, 1H), 4.26–4.50 (m, 4H), 4.68 (d, J = 10.7, 7.1 Hz, 1H), 4.26–4.50 (m, 4H), 4.68 (d, J = 10.7, 7.1 Hz, 1H), 4.26–4.50 (m, 4H), 4.68 (d, J = 10.7, 7.1 Hz, 1H), 4.26–4.50 (m, 4H), 4.68 (d, J = 10.7, 7.1 Hz, 1H), 4.26–4.50 (m, 4H), 4.68 (d, J = 10.7, 7.1 Hz, 1H), 4.26–4.50 (m, 4H), 4.68 (d, J = 10.7, 7.1 Hz, 1H), 4.26–4.50 (m, 4H), 4.68 (d, J = 10.7, 7.1 Hz, 1H), 4.26–4.50 (m, 4H), 4.68 (d, J = 10.7, 7.1 Hz, 1H), 4.26–4.50 (m, 4H), 4.68 (d, J = 10.7, 7.1 Hz, 1H), 4.26–4.50 (m, 4H), 4.68 (d, J = 10.7, 7.1 Hz, 1H), 4.26–4.50 (m, 4H), 4.68 (d, J = 10.7, 7.1 Hz, 1H), 4.58 (m, 4H), 4.68 (m, 4H), 14.8 Hz, 1H), 7.19 (d, J = 8.1 Hz, 1H), 7.26–7.36 (m, 5H), 7.89 (dd, J = 8.1, 1.7 Hz, 1H), 8.09 (d, J = 1.7 Hz, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 13.9 (CH₃), 14.1 (CH₃), 32.4 (CH), 34.4 (CH₂), 46.5 (CH₂), 50.28 (CH), 50.32 (CH₂), 52.2 (CH₃), 60.4 (C), 62.2 (CH₂), 62.7 (CH₂), 127.6 (CH), 128.2 (CH), 128.6 (C), 128.7 (CH), 128.9 (CH), 129.9 (CH), 132.2 (CH), 134.5 (C), 136.7 (C), 140.8 (C), 166.6 (C), 168.2 (C), 170.4 (C), 171.5 (C); ¹H NMR (400 MHz, CD₃CN) δ (ppm) 1.16 (t, *J* = 7.1 Hz, 3H), 1.28 (t, *J* = 7.1 Hz, 3H), 2.44 (ddddd, *J* = 13.1, 11.9, 9.8, 7.6, 5.3 Hz, 1H), 2.89 (dd, J = 17.0, 11.9 Hz, 1H), 3.04 (d, J = 13.1 Hz, 1H), 3.11 (dd, J = 17.0, 5.3 Hz, 1H), 3.11 (dd, J = 9.8, 9.2 Hz, 1H), 3.47 (dd, J = 9.2, 7.6 Hz, 1H), 3.87 (s, 3H), 4.07 (dq, J = 10.7, 7.1 Hz), 3.87 (s, 3H), 4.07 (dq, J = 10.7, 7.1 Hz)1H), 4.17-4.38 (m, 5H), 4.62 (d, J = 15.0 Hz, 1H), 7.28-7.33 (m, 4H), 7.36-7.39 (m, 2H), 7.86(dd, J = 8.0, 1.8 Hz, 1H), 7.94 (d, J = 1.8 Hz, 1H). Selected NOEs are between $\delta 2.44$ (C3a–H) and δ 3.47 (C3–HH). ¹³C NMR (100.6 MHz, CD₃CN) δ (ppm) 14.2 (CH₃), 14.4 (CH₃), 33.1 (CH), 34.6 (CH₂), 46.7 (CH₂), 50.6 (CH), 51.1 (CH₂), 52.8 (CH₃), 61.6 (C), 62.8 (CH₂), 63.1 (CH₂), 128.3 (CH), 128.8 (CH), 129.2 (C), 129.4 (CH), 129.6 (CH), 131.3 (CH), 132.4 (CH), 135.8 (C), 138.4 (C), 142.8 (C), 167.2 (C), 169.1 (C), 171.3 (C), 172.1 (C). Selected HMBC correlations are between δ 2.89 (C4–*H*H) and δ 51.1 (*C*3), between δ 3.47 (C3–H*H*) and δ 50.6 (C9a), between δ 2.89 (C4–*H*H), 3.47 (C3–H*H*), and δ 33.1 (*C*3a), and between δ 3.04 (C9a–*H*) and δ 61.6 (*C*9). IR (KBr) 2984, 2918, 1749, 1726, 1686, 1613, 1483, 1431, 1254, 1191, 1138, 1023 cm⁻¹; MS (FAB) *m*/*z* 502 ([M + Na]⁺), 480 ([M +H]⁺); HRMS (FAB) *m*/*z* [M + H]⁺ 480.2026 (calcd for C₂₇H₃₀NO₇ 480.2022), [M + Na]⁺ 502.1856 (calcd for C₂₇H₂₉NO₇Na 502.1842). Anal. Calcd for C₂₇H₂₉NO₇: C, 67.63; H, 6.10; N, 2.92. Found: C, 67.58; H, 6.12; N, 2.89.

7m: (1 mmol scale, 282 mg, 57%); $R_f = 0.4$ (hexane-ether = 1:8); colorless crystals; mp 128–129.5 °C (AcOEt); ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.23 (t, J = 7.1 Hz, 3H), 1.37 (t, J = 7.1 Hz, 3H), 1.38 (t, J = 7.1 Hz, 3H), 2.53 (ddddd, J = 12.9, 12.1, 9.8, 7.5, 5.1 Hz, 1H), 2.86 (dd, J = 16.6, 12.1 Hz, 1H), 3.05 (d, J = 12.9 Hz, 1H), 3.05 (dd, J = 9.8, 9.3 Hz, 1H), 3.08 (dd, J = 12.9 Hz, 1H), 3.05 (dd, J = 12.9 Hz, 1H), 3.08 (dd, J = 1 16.6, 5.1 Hz, 1H), 3.46 (dd, J = 9.3, 7.5 Hz, 1H), 4.13 (dq, J = 10.7, 7.1 Hz, 1H), 4.27–4.51 (m, 6H), 4.68 (d, J = 14.8 Hz, 1H), 7.19 (d, J = 8.0 Hz, 1H), 7.26–7.36 (m, 5H), 7.89 (dd, J = 8.0, 1.8) Hz, 1H), 8.10 (d, J = 1.8 Hz, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 13.9 (CH₃), 14.1 (CH₃), 14.3 (CH₃), 32.3 (CH), 34.3 (CH₂), 46.5 (CH₂), 50.2 (CH), 50.3 (CH₂), 60.4 (C), 61.0 (CH₂), 62.1 (CH₂), 62.7 (CH₂), 127.6 (CH), 128.2 (CH), 128.7 (CH), 128.85 (CH), 128.92 (C), 129.8 (CH), 132.1 (CH), 134.5 (C), 136.7 (C), 140.6 (C), 166.1 (C), 168.2 (C), 170.4 (C), 171.6 (C); ¹H NMR (400 MHz, CD₃CN) δ (ppm) 1.16 (t, J = 7.1 Hz, 3H), 1.28 (t, J = 7.1 Hz, 3H), 1.35 (t, J = 7.0 Hz, 3H), 2.44 (ddddd, J = 13.1, 11.9, 9.6, 7.4, 5.3 Hz, 1H), 2.89 (dd, J = 16.8, 11.9 Hz, 1H), 3.04 (d, J = 13.1 Hz, 1H), 3.11 (dd, J = 9.6, 9.2 Hz, 1H), 3.11 (dd, J = 16.8, 5.3 Hz, 1H), 3.47 (dd, J = 9.2, 7.4 Hz, 1H), 4.07 (dq, J = 10.7, 7.1 Hz, 1H), 4.18–4.39 (m, 6H), 4.62 (d, J =15.0 Hz, 1H), 7.28–7.33 (m, 4H), 7.35–7.39 (m, 2H), 7.87 (dd, J = 8.0, 1.8 Hz, 1H), 7.95 (d, J = 1.8 Hz, 1H). Selected NOEs are between δ 2.44 (C3a–H) and δ 3.47 (C3–HH), and between δ 2.89 (C4–*H*H) and δ 3.04 (C9a–*H*). ¹³C NMR (100.6 MHz, CD₃CN) δ (ppm) 14.9 (CH₃), 15.0 (CH₃), 15.2 (CH₃), 33.8 (CH), 35.3 (CH₂), 47.4 (CH₂), 51.2 (CH), 51.7 (CH₂), 62.2 (C), 62.5 (CH₂), 63.4 (CH₂), 63.8 (CH₂), 128.9 (CH), 129.4 (CH), 130.0 (CH), 130.2 (CH), 131.9 (CH), 133.0 (CH), 136.4 (C), 139.1 (C), 143.3 (C), 167.3 (C), 169.7 (C), 172.0 (C), 172.8 (C). Selected HMBC correlations are between δ 2.89 (C4–*H*H) and δ 51.7 (*C*3), between δ 3.47 (C3–H*H*), 2.89 (C4–*H*H), and δ 51.2 (*C*9a), between δ 2.89 (C4–*H*H), 3.04 (C9a–*H*), 3.47 (C3–H*H*), and δ 33.8 (*C*3a), and between δ 3.04 (C9a–*H*) and δ 62.2 (*C*9). IR (KBr) 2983, 1728, 1611, 1482, 1443, 1366, 1280, 1259, 1193, 1027 cm⁻¹; MS (EI) *m/z* 493 (M⁺, 100), 390 (72), 91 (55%); HRMS (EI) *m/z* M⁺ 493.2094 (calcd for C₂₈H₃₁NO₇ 493.2101).

7n: (0.5 mmol scale, 125 mg, 51%); $R_f = 0.7$ (ether); colorless crystals; mp 124–125 °C; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.23 (t, J = 7.1 Hz, 3H), 1.35 (t, J = 7.1 Hz, 3H), 2.54 (ddddd, J = 14.1, 12.2, 9.8, 7.5, 5.3 Hz, 1H), 2.87 (dd, J = 16.5, 12.2 Hz, 1H), 3.04 (d, J = 14.1Hz, 1H), 3.06 (dd, J = 9.8, 9.3 Hz, 1H), 3.08 (dd, J = 16.5, 5.3Hz, 1H), 3.47 (dd, J = 9.3, 7.5 Hz, 1H), 4.13 (dq, J = 10.7, 3.1 Hz, 1H), 4.26-4.47 (m, 4H), 4.69 (d, J = 14.8 Hz, 1H), 7.24 (d, J = 8.0 Hz, 1H),7.27–7.37 (m, 5H), 7.48 (dd, J = 8.0, 1.4 Hz, 1H), 7.67 (bs, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 13.8 (CH₃), 14.1 (CH₃), 32.4 (CH), 34.2 (CH₂), 46.5 (CH₂), 50.2 (CH), 50.3 (CH₂), 60.4 (C), 62.3 (CH₂), 62.8 (CH₂), 124.0 (C, q, $J_{CF} = 272$ Hz), 124.7 (CH, q, $J_{CF} = 3.8$ Hz), 127.6 (CH), 127.9 (CH, q, $J_{CF} = 3.8$ Hz), 128.2 (CH), 128.8 (CH), 128.9 (C, q, $J_{CF} = 33$ Hz), 130.3 (CH), 134.9 (C), 136.6 (C), 139.7 (C), 167.9 (C), 170.2 (C), 171.4 (C); ¹⁹F NMR (376 MHz, CDCl₃) δ (ppm) 62.71; ¹H NMR (400 MHz, CD3CN) δ (ppm) 1.16 (t, J = 7.0 Hz, 3H), 1.27 (t, J = 7.1 Hz, 3H), 2.46 (ddddd, J = 13.1, 12.1, 9.6, 7.4, 5.1 Hz, 1H), 2.91 (dd, J = 16.8, 12.1 Hz, 1H), 3.05 (d, J = 13.1 Hz, 1H), 3.12 (dd, J = 9.6, 9.2 Hz, 1H), 3.12 (dd, J = 16.8, 5.1 Hz, 1H), 3.48 (dd, J = 9.2, 7.4 Hz, 1H), 4.07 (dq, J = 10.7, 7.0 Hz, 1H), 4.18–4.37 (m, 4H), 4.63 (d, J = 15.0 Hz, 1H), 7.28–7.40 (m, 6H), 7.58 (d, J = 8.4 Hz, 1H), 7.59 (s, 1H). Selected NOEs are between δ 2.46 (C3a-H) and δ 3.48 (C3-HH), and between δ 2.91 (C4-HH) and δ 3.05 (C9a-H). ¹³C NMR (100.6 MHz, CD₃CN) δ (ppm) 14.1 (CH₃), 14.3 (CH₃), 33.1 (CH), 34.5 (CH₂), 46.7 (CH₂), 50.5 (CH), 51.0 (CH₂), 61.5 (C), 62.9 (CH₂), 63.2 (CH₂), 125.2 (C, q, $J_{CF} = 271$ Hz), 125.4 (CH, q, $J_{CF} = 3.8$ Hz), 128.2 (CH, q, $J_{CF} = 4.6$ Hz), 128.3 (CH), 128.7 (C, q, $J_{CF} = 32$ Hz), 128.8 (CH), 129.6 (CH), 131.9 (CH), 136.3 (C), 138.4 (C), 142.2 (C), 168.8 (C), 171.1 (C), 172.0 (C). Selected HMBC correlations are between δ 2.91 (C4–*H*H) and δ 51.0 (*C*3), between δ 3.48 (C3–H*H*), 2.91 (C4–*H*H), and δ 50.5 (*C*9a), between δ 2.91 (C4–*H*H), 3.05 (C9a–*H*), 3.48 (C3–H*H*), and δ 33.1 (*C*3a), and between δ 3.05 (C9a–*H*) and δ 61.5 (*C*9). IR (KBr) 2927, 1747, 1726, 1699, 1334, 1261, 1162, 1128 cm⁻¹; MS (EI) *m/z* 489 (M⁺, 25), 386 (15), 333 (14), 242 (29), 226 (36), 200 (100%); HRMS (EI) *m/z* M⁺ 489.1772 (calcd for C₂₆H₂₆F₃NO₅ 489.1763).

3n: (0.5 mmol scale, 14 mg, 6%); $R_f = 0.4$ (ether); colorless crystals; mp 164–165 °C; ¹H NMR $(400 \text{ MHz}, \text{CDCl}_3) \delta$ (ppm) 1.30 (t, J = 7.0 Hz, 3H), 1.34 (t, J = 7.1 Hz, 3H), 2.36 (ddd, J = 10.9, 6.6, 5.9 Hz, 1H), 2.64 (d, J = 10.9 Hz, 1H), 3.33 (dd, J = 10.9, 5.9 Hz, 1H), 3.82 (d, J = 14.3 Hz, 1H), 3.91 (d, J = 6.6 Hz, 1H), 4.03–4.16 (m, 2H), 4.24–4.37 (m, 3H), 5.00 (d, J = 14.3 Hz, 1H), 6.79 (d, J = 8.1 Hz, 2H), 7.30–7.32 (m, 2H), 7.40–7.44 (m, 3H), 7.49 (d, J = 8.1 Hz, 2H). Selected NOEs are between δ 2.36 (C5–H) and δ 3.33 (C4–HH), 6.79 (Ar-H), 3.91 (C1–H) and between δ 3.33 (C4–HH) and δ 3.91 (C1–H). ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 14.5 (CH₃), 15.0 (CH₃), 36.3 (CH), 41.1 (CH), 44.4 (CH₂), 46.3 (CH₂), 60.1 (CH₂), 65.0 (CH₂), 79.0 (CH), 79.7 (C), 123.8 (C, q, J_{CF} = 272 Hz), 125.7 (CH, q, J_{CF} = 3.8 Hz), 127.7 (CH), 128.1 (CH), 129.1 (CH), 129.2 (CH), 131.3 (C, q, $J_{CF} = 33$ Hz), 136.7 (C), 140.7 (C), 162.7 (C), 167.1 (C), 172.8 (C). Selected HMBC correlations are between δ 2.36 (C5–*H*), 2.64 (C4–*H*H), 3.91 (C1–*H*), and δ 172.8 (C2), between δ 2.36 (C5–H), 2.64 (C4–HH), 3.33 (C4–HH), 3.91 (C1–H), and δ 79.0 (C6), between δ 2.64 (C4–HH) and δ 41.1 (C1), and between δ 2.64 (C4–HH), 3.91 (C1–H), and δ 36.3 (C5). ¹⁹F NMR (376 MHz, CDCl₃) δ (ppm) –62.85; IR (KBr) 2984, 2931, 1699, 1668, 1621, 1327, 1164, 1124, 1068, 1020 cm⁻¹; MS (EI) m/z 489 (M⁺, 21), 291 (43), 205 (92), 200 (63), 91 (100%); HRMS (EI) *m/z* M⁺ 489.1789 (calcd for C₂₆H₂₆F₃NO₅ 489.1763).

13i: (1 mmol scale, 261 mg, 57%, including a small amount of impurity); $R_f = 0.5$ (hexane–ether = 1:2); colorless oil; ¹H NMR (400 MHz, CDCl₃) (2 rotamers, ratio 1:1) δ (ppm) 3.95 (dd, J = 6.2, 1.4 Hz, 2H × 0.5), 4.20 (dd, J = 6.6, 1.0 Hz, 2H × 0.5), 4.50 (s, 2H × 0.5), 4.74 (s, 2H × 0.5), 5.90 (dt, J = 15.6, 6.2 Hz, 1H × 0.5), 6.06 (dt, J = 15.6, 6.6 Hz, 1H × 0.5), 6.95 (d, J = 15.6 Hz, 1H × 0.5), 6.98 (d, J = 15.6 Hz, 1H × 0.5), 7.15 (s, 1H × 0.5), 7.23–7.54 (m, 7H+1H × 0.5), 7.58–7.62 (m, 1H), 7.98 (dd, J = 8.2, 1.2 Hz, 1H × 0.5), 8.00 (dd, J = 8.2, 1.2 Hz, 1H × 0.5); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 46.1 (CH₂), 47.7 (CH₂), 49.1 (CH₂), 51.1 (CH₂), 120.1 (C, q, $J_{CF} = 275$ Hz), 120.3 (C, broad q, $J_{CF} = 275$ Hz), 124.7 (CH), 123.45–124.46 (C, m), 124.8 (CH), 127.4 (CH), 127.5 (CH), 127.6 (CH), 128.1 (CH), 128.6 (CH), 128.7 (CH), 128.8 (CH), 128.95 (CH), 129.00 (CH), 129.1 (CH), 129.3 (CH), 130.35 (CH), 130.43 (CH), 131.9 (C), 132.3 (C), 133.5 (CH), 133.6 (CH), 134.5 (C), 135.7 (C), 136.0 (CH, m), 136.2 (CH, m), 147.59 (C), 147.63 (C), 162.6 (C), 162.7 (C); ¹⁹F NMR (376 MHz, CDCl₃) δ (ppm) –66.43 (q, $J_{FF} = 6.5$ Hz), -66.65 (q, $J_{FF} = 6.5$ Hz), -69.89 (q, $J_{FF} = 6.5$ Hz), -69.89 (q, $J_{FF} = 6.5$ Hz), -69.89 (q, $J_{FF} = 6.5$ Hz), 128. (CH), 133.5 (CH), 134.5 (C), 135.7 (C), 136.0 (CH, m), 136.2 (CH, m), 147.59 (C), 147.63 (C), 162.6 (C), 162.7 (C); ¹⁹F NMR (376 MHz, CDCl₃) δ (ppm) –66.43 (q, $J_{FF} = 6.5$ Hz), -66.65 (q, $J_{FF} = 6.5$ Hz), -69.89 (q, $J_{FF} = 6.5$ Hz), -69.89 (R) $J_{FF} = 6.5$ Hz), -69.99 (R) $J_{FF} = 6.5$ Hz); IR (neat) 3068, 3032, 2931, 1651, 1608, 1524, 1435, 1386, 1348, 1286, 12

Typical experimental procedure for Table 3-5 (entry 5). A solution of **13i** (261 mg, 0.57 mmol) in toluene (1.0 mL) was heated at 110 °C for 20 h. The mixture was concentrated *in vacuo*. The residue was purified by column chromatography over silica gel with hexane–ether as eluent to give **16i** (231 mg, 89%).

16i: (0.57 mmol scale, 231 mg, 89%); $R_f = 0.2$ (hexane–ether = 2:1); colorless crystals; mp 219–220 °C (AcOEt); ¹H NMR (400 MHz, CDCl₃) δ (ppm) 2.64 (ddddd, J = 13.5, 12.1, 9.6, 7.2, 4.3 Hz, 1H), 2.81 (dd, J = 13.5, 1.2 Hz, 1H), 2.95 (dd, J = 17.2, 12.1 Hz, 1H), 3.06 (dd, J = 9.6, 9.4 Hz, 1H), 3.12 (dd, J = 17.2, 4.3 Hz, 1H), 3.28 (dd, J = 9.4, 7.2 Hz, 1H), 4.44 (d, J = 14.8 Hz, 1H), 4.62 (d, J = 14.8 Hz, 1H), 7.25 (d-like, J = 7.4 Hz, 2H), 7.28–7.37 (m, 3H), 7.51 (dd, J = 8.4, 8.0 Hz, 1H), 7.90 (dd, J = 8.0, 1.2 Hz, 1H), 8.11 (d. J = 8.4 Hz, 1H). Selected NOEs are between

δ 2.64 (C3a–*H*) and δ 3.28 (C3–H*H*) and between δ 2.95 (C4–*H*H) and δ 2.81 (C9a–*H*). ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 30.9 (CH₂), 32.4 (CH, q, *J*_{CF} = 2.3 Hz), 46.6 (CH), 47.2 (CH₂), 48.7 (CH₂), 56.9 (C, septet, *J*_{CF} = 27 Hz), 123.6 (C, q, *J*_{CF} = 288 Hz), 124.3 (C, q, *J*_{CF} = 285 Hz), 125.5 (CH), 127.6 (CH), 128.0 (CH), 128.3 (CH), 129.0 (CH), 129.5 (C), 132.9 (C), 135.5 (CH, septet, *J*_{CF} = 3.8 Hz), 136.0 (C), 151.1 (C), 167.4 (C). Selected HMBC correlations are between δ 3.28 (C3–H*H*) and 2.95 (C4–*H*H), between δ 46.6 (*C*9a), δ 2.95 (C4–*H*H), 3.28 (C3–H*H*), and δ 32.4 (*C*3a), and between δ 2.81 (C9a–*H*) and δ 56.9 (*C*9). ¹⁹F NMR (376 MHz, CDCl₃) δ (ppm) –66.23 (q, *J*_{FF} = 6.9 Hz), -70.27 (q, *J*_{FF} = 6.9 Hz); IR (KBr) 3033, 2929, 1699, 1530, 1431, 1349, 1263, 1245, 1195, 1080 cm⁻¹;MS (EI) *m*/*z* 458 (M⁺, 71), 91 (100%); HRMS (EI) *m*/*z* 458.1064 (calcd for C₂₁H₁₆F₆N₂O₃ 458.1065). Anal. Calcd for C₂₁H₁₆F₆N₂O₃: C, 55.03; H, 3.52; N, 6.11. Found: C, 55.01; H, 3.55; N, 6.15.

References

- ¹ (a) T. Bach, C. Pelkmann, K. Harms, *Tetrahedron Lett.* 1999, 40, 2103. (b) T. Bach, C. Krüger,
 K. Harms, Synthesis 2000, 2000, 305.
- ² (a) L, H, Klemm. T. M. McGuire, K. W. Gopinath, *J. Org. Chem.* **1976**, 41, 15, 2571. (b) T, Ozawa, T. Kurahashi, S. Matsubara, *Org. Lett.* **2011**, 13, 19, 5390.
- ³ (a) C. Ko, J. B. Feltenberger, S. K. Ghosh, R. Hsung, *P. Org. Lett.* 2008, 10, 1971. (b) S. Nakano, K. Kakugawa, T. Nemoto, Y. Hamada, *Adv. Synth. Catal.* 2014, 356, 2088. (c) M. R. Luzung, P. Mauleón, F. D. Toste, *J. Am. Chem. Soc.* 2007, 129, 12402. (d) B. Ranieri, C. Obradors, M. Mato, A. M. Echavarren, *Org. Lett.* 2016, 18, 1614. (e) M. Gulías, A. Collado, B. Trillo, F. López, E. Oñate, M. A. Esteruelas, J. L. Mascareñas, *J. Am. Chem. Soc.* 2011, 133, 7660.
 ⁴ B. B. Snider, D. M. Roush. *J. Org. Chem.* 1979, 44, 24, 4229.
- ⁵ (a) S. Yamazaki, Y. Yamamoto, Y. Fukushima, M. Takebayashi, T. Ukai, Y. Mikata, J. Org. Chem. 2010, 75, 5216. (b) S. Yamazaki, M. Takebayashi, K. Miyazaki, J. Org. Chem. 2010, 75, 1188. (c) S. Yamazaki, Y. Iwata, Y. Fukushima, Org. Biomol. Chem. 2009, 7, 655. (d) S. Morikawa,

S. Yamazaki, M. Tsukada, S. Izuhara, T. Morimoto, K. Kakiuchi, J. Org. Chem. 2007, 72, 6459. (e)
S. Yamazaki, Y. Iwata, J. Org. Chem. 2006, 71, 739. (f) S. Yamazaki, K. Ueda, Y. Fukushima, A. Ogawa, K. Kakiuchi, Eur. J. Org. Chem. 2014, 2014, 7023. (g) S. Yamazaki, Y. Maenaka, K. Fujinami, Y. Mikata, RSC Adv. 2012, 2, 8095. (h) S. Yamazaki, K. Fujinami, Y. Maitoko, K. Ueda, K. Kakiuchi, J. Org. Chem. 2013, 78, 8405. (i) Y. Fukushima, S. Yamazaki, A. Ogawa, Org. Biomol. Chem. 2014, 12, 3964. (j) S. Yamazaki, J. Wada, K. Kakiuchi, Can. J. Chem. 2015, 93, 1122.
⁶ S. Yamazaki, H. Sugiura, M. Niina, Y. Mikata, A. Ogawa, Heterocycles 2016, 92, 485.

⁷ S. Yamazaki, H. Sugiura, S. Ohashi, K. Ishizuka, R. Saimu, Y. Mikata, A. Ogawa, *J. Org. Chem.* **2016**, 81, 10863.

⁸ The byproducts by the reaction of ClCH₂CH₂Cl and HOBt in the presence of Et₃N were formed and removed by column chromatography. (a) J.-G. Ji, D.-Y. Zhang, Y.-H. Ye, Q.-Y. Xing, *Tetrahedron Lett.* **1998**, 39, 6515. (b) W. A. Feld, D. G. Evans, *J. Chem. Eng. Data* **1983**, 28, 138.

⁹ S. Yamazaki, K. Ohnitsu, K. Ohi, T. Otsubo, K. Moriyama, Org. Lett. 2005, 7, 759.

- ¹⁰ (a) H. Anan, A. Tanaka, R. Tsuzuki, M. Yokota, T. Yatsu, T. Fujikura, *Chem. Pharm. Bull.* 1996, 44, 1865. (b) A. F. Abdel-Magid, K. G. Carson, B. D. Harris, C. A. Maryanoff, R. D. Shah, *J. Org. Chem.* 1996, 61, 3849.
- ¹¹ K. K. Singh, C. S. Mathela, *Indian J. Chem.* **2014**, 53B, 907.
- ¹² Nishimura, T. Bull. Chem. Soc. Jpn. **1952**, 25, 54.
- ¹³ G. Battistuzzi, S. Cacchi, G. Fabrizi, Org. Lett. 2003, 5, 777.
- ¹⁴ J. Zhu, J. Liu, R. Ma, H. Xie, J. Li, H. Jiang, W. Wang, Adv. Synth. Catal. 2009, 351, 1229.
- ¹⁵ (a) H. Saikachi, Y. Taniguchi, H. Ogawa, Yakugaku Zasshi 1963, 83, 578; Chem. Abstr. 1963,
- 59, 11397. (b) T. M. Cresp, M. V. Sargent, P. Vogel, J. Chem. Soc., Perkin Trans. 1 1974, 37. (c)
- R. Olstein, E. F. M. Stephenson, Aust. J. Chem. 1979, 32, 681.

Chapter 4

SequentialIntramolecularCyclizationofDiarylpropenamineswithElectron-DeficientCarboxylic Acids

4-1 Introduction

The intramolecular Diels-Alder reaction (IMDA) between alkenes and dienes is a powerful tool for the facile construction of multicyclic skeletons. ¹ The IDMA reaction of vinyl benzene (styrene) as a diene requires relatively harsh conditions, because of involving dearomatization of benzene ring.²

In chapter 3, intramolecular [2 + 2] and [4 + 2] cycloaddition (IMDA) reactions of 1,1-diethyl 2-hydrogen ethenetricarboxylate with *E*-cinnamylamines under the amide formation conditions in sequential processes were described.³ In addition to the cycloaddition reactions of styrenes, the inter- and intramolecular reactions of diarylethenes were studied.^{2c, 4} Furthermore, one of the important selectivities of the IMDA reaction in general are formation of *trans*- or *cis*-fused ring,^{1b,5} and the investigation on the stereoselectivity of the IMDA reaction of styrenes with various substituents is of considerable interest.^{2c, 2k}

In this chapter, the sequential amide formation/IMDA/H⁺-shift reactions of electron-deficient alkenyl carboxylic acids with Z-cinnamyl amines and various 3,3-diaryl-2-propen-1-amines have been studied. This reaction proceeds to give tricyclic compounds, functionalized hexahydrobenzo[f]isoindoles. Some biologically active compounds such as podophyllotoxin and 4-phenylbenzo[f]isoindoles have the skeletons of the [4 + 2] cycloaddition products of 3,3-diaryl-2-propen-1-amines. In order to understand the factors to

control *cis*- and *trans*-fused stereochemistry in the present results, DFT calculations have been carried out.

4-2 Results and Discussion

In chapter 3, reactions of 1,1-diethyl 2-hydrogen ethenetricarboxylate 1 and *E*-cinnamylamines bearing *p*-H, halogen and MeO groups in the presence of EDCI/HOBt/Et₃N at room temperature to give cyclobutane-fused pyrrolidines are described.³ The products may be formed via amide formation/intramolecular [2 + 2] cycloaddition. First, reaction with *Z*-cinnamylamines has been examined. Reaction with benzyl or cyclohexylmethyl cinnamylamines **2a,b** (*Z*:*E* = ca. 5:1) in the presence of EDCI/HOBt/Et₃N in THF at room temperature gave a complex mixture and the possible [2 + 2] cycloadducts were not detected. The reaction at 80-110 °C gave [4 + 2] cycloadducts, *cis*-fused tricyclic compounds **3a,b** as the major products (Table 4-1). The *cis*-fused stereochemistry of **3a,b** was determined by NOEs in C₆D₆.

$EtO_{2}C CO_{2}Et + HN + H$									
Entry	2	R	Solvent	Temp.	Product	Yield (%)			
1	2a ^a	CH ₂ Ph	THF	r.t.	b	-			
2	2a ^a	CH ₂ Ph	Benzene	80°C	3a	62			
3	2a ^a	CH ₂ Ph	Toluene	110 °C	3a	58			
4	2b ^a	CH ₂ Cyclohexyl	THF	r.t.	b	-			
5	2b ^a	CH ₂ Cyclohexyl	Benzene	80 °C	3b	50			
6	2b ^a	CH ₂ Cyclohexyl	Toluene	110 °C	3b	61			
a 7. F -	- 5.1	^b A complex mix	turo						

 Table 4-1. Reactions of 1,1-diethyl 2-hydrogen ethenetricarboxylate 1 and Z-cinnamylamines 2.

^a Z:E = 5:1. ^b A complex mixture.

Next, the reaction of **1** and 3,3-diaryl-2-propenylamines **4** in the presence of the amide condensation reagents has been examined (Table 4-2). The influence of the second aromatic ring on stereochemistry by both steric and electronic effects is of mechanistic and synthetic interests.

3,3-Diaryl-2-propen-1-amines 4 were prepared from the corresponding alcohols.⁶ The reaction of 1 and N-benzyl 3,3-diphenyl-2-propen-1-amine 4a with EDCI/HOBt/Et₃N in THF at room temperature gave a complex mixture. However, the reaction in DMF gave trans-fused hexahydrobenzo[f] isoindole **5a**-trans in 52% yield via [4 + 2] cycloaddition. Interestingly, the reaction of **4a,b** in benzene or toluene at 80 110 °C gave or cis-fused hexahydrobenzo[f]isoindoles 5a,b-cis as major products. Similarly, the reaction of N-benzyl-3,3-bis(4-fluorophenyl)-2-propenylamine 4c in DMF at room temperature gave *trans*-fused tricyclic product **5***c-trans* in 53% yield and the reaction of **4***c* in benzene or toluene at 80 or 110 °C gave cis-fused product 5c-cis in 87 and 98% yields, respectively. On the other hand, reaction of 1 and 4,4'-dichloro and 3,3'-di(trifluoromethyl) derivatives 4d,e in THF, DMF and benzene at room temperature, 80 and 110 °C gave trans-fused tricyclic products 5d,e-trans in high yields. Product 5e-trans from 3,3'-CF3 substituted amine substrate 4e was obtained as a single regioisomer (3,3'-CF₃). The reaction of 1 and 4,4'-dimethyl derivative 4f in DMF gave a complex mixture at room temperature and the reaction in DMF, benzene and toluene at 80 and 110 °C gave cis-fused tricyclic product 5f-cis selectively. The relative configurations of 5 were determined by NOEs in CDCl₃ or C₆D₆. The reaction of 1 and 4,4'-dimethoxy derivative 4g gave a complex mixture in THF, DMF, benzene and toluene at room temperature, 80 and 110 °C.

Thus, the reaction with 3,3-diaryl-2-propen-1-amines gave *cis*- and *trans*-fused tricyclic compounds selectively, depending on the substituents on the benzene ring, reaction temperature and solvent.

EtO ₂ C HO ₂ C	CO ₂ Et +	X 3 2 1 R N H 4	$\begin{array}{c} 4\\ \\ 1\\ \\ 2'\\ 3' \end{array}$	EtO ₂ C O H R-N H 5- <i>trans</i> 4	$CO_{2}Et$ X $1'$ $2'$ $3'$ X	EtO ₂ C O H R-N H 5-cis	2 CO ₂ Et 4' 1' $2'3'4'3'4'3'4'3'4'3'3'4'3'$
Entry	4 ^a	Х	R	Solvent	Temp.	Product	Yield (%)
1	4a	Н	CH ₂ Ph	THF	r.t.	b	-
2	4 a	Н	CH ₂ Ph	DMF ^c	r.t.	5a <i>-trans</i>	52
3	4 a	Н	CH ₂ Ph	Benzene	80 °C	5a- <i>cis</i>	48 ^d
4	4b	Н	CH ₂ Cyclohexyl	Benzene	80 °C	5b- <i>cis</i>	55 ^d
5	4c	4,4'-F	CH ₂ Ph	THF	r.t.	e	-
6	4c	4,4'-F	CH ₂ Ph	DMF ^c	r.t.	5c-trans	53
7	4c	4,4'-F	CH ₂ Ph	Benzene	80 °C	5c-cis	87
8	4c	4,4'-F	CH ₂ Ph	Toluene	110 °C	5c- <i>cis</i>	98
9	4d	4,4'-Cl	CH ₂ Ph	THF	r.t.	5d-trans	89
10	4d	4,4'-Cl	CH ₂ Ph	$\mathrm{DMF}^{\mathrm{f}}$	r.t.	5d-trans	93
11	4d	4,4'-Cl	CH ₂ Ph	Benzene	80 °C	5d-trans	67
12	4 e	3,3'-CF ₃	CH ₂ Ph	THF	r.t.	5e-trans	99
13	4 e	3,3'-CF ₃	CH ₂ Ph	$\mathrm{DMF}^{\mathrm{f}}$	r.t.	5e-trans	93
14	4 e	3,3'-CF ₃	CH ₂ Ph	Benzene	80 °C	5e-trans	93
15	4f	4,4'-Me	CH ₂ Ph	Toluene	110 °C	5f- <i>cis</i>	48 ^g
16	4f	4,4'-Me	CH ₂ Ph	DMF	110 °C	5f-cis	55

 Table 4-2. Reactions of 1,1-diethyl 2-hydrogen ethenetricarboxylate 1 and

3,3-diaryl-2-propen-1-amines 4.

^a The reaction of 4g (X = 4,4'-OMe) gave a complex mixture. ^b A complex mixture. ^c The reaction in DMF at 80 or 110 °C gave a mixture of 5-*cis* and 5-*trans*. ^d The reaction in toluene at 110 °C gave 5-*cis* in lower yields (5a-*cis* 32%, 5b-*cis* 26%) and a complex mixture. ^e A mixture containing 5c-*trans*. ^f The reaction in DMF at 80 or 110 °C also gave 5-*trans* in 89-95% yields. ^g The reaction in benzene at 80 °C gave 5f-*cis* in lower yield (17%) and a complex mixture.

To obtain some insights into the mechanism, the reactions with dissymmetrically substituted 3,3-diaryl-2-propen-1-amines have been studied (Table 4-3). Reaction of **1** with (*Z*) and (*E*)-3-(2- or 4-chlorophenyl)-3-phenyl-2-propen-1-amines **4h-j** in the presence of EDCI/HOBt/Et₃N at room temperature gave *trans*-fused hexahydrobenzo[*f*]isoindoles **5h,i,j**-*trans* stereoselectively in 30-86% yields (entries 1,2,5,6,11). At higher temperature in benzene or toluene, *cis*-fused hexahydrobenzo[*f*]isoindoles **5h,i,j**-*cis* were also formed (entries 3,4,8,9,13,14). Reaction of **1** with (*Z*)-3-(4-nitrophenyl)-3-phenyl-2-propen-1-amine **4k** in the presence of EDCI/ HOBt/Et₃N at 110 °C gave *trans*-fused hexahydrobenzo[*f*]isoindoles **5k**-*trans* as the major product in 55% yield. Highly electron-deficient NO₂ group on phenyl group may decrease the [4 + 2] cycloaddition reaction rate. Thus, *E*-substituted aryl group reacted selectively as a styrene component.

Next the reactions of other electron-deficient alkenes **6** with carboxyl group and 3,3-diaryl-2-propen-1-amines **4** were carried out in order to examine the generality of the reaction (Table 4-4). Reaction of β -substituted (CO₂Me, CF₃, bisCF₃) α , β -unsaturated carboxylic acids **6a-c** and 3,3-diaryl-2-propen-1-amines (Ar=Ph, C₆H₄-4-Cl) **4a**,**d** with EDCI/HOBt/Et₃N at room temperature gave the corresponding amides **7**. On the other hand, the reaction of **6a**,**c** and **4a**, **6a-c** and **4d** on heating (at 60-160 °C) gave *trans*-fused tricyclic compounds **8** as the major products.

Transformation of the amides 7 to the tricyclic compounds 8 was also examined. Thermal reaction of 7a,c,d,f with and without acid or base gave 8 in 21-88% yields (Table 4-5).

In order to understand the reaction mechanism of the cycloadditions and find the factors to control *cis-* and *trans-*fused stereochemistry, a theoretical study was carried out by DFT calculation.⁷ Some theoretical studies on *cis* and *trans-*fused stereoselectivity of IMDA reactions have been reported.^{8c-f} The selectivity varies depending on the steric and electronic effects of linkers and substituents.

EtO ₂ 0	c v v	0₂Et		3 X ¹ 4	Et	aN	O H 9 8 7	O H CO ₂ Et
HO ₂ 0		+		2	H(El	ÖBt DCI/──I	$\begin{array}{c} N \\ 2 \\ 2 \\ \end{array} \xrightarrow{9a} \\ 6 \\ 6 \\ 6 \\ 6 \\ 6 \\ 6 \\ 6 \\ 6 \\ 6 \\ $	
	1	Ph	N H			_→ Ph		Ĥ
				2'` 4h-k	4'			
-			1	2		5-	trans X'	5-cis X1
-	Entry	4	X	X^2	Solvent	Temp.	5-trans (Yield %)	5-cis (Yield %)
	1	4h	2-C1	Н	THF	r.t.	5h- <i>trans</i> (30)	
	2	4h	2-Cl	Н	DMF	r.t.	5h- <i>trans</i> (33)	
	3	4h	2-Cl	Н	Benzene	80 °C	5h- <i>trans</i> (18) ^a	5h- <i>cis</i> (32) ^a
	4	4h	2-Cl	Н	Toluene	110 °C	5h- <i>trans</i> (28) ^a	5h- <i>cis</i> (56) ^a
	5	4i	4-Cl	Н	THF	r.t.	5i- <i>trans</i> (59)	
	6	4i	4-Cl	Н	DMF	r.t.	5i- <i>trans</i> (72)	
	7	4i	4-Cl	Н	DMF	80 °C	5i- <i>trans</i> (58)	
	8	4i	4-Cl	Н	Benzene	80 °C	5 i <i>-trans</i> (48) ^a	5i- <i>cis</i> (27) ^a
	9	4i	4-C1	Н	Toluene	110 °C	5 <i>i</i> - <i>trans</i> (5) ^a	5i- <i>cis</i> (39) ^a
	10	4j	Н	4'-Cl	THF	r.t.	5j-trans (-) ^b	
	11	4j	Н	4'-Cl	DMF	r.t.	5j-trans (86)	
	12	4j	Н	4'-Cl	DMF	80 °C	5 j <i>-trans</i> (71)	
	13	4j	Н	4'-Cl	Benzene	80 °C	5 j <i>-trans</i> (61) ^a	5 <i>j</i> - <i>cis</i> (18) ^a
	14	4j	Н	4'-Cl	Toluene	110 °C	5 <i>j</i> - <i>trans</i> (47) ^a	5j- <i>cis</i> (31) ^a
	15	4k	4-NO ₂	Н	THF	r.t.	5k- <i>trans</i> (32) ^c	
	16	4k	4-NO ₂	Н	DMF	r.t.	5k- <i>trans</i> (34) ^c	
	17	4k	4-NO ₂	Н	Toluene	110 °C	5k- <i>trans</i> (55)	
	18	4k	4-NO ₂	Н	DMF	110 °C	5k- <i>trans</i> (36) ^d	

 Table 4-3. Reactions of 1,1-diethyl 2-hydrogen ethenetricarboxylate 1 and dissymmetrically

 substituted 3,3-diaryl-2-propen-1-amines 4h-k.

^a The yields were calculated by ¹H NMR. ^b A complex mixture containing a small amount of **5***j*-*trans*. ^c A complex mixture containing unidentified compounds was formed along with **5***k*-*trans*. ^d A small amount of possible phenyl epimer of **5***k*-*trans* was also formed but could not be purified.

 E^2 HOBt E F² E EDCI Et₃N C 8a HO₂C Ph Ph N H Ρh š 20 h Ĥ 6 Х 4 7 8 E^1 E^2 Yield (%) 4 Х Solvent Product Entry 6 Temp. Η CO₂Me 4a Η THF **7a**^a ca. 77 1 6a r.t. 2 Toluene 47 6a Η CO₂Me 4a Η 110 °C **8**a THF 3 6b Η CF_3 Η 7b 89 4a r.t. с 4 **6b** Η CF₃ 4a Η Xylene 140 °C Toluene^b с 5 Η CF₃ Η 160 °C **6b** 4a THF 6 CF₃ CF₃ Η 7c^a ca. 39 6c 4a r.t. 7 CF_3 CF_3 Benzene 37 6c 4a Η 80 °C 8c 8 CF_3 CF_3 Η Toluene 48 6c 4a 110 °C 8c 9 CO₂Me Cl THF 6a Η 4d r.t. **7d**^a ca. 86 10 Η CO₂Me Cl THF 60 °C 80 6a 4d 8d 11 Η CO₂Me Cl Benzene 75 6a 4d 80 °C 8d 12 6a Η CO₂Me 4d Cl Toluene 110 °C 69 8d THF 13 6b Η CF₃ 4d Cl r.t. 7e 77 Cl Toluene 110 °C 14 **6b** Η CF₃ 4d 7e 67 15 Η Cl Xylene 27 **6b** CF₃ 4d 140 °C **8e** Toluene^b 16 Η CF₃ Cl 160 °C 38 **6b** 4d **8e** 17 THF 7f^a 6c CF₃ CF₃ 4d Cl r.t. ca. 57 CF₃ Cl Benzene 18 6c CF_3 4d 80 °C **8f** 53 19 CF₃ CF₃ Cl Toluene 45 6c 4d 110 °C 8f

Table 4-4. Reactions of electron-deficient alkenes 6 and 3,3-diaryl-2-propen-1-amines 4.

^a Products 7a,c,d,f are unstable and decompose to give complex mixtures gradually.

^b In a closed vessel. ^c A complex mixture.

$\begin{array}{c} E^{1} \\ O \\ Ph \\ N \\ 7 \\ X \end{array}$											
Entry	7	E^1	E ²	Х	Solvent	Additive	Temp.	Product	Yield (%)		
1	7a	Н	CO ₂ Me	Н	Toluene	none	110 °C	8 a	21		
2	7a	Н	CO ₂ Me	Η	Toluene	HCl (1 equiv) ^a	110 °C	8 a	67		
3	7a	Н	CO ₂ Me	Н	Toluene	Et ₃ N (1 equiv)	110 °C	8 a	83		
4	7b	Н	CF ₃	Н	Toluene	none	110 °C	b			
5	7b	Н	CF ₃	Н	Toluene	HCl (1 equiv) ^a	110 °C	b			
6	7b	Н	CF ₃	Н	Toluene	Et ₃ N (1 equiv)	110 °C	b			
7	7c	CF ₃	CF ₃	Н	Toluene	Et ₃ N (1 equiv)	110 °C	8c	55		
8	7d	Н	CO ₂ Me	Cl	THF	Et ₃ N (1 equiv)	60 °C	8d	88		
9	7e	Н	CF ₃	Cl	Toluene	none	110 °C	b			
10	7e	Н	CF ₃	Cl	Toluene	HCl (1 equiv) ^a	110 °C	b			
11	7e	Н	CF ₃	Cl	Toluene	Et ₃ N (1 equiv)	110 °C	b			
12	7f	CF ₃	CF ₃	Cl	Toluene	Et ₃ N (1 equiv)	110 °C	8f	87		

 Table 4-5. Transformation of amides 7 to hexahydrobenzo[f]isoindoles 8.

^a 1M HCl in ether was added. ^b No reaction.

The *cis*- and *trans*-fused stereoselectivity for reaction of diaryl propenyl amides in the [4 + 2] cycloaddition path has been examined by DFT calculations (Scheme 4-1). For formation of *trans*-fused [4 + 2] cycloadduct **CM1**-*trans*, the asynchronous and concerted path for X = H was obtained. For formation of *cis*-fused [4 + 2] cycloadduct **CM2**-*cis*, stepwise path via zwitter-ionic intermediate **BM2**-*cis* was obtained. The activation energy TSa (+21.47 kcal/mol) leading to **CM1**-*trans* is lower than TSb (+22.51 kcal/mol) and TSc (+24.06 kcal/mol) leading to **CM2**-*cis*. However, the cycloadduct **CM2**-*cis* (+5.96 kcal/mol) is more stable than **CM1**-*trans* (+11.03 kcal/mol). The stability of **CM2**-*cis* may be partially attributed to **8a**,**9a**-*cis* (1,3-diequatorial-like)

conformation of the cyclohexene ring. At higher temperature, the reaction leads to the more stable [4 + 2] cycloadduct CM2-*cis*. The path reacting with *Z*-phenyl group (via AM3 \rightarrow BM3-*cis* \rightarrow CM3-*cis*) was also calculated. However, the path to give CM3-*cis* with 8a,9a-*trans* stereochemistry is unfavorable. The paths reacting with *Z*-phenyl group to give 9a-*trans* stereochemistry could not be obtained because of the steric hindrance. Thus, the reaction paths at *E*-substituted phenyl group of diphenyl propenyl substrates as a diene moiety were preferentially obtained for both *trans* and *cis*-fused products by the DFT calculations. This is in agreement with the experimental results of reactions of dissymmetrically substituted diaryl-2-propen-1-amines 4h-k.

The 1,3-H shift may not be a concerted process, ⁹ and the possible stepwise protonation-deprotonation $(1,3-H^+ \text{ shift})^{10}$ is also considered to play an important role to *cis*and *trans*- fused stereoselectivity in these cases. Selective formation of the stereochemistry at the 4-aryl group may arise from the protonation from less hindered side (*cis* to adjacent H). The result is in agreement with predominant formation of *trans*-benz[*f*]isoindoline in the [4 + 2] cycloaddition reaction of *N*-allyl-*N*-diphenylallyl amide at 180 °C for 6 days reported by Oppolzer et al.^{2c}

The reaction at room temperature proceeds favorably to give CM1-*trans* and the use of polar solvent such as DMF facilitates the stepwise protonation-deprotonation and leads to the *trans*-fused rearomatized product 5M-*trans*.

The reaction of bis(4-chlorophenyl) and bis(4-methylphenyl) substrate models (X = Cl, Me) were also calculated. In the reaction of bis(4-methylphenyl) substrate model (X = Me), the path via intermediate **BM1-***trans* was obtained.

The stepwise 1,3-H⁺ shift of CM2-*cis* (X = H and Me) by catalytic acid in situ leads to rearomatized **5M**-*cis*. The path through TSb with partial ionic character and zwitter-ionic intermediate **BM2**-*cis* (X = Cl) is less stable than those of (X = H and Me) because of destabilization by electron-withdrawing chloro-substituents.

The observed difference on *cis*- and *trans*-fused selectivity by substituents (Table 4-2) may be correlated to the Hammet constants σ .¹¹ σ (*p*-Me; -0.07) gave *cis*-fused product. σ (H; 0, *p*-F; 0.06) gave *cis*- and *trans*-fused products depending on the reaction conditions. Positive values of σ (*p*-Cl; 0.23, m-CF₃; 0.43) only gave *trans*-fused cycloadducts. Larger negative value of σ (*p*-OMe; -0.27) gave a complex mixture, probably because of the formation of the byproducts. The selective formation of *trans*-fused ring of mono-NO₂-substituted **5k**-*trans* (Table 4-3) may be attributed to the destabilization of the cation intermediate **BM2**-*cis* similar to bis-Cl and CF₃-substituted substrates.



Scheme 4-1. [4 + 2] Cycloaddition reaction paths for model compounds of diaryl propenyl amides. ΔE 's (sum of electronic and zero-point energies) by B3LYP/6-311+G(d,p) SCRF = (PCM, solvent = THF) // B3LYP/6-31G* SCRF = (PCM, solvent = THF) relative to AM1 are shown.

Next, the [4 + 2] cycloaddition reaction path of less reactive methyl (2*E*)-4-amino-4-oxo-2-butenoates 7 has been examined (Scheme 4-2). For 7**M**, the models of 7**a**,**d**, the concerted paths lead to both *cis* and *trans* adducts. The activation energies of TSFa and TSFb for 7**M** leading to cycloadducts **CMF**-*trans* and **CMF**-*cis* are substantially higher than those of TSa, TSb, and TSc in Scheme 4-2. The path with lower activation barrier TSFa may give *trans*-[4 + 2] cycloadduct **CMF**-*trans* and the final stable aromatic ring-reproduced product **8M**-*trans* by the stepwise protonation-deprotonation (1,3-H⁺ shift) under the reaction conditions.



Scheme 4-2. [4 + 2] Cycloaddition reaction paths for the models of 7a,d. ΔE 's (sum of electronic and zero-point energies) by B3LYP/6-311+G(d,p) SCRF = (PCM, solvent = THF) // B3LYP/6-31G* SCRF = (PCM, solvent = THF) are shown.

The reaction of less reactive amides 7a,c,d,f with HCl or Et₃N has been shown to give the cyclized products. The 1,3-H⁺ shift under thermal conditions without acid or base for 7a may proceed intermolecularly as well. The [4 + 2] cycloaddition may be reversible and the catalysts accelerate the 1,3-H⁺ shift step.

4-3 Conclusion

In summary, IMDA reactions of various 3,3-diarylpropenylamides of electron-deficient alkenes to give hexahydrobenzo[*f*]isoindoles were investigated. Reaction of 1,1-diethyl 2-hydrogen ethenetricarboxylate with 3,3-diarylpropenylamines or *Z*-cinnamyl amines under the amide formation conditions gave the tricyclic compounds in sequential processes involving IMDA reaction. When the electron-withdrawing substituents (positive values of Hammet constants σ) are present on benzene ring, the [4 + 2] cycloaddition proceeds in *trans*-fused manner. When the substituents such as H and F (σ near 0) are present, the reaction gives *cis* and *trans*-fused products depending on the reaction conditions. When the substituents (slightly negative values of σ) are present, the reaction gives *cis*-fused product. These processes are controlled by the substituents on the benzene ring, reaction temperature, and solvent. Reaction of electron-deficient alkenic carboxylic acids such as fumarate and 3,3-diaryl-2-propen-1-amines under the amide formation conditions at room temperature gave the corresponding amides, and the reaction on heating gave *trans*-fused hexahydrobenzo[*f*]isoindoles. The origin of observed stereoselectivity of the fused rings has been examined by the DFT calculations.

4-4 Experimental Section

General Procedures

¹H Chemical shifts are reported in ppm relative to Me₄Si. ¹³C Chemical shifts are reported in ppm relative to CDCl₃ (77.1 ppm). ¹⁹F Chemical shifts are reported in ppm relative to CFCl₃. ¹³C

multiplicities were determined by DEPT and HSQC. Mass spectra were recorded at an ionizing voltage of 70 eV by EI, FAB or ESI. Mass analyzer type used for EI and FAB is double-focusing and that for ESI is TOF in the HRMS measurements. All reactions were carried out under a nitrogen atmosphere. Column chromatography was performed on silica gel (75-150 μ m).

1,1-Diethyl 2-hydrogen ethenetricarboxylate **1** was prepared according to the literature.¹² (*Z*)-Cinnamyl alcohol¹³ was prepared by hydrogenation of 3-phenyl-2-propyn-1-ol with Lindlar catalyst in methanol.

(*Z*)-Cinnamyl bromide¹⁴ was prepared by reaction of (*Z*)-cinnamyl alcohol with PBr₃ and pyridine in ether and used without further purification. (*Z*)-Cinnamylamines **2a-b** were prepared by reaction of benzylamine or cyclohexylamine (2 equiv) with (*Z*)-cinnamyl bromide in ether according to the literature procedure.¹⁵

(*Z*)-Benzyl cinnamylamine (2a): *Z*:*E* = 5:1 (2.4 mmol scale, 274 mg, 51%); R_f = 0.1 (hexane-ether = 1 : 1); colorless oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) For the major isomer, 1.68 (bs, 1H), 3.55 (dd, *J* = 6.6, 1.9 Hz, 2H), 3.77 (s, 2H), 5.65 (dt, *J* = 11.6, 6.6 Hz, 1H), 6.53 (d, *J* = 11.6 Hz, 1H), 7.19-7.37 (m, 10H). Selected NOEs are between δ 5.65 (=C-H) and δ 6.53 (=C-H).; ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) For the major isomer, 46.99 (CH2), 53.53 (CH2), 126.93 (CH), 127.01 (CH), 128.19 (CH), 128.27 (CH), 128.41 (CH), 128.80 (CH), 130.78 (CH), 130.94 (CH), 137.07 (C), 140.04 (C); IR (neat) 3313, 3024, 2831, 1599, 1494, 1453, 1115, 1028 cm⁻¹; MS (EI) *m/z* 223 (M⁺, 35), 222 (17), 132 (100%); HRMS (EI) *m/z* 223.1372 (calcd for C₁₆H₁₇N 223.1361).

(Z)-Cinnamyl cyclohexylmethylamine (2b): Z:E = 5:1 (20 mmol scale, 2.24 g, 49%); $R_f = 0.4$ (hexane-ether = 1 : 4); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) For the major isomer, 0.834-0.957 (m, 2H), 1.08-1.29 (m, 3H), 1.36-1.48 (m, 1H), 1.62-1.77 (m, 5H), 2.43 (d, J = 6.6 Hz, 2H), 3.50 (dd, J = 6.5, 1.9 Hz, 2H), 5.76 (dt, J = 11.7, 6.5 Hz, 1H), 6.49 (d, J = 11.7 Hz, 1H), 7.16-7.37 (m, 5H). Selected NOEs are between δ 5.76 (=C-*H*) and δ 6.49 (=C-*H*) and between δ 3.50 (*CH*₂) and δ 7.16-7.37 (Ph).; ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) For the major isomer,

26.03 (CH₂), 26.66 (CH₂), 31.45 (CH₂), 38.01 (CH), 47.92 (CH₂), 56.48 (CH₂), 126.76 (CH), 128.08 (CH), 128.74 (CH), 130.14 (CH), 131.56 (CH), 137.18 (C); IR (neat) 3323, 3023, 2923, 1600, 1494, 1447, 1124 cm⁻¹; MS (EI) *m/z* 229 (M⁺, 9.3), 146 (19), 117 (100%); HRMS (EI) *m/z* 229.1836 (calcd for C₁₆H₂₃N 229.1830).

Typical experimental procedure for preparation 3 (Table 4-1, entry2). To a solution of 1,1-diethyl 2-hydrogen ethenetricarboxylate (1) (prepared from 1,1-diethyl 2-*tert*-butyl ethenetricarboxylate (272 mg, 1 mmol) upon treatment with CF₃CO₂H (4 mL))¹² in benzene (0.7 mL) were added *Z*-benzyl cinnamylamine (**2a**) (223 mg, 1 mmol) in benzene (0.7 mL), Et₃N (0.14 mL, 102 mg, 1 mmol), HOBt (1-hydroxybenzotriazole) (270 mg, 2 mmol), and EDCI (1-[3-(dimethylamino)propyl]-3-ethylcarbodiimide hydrochloride) (199 mg, 1.04 mmol) at 0 °C. The reaction mixture was stirred for 1 h at 0 °C, and then heated at 80 °C and stirred for 20 h. The reaction mixture was diluted with CH₂Cl₂. The organic phase was washed with saturated aqueous NaHCO₃ solution, 2M aqueous citric acid, saturated aqueous NaHCO₃ and water, dried (Na₂SO₄), and evaporated *in vacuo*. The residue was purified by column chromatography over silica gel eluting with hexane-Et₂O to give **3a** (261 mg, 62%).

3a: $R_f = 0.5$ (hexane-ether = 1 : 4); pale yellow oil; ¹H NMR (400 MHz, CDCl3) δ (ppm) 1.17 (t, J = 7.1 Hz, 3H), 1.33 (t, J = 7.1 Hz, 3H), 2.34 (dd, J = 17.4, 11.1 Hz, 1H), 2.86-2.98 (m, 3H), 3.49 (dd, J = 8.5, 8.5 Hz, 1H), 3.59 (d, J = 11.1 Hz, 1H), 4.07-4.26 (m, 2H), 4.27 (d, J = 14.7 Hz, 1H), 4.35-4.40 (m, 2H), 4.67 (d, J = 14.7 Hz, 1H), 7.08 (d, J = 7.2 Hz, 1H), 7.15-7.18 (m, 2H), 7.20-7.30 (m, 5H), 7.53 (dd, J = 7.7, 1.3 Hz, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 13.80 (CH₃), 13.99 (CH₃), 29.05 (CH), 33.21 (CH₂), 46.85 (CH₂), 48.41 (CH), 52.48 (CH₂), 60.72 (C), 61.72 (CH₂), 62.29 (CH₂), 126.96 (CH), 127.34 (CH), 127.77 (CH), 127.82 (CH), 127.97 (CH), 128.22 (CH), 128.49 (CH), 135.36 (C), 136.14 (C), 136.35 (C), 168.81 (C), 169.88 (C), 172.77 (C); ¹H NMR (400 MHz, C₆D₆) δ (ppm) 0.843 (t, J = 7.1 Hz, 3H), 0.949 (t, J = 7.1 Hz, 3H), 2.09

(dd, J = 14.8, 8.6 Hz, 1H), 2.35 (ddddd, J = 11.5, 9.0, 8.6, 7.2, 6.4 Hz, 1H), 2.50 (dd, J = 14.8, 7.2 Hz, 1H), 2.58 (dd, J = 9.0 Hz, 6.4 Hz, 1H), 2.88 (dd, J = 9.0, 9.0 Hz, 1H), 3.62 (d, J = 11.5 Hz, 1H), 3.81-3.89 (m, 1H), 4.01-4.17 (m, 3H), 4.17 (d, J = 14.8 Hz, 1H), 4.35 (d, J = 14.8 Hz, 1H), 6.78 (d, J = 7.0 Hz, 1H), 6.93-7.08 (m, 7H), 7.83 (dd, J = 7.8, 1.0 Hz, 1H). Selected NOEs are between δ 2.35 (C3a-*H*) and δ 3.62 (C9a-*H*), 2.88 (C3-H*H*) and 2.50 (C4-H*H*). Atom numbering is shown in Table 4-1.; ¹³C NMR (100.6 MHz, C₆D₆) δ (ppm) 13.85 (CH₃), 13.95 (CH₃), 29.30 (CH), 33.24 (CH₂), 46.94 (CH₂), 48.68 (CH), 52.21 (CH₂), 61.08 (C), 61.61 (CH₂), 62.17 (CH₂), 127.13 (CH), 127.38 (CH), 127.89 (CH), 128.24 (CH), 128.49 (CH), 128.66 (CH), 128.73 (CH), 136.15 (C), 136.78 (C), 137.29 (C), 168.87 (C), 169.95 (C), 172.26 (C). Selected HMBC correlations are between δ 2.35 (C3a-*H*), 2.58 (C3-*H*H), 2.50 (C4-H*H*), 3.62 (C9a-*H*), 6.78 (C5-*H*) and δ 33.24 (C4), and between δ 3.62 (C9a-*H*) and δ 61.08 (C9).; IR (neat) 2980, 1734, 1689, 1485, 1444, 1247, 1096, 1031 cm⁻¹; MS (EI) *m/z* 421 (M⁺, 45), 347 (8.4), 303 (14), 301 (12), 91 (100%); HRMS (EI) *m/z* 421.1906 (calcd for C₂₅H₂₇NO₅ 421.1889).

3b: (1 mmol scale, toluene, 110 °C, 262 mg, 61%); $R_f = 0.3$ (hexane-ether = 1 : 4); pale yellow oil; ¹H NMR (400 MHz, CDCl3) δ (ppm) 0.803-0.925 (m, 2H), 1.09-1.20 (m, 3H), 1.17 (t, J = 7.0 Hz, 3H), 1.30 (t, J = 7.1 Hz, 3H), 1.43-1.70 (m, 6H), 2.40 (dd, J = 17.5, 11.0 Hz, 1H), 2.87 (dd, J = 13.5, 6.7 Hz, 1H), 2.96-3.01 (m, 2H), 3.05 (dd, J = 9.0, 5.9 Hz, 1H), 3.30 (dd, J = 13.5, 7.8 Hz, 1H), 3.55 (d, J = 10.7 Hz, 1H), 3.66 (dd, J = 9.0, 8.6 Hz, 1H), 4.08-4.26 (m, 2H), 4.34 (q, J = 7.1 Hz, 2H), 7.11-7.13 (m, 1H), 7.20-7.28 (m, 2H), 7.54-7.57 (m, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 13.83 (CH₃), 13.98 (CH₃), 25.77 (CH₂), 25.83 (CH₂), 26.38 (CH₂), 29.17 (CH), 30.74 (CH₂), 30.78 (CH₂), 33.34 (CH₂), 35.85 (CH), 48.52 (CH), 49.40 (CH₂), 54.10 (CH₂), 60.54 (C), 61.65 (CH₂), 62.24 (CH₂), 126.94 (CH), 127.71 (CH), 127.95 (CH), 127.97 (CH), 135.36 (C), 136.14 (C), 168.75 (C), 169.90 (C), 172.96 (C); ¹H NMR (400 MHz, C₆D₆) δ (ppm) 0.689-0.850 (m, 2H), 0.965-1.12 (m, 3H), 0.938 (t, J = 7.1 Hz, 3H), 0.993 (t, J = 7.0 Hz, 3H),

1.36-1.45 (m, 2H), 1.49-1.63 (m, 4H), 2.26 (dd, J = 14.8, 8.2 Hz, 1H), 2.53 (ddddd, J = 11.3, 8.8, 8.2, 7.0, 6.4 Hz, 1H), 2.69 (dd, J = 14.8, 7.0 Hz, 1H), 2.73 (dd, J = 9.0, 6.4 Hz, 1H), 2.90 (dd, J = 13.5, 6.8 Hz, 1H), 3.04 (dd, J = 9.0, 8.8 Hz, 1H), 3.12 (dd, J = 13.5, 7.2 Hz, 1H), 3.71 (d, J = 11.3 Hz, 1H), 3.91-3.99 (m, 1H), 4.07-4.24 (m, 3H), 6.92 (d, J = 7.4, 0.6 Hz, 1H), 7.05 (ddd, J = 7.6, 7.4, 1.2 Hz, 1H), 7.14 (ddd, J = 7.8, 7.6, 1.4 Hz, 1H), 8.01 (d, J = 7.8 Hz, 1H). Selected NOEs are between δ 2.53 (C3a-*H*) and δ 3.71 (C9a-*H*), 3.04 (C3-H*H*) and 2.69 (C4-H*H*).; ¹³C NMR (100.6 MHz, C₆D₆) δ (ppm) 13.85 (CH₃), 13.90 (CH₃), 26.15 (CH₂), 26.70 (CH₂), 29.43 (CH), 30.88 (CH₂), 31.06 (CH₂), 33.44 (CH₂), 36.16 (CH), 48.76 (CH), 49.47 (CH₂), 53.77 (CH₂), 60.88 (C), 61.53 (CH₂), 62.10 (CH₂), 127.14 (CH), 127.76 (CH), 128.24 (CH), 128.98 (CH), 136.10 (C), 136.71 (C), 168.76 (C), 169.93 (C), 172.34 (C). Selected HMBC correlations are between δ 2.26 (C4-*H*H), 2.69 (C4-HH), 2.73 (C3-*H*H), 3.04 (C3-HH), 3.71 (C9a-*H*) and δ 29.30 (*C*3a), δ 2.58 (C3-*H*H), 2.88 (C3-HH), 3.62 (C9a-*H*), 6.92 (C5-*H*) and δ 29.43 (C4), and between δ 3.62 (C9a-*H*) and δ 60.88 (C9).; IR (neat) 2925, 2852, 1733, 1690, 1485, 1448, 1366, 1236, 1118, 1095, 1035 cm⁻¹; MS (EI) *m*/*z* 427 (M⁺, 9.9), 345 (11), 117 (10), 84 (100%); HRMS (EI) *m*/*z* 427.2368 (calcd for C₂₅H₃₃NO₅ 427.2359).

Arylpropenyl esters, ethyl 3,3-diphenylacrylate **Xa** (for **4a-b**), ethyl 3,3-bis-(4-fluorophenyl)acrylate **Xc** (for **4c**), ethyl 3,3-bis[3-(trifluoromethyl)phenyl]acrylate **Xe** (for **4e**) and ethyl (2*Z*)-3-(2-chlorophenyl)-3-phenylprop-2-enoate **Xh** (for **4h**) and the corresponding alcohols, 3,3-diphenylprop-2-en-1-ol **Ya**, 3,3-bis(4-fluorophenyl)prop-2-en-1-ol **Yc**, 3,3-bis(4-chlorophenyl)prop-2-en-1-ol **Yd**, 3,3-bis[3-(trifluoromethyl)phenyl]prop-2-en-1-ol **Ye**, and 3-(2-chlorophenyl)-3-phenylprop-2-en-1-ol **Yh** were prepared according to the literature. ⁶ Arylpropenyl esters **Xf** (for **4f**), **Xg** (for **4g**), **Xi** (for **4i**), **Xj** (for **4j**), **Xk** (for **4k**) and the corresponding alcohols **Yf** (for **4f**), **Yg** (for **4g**), **Yi** (for **4i**), **Yk** (for **4k**) were prepared by the literature methods.⁶ The stereochemistry of ethyl 3-(2-chlorophenyl)-3-phenylprop-2-enoate **Xh** obtained as the major product by the literature method was reported as E^6 but it was found to be Z by the observed NOE's in C₆D₆. The ¹H NMR spectra of **Xh** in CDCl₃ were in accord with the reported data.

(2*E*)-3-(4-chlorophenyl)-3-phenylprop-2-en-1-ol (**Yj**) (for **4j**) was prepared by the Suzuki cross-coupling reaction of (*E*)-3-bromo-3-phenylprop-2-en-1-ol and 4-chlorophenylboronic acid by the literature method. ¹⁶

Ethyl 3,3-bis(4-methylphenyl)acrylate (Xf): (2.7 mmol scale, 721 mg, 95%); $R_f = 0.4$ (hexane-ether = 4 : 1); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.14 (t, J = 7.1 Hz, 3H), 2.34 (s, 3H), 2.39 (s, 3H), 4.06 (q, J = 7.1 Hz, 2H), 6.30 (s, 1H), 7.08-7.12 (m, 4H), 7.17-7.20 (m, 4H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 14.13 (CH₃), 21.29 (CH₃), 21.44 (CH₃), 59.96 (CH₂), 116.22 (CH), 128.38 (CH), 128.59 (CH), 129.09 (CH), 129.21 (CH), 136.16 (C), 137.94 (C), 138.34 (C), 139.61 (C), 156.94 (C), 166.31 (C); IR (neat) 2980, 1723, 1604, 1508, 1368, 1264, 1160, 1038 cm⁻¹; MS (EI) *m/z* 280 (M⁺, 98), 235 (100%); HRMS (EI) *m/z* 280.1459 (calcd for C₁₉H₂₀O₂ 280.1463).

Ethyl 3,3-bis(4-methoxyphenyl)acrylate (Xg): (2.7 mmol scale, 537 mg, 64%); $R_f = 0.4$ (hexane-ether = 1 : 1); colorless oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.16 (t, J = 7.1 Hz, 3H), 3.81 (s, 3H), 3.84 (s, 3H), 4.07 (q, J = 7.1 Hz, 2H), 6.23 (s, 1H), 6.84 (d-like, J = 9.0 Hz, 2H), 6.91 (d-like, J = 8.7 Hz, 2H), 7.15 (d-like, J = 8.7 Hz, 2H), 7.24 (d-like, J = 9.0 Hz, 2H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 14.21 (CH₃), 55.27 (CH₃), 55.39 (CH₃), 59.90 (CH₂), 113.26 (CH), 113.74 (CH), 114.97 (CH), 130.04 (CH), 130.91 (CH), 131.33 (C), 133.89 (C), 156.43 (C), 159.71 (C), 160.79 (C), 166.50 (C); IR (neat) 2979, 2837, 1717, 1600, 1513, 1250, 1174, 1149, 1034 cm⁻¹; MS (EI) *m/z* 312 (M⁺, 100), 267 (33), 240 (49%); HRMS (EI) *m/z* 312.1357 (calcd for C₁₉H₂₀O₄ 312.1362).

Ethyl (2*Z*)-3-(2-chlorophenyl)-3-phenylprop-2-enoate (Xh): ¹H NMR (400 MHz, C₆D₆) δ (ppm) 0.824 (t, *J* = 7.1 Hz, 3H), 3.88 (q, *J* = 7.1 Hz, 2H), 6.61 (s, 1H), 6.86 (ddd, *J* = 7.6, 7.5, 2.0

Hz, 1H), 6.91 (ddd, J = 7.5, 7.4, 1.4 Hz, 1H), 6.95-7.03 (m, 5H), 7.18-7.20 (m, 2H), 7.24 (dd, J = 7.6, 1.4 Hz, 1H). Selected NOEs are between δ 6.61 (C2-*H*) and δ 7.18-7.20 (2-*H* of Ph).; ¹³C NMR (100.6 MHz, C₆D₆) δ (ppm) 13.99 (CH₃), 59.91 (CH₂), 119.56 (CH), 126.54 (CH), 127.67 (CH), 128.79 (CH), 129.05 (CH), 129.57 (CH), 130.62 (CH), 133.14 (C), 138.74 (C), 139.06 (C), 153.03 (C), 164.89 (C).

Xi and Xj were obtained as a ca. 1:1 mixture (20 mmol scale, 5.46 g, 95%). Xi (2.15 g, 38%) was partially isolated by fractional crystallization of the mixture (from hexane). Xj (3:1 (Xj:Xi) mixture) was obtained from the filtrates. The stereochemistries of Xi and Xj were assigned by the NOE's of the corresponding alcohols Yi and Yj obtained by DIBAL-H reduction. The spectral data for Yj obtained from Xj were in accord with those by the Suzuki cross-coupling reaction.

Ethyl (2*Z*)-3-(4-chlorophenyl)-3-phenylprop-2-enoate (Xi): $R_f = 0.5$ (hexane-ether = 4 : 1); colorless crystals; mp 67-68 °C (hexane); ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.16 (t, δ = 7.1 Hz, 3H), 4.07 (q, *J* = 7.1 Hz, 2H), 6.37 (s, 1H), 7.15 (d-like, *J* = 8.4 Hz, 2H), 7.26-7.39 (m, 7H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 14.12 (CH₃), 60.27 (CH₂), 117.92 (CH), 128.24 (CH), 128.33 (CH), 128.56 (CH), 129.69 (CH), 130.69 (CH), 134.26 (C), 137.43 (C), 140.49 (C), 155.46 (C), 165.96 (C); IR (KBr) 2981, 1718, 1488, 1368, 1271, 1157, 1089 cm⁻¹; MS (EI) *m/z* 288 (M⁺, 27), 286 (M⁺, 81), 241 (100%); HRMS (EI) *m/z* 286.0759, 288.0732 (calcd for C₁₇H₁₅ClO₂ 286.0761, 288.0731); Anal. Calcd for C₁₇H₁₅ClO₂: C, 71.20; H, 5.27. Found: C, 71.27; H, 5.29.

Ethyl (2*E*)-3-(4-chlorophenyl)-3-phenylprop-2-enoate (Xj): $R_f = 0.4$ (hexane-ether = 4 : 1); pale yellow oil; (Xj:Xi = 3:1) ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.10 (t, *J* = 7.1 Hz, 3H), 4.04 (q, *J* = 7.1 Hz, 2H), 6.33 (s, 1H), 7.14-7.23 (m, 3H), 7.25-7.39 (m, 6H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 13.98 (CH₃), 60.14 (CH2), 117.81 (CH), 128.00 (CH), 128.61 (CH), 129.08 (CH), 129.55 (CH), 130.62 (CH), 135.50 (C), 138.51 (C), 139.26 (C), 155.10 (C), 165.86 (C); IR (neat) 2981, 1724, 1618, 1489, 1368, 1263, 1165, 1092 cm⁻¹; MS (EI) *m/z* 288 (M⁺, 27), 286 (M⁺, 79), 241 (89), 214 (51), 178 (100%); HRMS (EI) *m/z* 286.0767, 288.0743 (calcd for C₁₇H₁₅ClO₂ 286.0761, 288.0731).

Ethyl (2*Z***)-3-(4-nitrophenyl)-3-phenylprop-2-enoate (Xk)**: (2.7 mmol scale, 424 mg, purified by recrystalization, 52%); $R_f = 0.4$ (hexane- AcOEt = 4 : 1); yellow crystals; mp 88.5-90.0 °C (hexane-MeOH); ¹H NMR (400 MHz, CDCl3) *δ* (ppm) 1.17 (t, *J* = 7.1 Hz, 3H), 4.07 (d, *J* = 7.1 Hz, 2H), 6.48 (s, 1H), 7.24-7.27 (m, 2H), 7.33-7.43 (m, 5H), 8.27 (d-like, *J* = 8.8 Hz, 2H). Selected NOEs are between *δ* 6.48 (C2-*H*) and *δ* 7.24-7.27 (2-*H* of Ph).; ¹³C NMR (100.6 MHz, CDCl₃) *δ* (ppm) 14.12 (CH₃), 60.54 (CH₂), 118.62 (CH), 123.34 (CH), 128.08 (CH), 128.81 (CH), 130.08 (CH), 130.14 (CH), 139.24 (C), 146.13 (C), 147.57 (C), 154.50 (C), 165.49 (C). Selected HMBC correlations are between *δ* 6.48 (C2-*H*) and *δ* 139.24 (1-*C* of Ph), and *δ* 7.24-7.27 (2-*H* of Ph) and *δ* 154.50 (C3).; IR (KBr) 2985, 1719, 1619, 1594, 1515, 1350, 1267, 1172, 1034 cm⁻¹; MS (EI) *m*/*z* 297 (M⁺, 80), 252 (100%); HRMS (EI) *m*/*z* 297.1010 (calcd for C₁₇H₁₅NO4 297.1001).

3,3-Bis(4-methylphenyl)prop-2-en-1-ol (Yf): (26 mmol scale, 5.915 g, 95%); colorless crystals: mp 68.5-69.0 °C; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.57 (bs, 1H), 2.33 (s, 3H), 2.37 (s, 3H), 4.19 (d, *J* = 6.8 Hz, 2H), 6.17 (t, *J* = 6.8 Hz, 1H), 7.04 (d-like, *J* = 8.0 Hz, 2H), 7.08 (d-like, *J* = 8.0 Hz, 2H), 7.13-7.17 (m, 4H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 21.15 (CH₃), 21.29 (CH₃), 60.81 (CH2), 126.46 (CH), 127.62 (CH), 128.91 (CH), 129.72 (CH), 136.28 (C), 137.26 (C), 137.42 (C), 139.27 (C), 144.13 (C); IR (KBr) 3267, 2916, 1510, 1012 cm⁻¹; MS (EI) *m/z* 238 (M⁺, 60), 223 (68), 195 (100%); HRMS (EI) *m/z* 238.1361 (calcd for C₁₇H₁₈O 238.1358).

3,3-Bis(4-methoxyphenyl)prop-2-en-1-ol (Yg): (19.4 mmol scale, 5.26 g, 100%); $R_f = 0.1$ (hexane-ether = 1 : 1); colorless oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.75 (bs, 1H), 3.79 (s, 3H), 3.82 (s, 3H), 4.20 (d. J = 7.0 Hz, 2H), 6.09 (t, J = 7.0 Hz, 1H), 6.81 (d-like, J = 8.9 Hz, 2H), 6.88 (d-like, J = 8.8 Hz, 2H), 7.07 (d-like, J = 8.8 Hz, 2H), 7.18 (d-like, J = 8.9 Hz, 2H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 55.29 (CH₃), 55.31 (CH₃), 60.81 (CH₂), 113.54 (CH), 113.56

(CH), 125.46 (CH), 128.92 (CH), 131.01 (CH), 131.60 (C), 134.86 (C), 143.48 (C), 159.01 (C), 159.24 (C); IR (neat) 3344, 2933, 2835, 1607, 1511, 1245, 1174, 1034 cm⁻¹; MS (EI) *m/z* 270 (M⁺, 52), 242 (29), 227 (100), 135 (65%); HRMS (EI) *m/z* 270.1253 (calcd for C₁₇H₁₈O₃ 270.1256).

(2*Z*)-3-(2-Chlorophenyl)-3-phenylprop-2-en-1-ol (Yh): (11.4 mmol scale, 2.67 g, 96%); $R_f = 0.1$ (hexane-ether = 1 : 1); colorless oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.55 (bs, 1H), 4.03-4.08 (m, 2H), 6.44 (t, J = 6.8 Hz, 1H), 7.18-7.20 (m, 1H), 7.22-7.33 (m, 7H), 7.43-7.48 (m, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 60.99 (CH₂), 126.58 (CH), 126.96 (CH), 127.83 (CH), 128.46 (CH), 128.60 (CH), 129.15 (CH), 129.82 (CH), 131.50 (CH), 133.61 (C), 137.75 (C), 139.66 (C), 140.79 (C); IR (neat) 3327, 3056, 2867, 1597, 1494, 1472, 1446, 1052, 1035 cm⁻¹; MS (EI) *m*/*z* 246 (M⁺, 11), 244 (M⁺, 32), 209 (100%); HRMS (EI) *m*/*z* 244.0651, 246.0625 (calcd for C₁₅H₁₃ClO 244.0655, 246.0625); ¹H NMR (400 MHz, C₆D₆) δ (ppm) 0.43 (bs, 1H), 3.93 (bs, 2H), 6.33-6.39 (m, 1H), 6.78-6.88 (m, 2H), 6.92-6.97 (m, 1H), 7.00-7.09 (m, 3H), 7.19 (dd, J = 7.7, 1.5 Hz, 1H), 7.22-7.25 (m, 2H). Selected NOEs are between δ 6.33-6.39 (C*H*=CPh) and δ 7.22-7.25 (2'-*H* of CH=*CPh*) and between δ 3.93 (CH₂) and δ 6.92-6.97 (6-*H* of 2-chlorophenyl) in C₆D₆; ¹³C NMR (100.6 MHz, C₆D₆) δ (ppm) 60.75 (CH₂), 126.82 (CH), 126.90 (CH), 127.76 (CH), 128.65 (CH), 129.05 (CH), 129.89 (CH), 129.91 (CH), 131.77 (CH), 134.08 (C), 138.37 (C), 140.17 (C), 140.25 (C).

(2Z)-3-(4-Chlorophenyl)-3-phenylprop-2-en-1-ol (Yi): (8.75 mmol scale, 2.37 g, 94%); $R_f = 0.4$ (hexane-ether = 1 : 1); colorless crystals; mp 88.5-89.0 °C; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.48 (bs, 1H), 4.20 (d, J = 6.8 Hz, 2H), 6.24 (t, J = 6.8 Hz, 1H), 7.11 (d-like, J = 8.6 Hz, 2H), 7.22-7.32 (m, 5H), 7.35 (d-like, J = 8.6 Hz, 2H). Selected NOEs are between δ 4.20 (CH₂) and δ 7.11 (2-*H* of 4-chlorophenyl).; ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 60.66 (CH₂), 127.88 (CH), 127.91 (CH), 128.02 (CH), 128.37 (CH), 128.57 (CH), 131.21 (CH), 133.67 (C), 137.55 (C), 141.47 (C), 143.26 (C); IR (KBr) 3270, 2860, 1594, 1491, 1091, 1015 cm⁻¹; MS (EI) *m/z* 246 (M⁺, 23), 244 (M⁺, 70), 201 (100%); HRMS (EI) *m/z* 244.0647, 246.0615 (calcd for C₁₅H₁₃ClO 244.0655, 246.0625).

(2*Z*)-3-(4-Nitrophenyl)-3-phenylprop-2-en-1-ol (Yk): (5 mmol scale, 854 mg, 88%); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 2.12 (bs, 1H), 4.19 (d, *J* = 6.9 Hz, 2H), 6.34 (t, *J* = 6.9 Hz, 1H), 7.18-7.21 (m, 2H), 7.28-7.32 (m, 3H), 7.35 (d-like, *J* = 8.8 Hz, 2H), 8.21 (d-like, *J* = 8.8 Hz, 2H). Selected NOEs are between δ 6.34 (*CH*=CPh) and δ 7.18-7.21 (2'-*H* of CH=C*Ph*) between δ 4.19 (*CH*₂) and δ 7.35 (2-*H* of 4-nitrophenyl).; ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 60.27 (CH₂), 123.53 (CH), 127.56 (CH), 128.21 (CH), 128.51 (CH), 129.34 (CH), 130.75 (CH), 140.57 (C), 142.29 (C), 146.10 (C), 147.21 (C); IR (neat) 3359, 3078, 1520, 1347, 1107, 1014 cm⁻¹; MS (EI) *m*/*z* 255 (M⁺, 100), 237 (79), 212 (61), 165 (78%); HRMS (EI) *m*/*z* 255.0892 (calcd for C₁₅H₁₃NO₃ 255.0895).

Preparation of Yj. A mixture of $Pd(OAc)_2$ (22 mg, 0.1 mmol), PPh₃ (52 mg, 0.2 mmol), 4-chlorophenylboronic acid (2.4 mmol), KOH (224 mg, 4 mmol), MeOH (8 mL), and THF (8 mL) was heated at 60 °C overnight. After cooling to room temperature, the solution was taken up in Et₂O (30 mL) and the Et₂O layer was washed with aq 1.0 M NaOH (10 mL) and brine (2 × 5 mL). The organic layer was dried (MgSO₄), filtered, and concentrated under reduced pressure. The crude material was purified by column chromatography on silica gel using hexane/ethyl acetate as eluent. The *E*-configuration of the products was assigned by 2D-NOESY.

(2*E*)-3-(4-Chlorophenyl)-3-phenylprop-2-en-1-ol (Yj): (6.8 mmol scale, 1.49 g, 90%); $R_f = 0.3$ (hexane-AcOEt = 1 : 1); colorless crystals; mp 72.0-73.0 °C; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.53 (bs, 1H), 4.21 (d, J = 6.8 Hz, 2H), 6.22 (t, J = 6.8 Hz, 1H), 7.12-7.15 (m, 2H), 7.18 (d-like, J = 8.6 Hz, 2H), 7.25 (d-like, J = 8.6 Hz, 2H), 7.32-7.40 (m, 3H). Selected NOEs are between δ 4.21 (CH₂) and δ 7.12-7.15 (2'-*H* of CH=CP*h*) and between δ 6.22 (C*H*=CAr) and δ 7.18 (2-*H* of 4-chlorophenyl).; ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 60.71 (CH₂), 127.88 (CH), 127.97 (CH), 128.42 (CH), 128.95 (CH), 129.73 (CH), 133.57 (C), 138.64 (C), 140.33 (C), 143.15 (C); IR (neat) 3261, 2848, 1488, 1092, 1010 cm⁻¹; MS (EI) *m/z* 246 (M⁺, 22), 244 (M⁺, 67), 226 (37), 209 (61), 201 (100%); HRMS (EI) *m/z* 244.0651, 246.0625 (calcd for C₁₅H₁₃ClO 244.0655, 246.0625).

3,3-Diaryl-2-propen-1-amines **4a-k** were prepared from the corresponding alcohols **Ya**, **Yc-k**. The corresponding bromides were prepared by reaction of the alcohols with PBr₃ in ether and used without further purification. 3,3-Diaryl-2-propen-1-amines **4a-k** were prepared by reaction of benzylamine (2 equiv) with the corresponding bromide in ether according to the literature procedure.¹⁵

4a: (1.5 mmol scale, 216 mg, 48%); $R_f = 0.1$ (hexane-ether = 1 : 1); colorless oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 2.00 (bs, 1H), 3.35 (d, J = 7.0 Hz, 2H), 3.75 (s, 2H), 6.22 (t, J = 7.0 Hz, 1H), 7.14-7.16 (m, 2H), 7.19-7.36 (m, 13H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 48.02 (CH₂), 53.43 (CH₂), 127.05 (CH), 127.30 (CH), 127.34 (CH), 127.48 (CH), 128.19 (CH), 128.24 (CH), 128.26 (CH), 128.44 (CH), 129.81 (CH), 139.64 (C), 139.99 (C), 142.21 (C), 143.83 (C); IR (neat) 3330, 3026, 2836, 1598, 1495, 1443, 1114, 1073, 1028 cm⁻¹; MS (EI) *m/z* 299 (M⁺, 70), 298 (21), 222 (30), 208 (66), 132 (49), 91 (100%); HRMS (EI) *m/z* 299.1678 (calcd for C₂₂H₂₁N 299.1674).

4b: (10 mmol scale, 1.31 g, 43%); $R_f = 0.1$ (hexane-ether = 1 : 1); colorless crystals; mp 67.5-68.5 °C; 1H NMR (400 MHz, CDCl₃) δ (ppm) 0.803-0.901 (m, 2H), 1.06-1.24 (m, 4H), 1.31-1.42 (m, 1H), 1.62-1.70 (m, 5H), 2.38 (d, J = 6.6 Hz, 2H), 3.27 (d, J = 6.8 Hz, 2H), 6.18 (t, J = 6.8 Hz, 1H), 7.14-7.34 (m, 5H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 25.93 (CH₂), 26.57 (CH₂), 31.29 (CH₂), 37.80 (CH), 48.60 (CH₂), 56.19 (CH₂), 126.99 (CH), 127.20 (CH), 127.94 (CH), 127.98 (CH), 128.04 (CH), 129.60 (CH), 139.61 (C), 142.06 (C), 142.96 (C); IR (neat) 3317, 3076, 2925, 2846, 1596, 1493, 1444, 1363, 1118 cm⁻¹; MS (EI) *m/z* 305 (M⁺, 46), 302 (11),

222 (11), 193 (100%); HRMS (EI) *m/z* 305.2145 (calcd for C₂₂H₂₇N 305.2143); Anal. Calcd for C₂₂H₂₇N: C, 86.51; H, 8.91; N, 4.59. Found: C, 86.35; H, 8.66; N, 4.49.

4c: (2 mmol scale, 501 mg, 74%); $R_f = 0.4$ (hexane-ether = 4 : 1); colorless oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.60 (bs, 1H), 3.32 (d, J = 6.8 Hz, 2H), 3.75 (s, 2H), 6.13 (t, J = 6.8 Hz, 1H), 6.92-6.98 (m, 2H), 7.01-7.06 (m, 2H), 7.07-7.12 (m, 2H), 7.14-7.19 (m, 2H), 7.21-7.32 (m, 5H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 48.00 (CH₂), 53.59 (CH₂), 115.10 (CH, d, $J_{FC} = 21$ Hz), 115.27 (CH, d, $J_{FC} = 21$ Hz), 127.09 (CH), 127.90 (CH), 128.19 (CH), 128.47 (CH), 129.03 (CH, d, $J_{FC} = 8.4$ Hz), 131.39 (CH, d, $J_{FC} = 8.4$ Hz), 135.28 (C, d, $J_{FC} = 3.1$ Hz), 138.22 (C, d, $J_{FC} = 3.1$ Hz), 140.06 (C), 141.63 (C), 162.13 (C, d, $J_{FC} = 247$ Hz), 162.32 (C, d, $J_{FC} = 247$ Hz); ¹⁹F NMR (376 MHz, CDCl₃) δ (ppm) -115.14 (m), -114.68 (m); IR (neat) 3330, 3028, 2835, 1601, 1508, 1453, 1224, 1159, 1095 cm⁻¹; MS (EI) *m/z* 335 (M⁺, 94), 244 (70), 132 (67), 91 (100%); HRMS (EI) *m/z* 335.1482 (calcd for C₂₂H₁₉F₂N 335.1486).

4d: (2.5 mmol scale, 446 mg, 48%); colorless oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.61 (bs, 1H), 3.31 (d, J = 6.8 Hz, 2H), 3.73 (s, 2H), 6.17 (t, J = 6.8 Hz, 1H), 7.05 (d-like, J = 8.7 Hz, 2H), 7.11 (d-like, J = 8.7 Hz, 2H), 7.20-7.32 (m, 9H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 47.98 (CH₂), 53.62 (CH₂), 127.08 (CH), 128.15 (CH), 128.43 (CH), 128.47 (CH), 128.60 (CH), 128.69 (CH), 128.89 (CH), 131.10 (CH), 133.41 (C), 133.49 (C), 137.57 (C), 140.09 (C), 140.29 (C), 141.37 (C); IR (neat) 3028, 2837, 1591, 1492, 1091, 1014 cm⁻¹; MS (EI) *m/z* 369 (M⁺, 6.5), 367 (M⁺, 10), 278 (7.3), 276 (10), 165 (60), 164 (58), 106 (100%); HRMS (EI) *m/z* 367.0887, 369.0887 (calcd for C₂₂H₁₉Cl₂N 367.0895, 369.0865).

4e: (1.9 mmol scale, 592 mg, 72%); $R_f = 0.3$ (hexane-ether = 1 : 4); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.52 (bs, 1H), 3.34 (d, J = 6.8 Hz, 2H), 3.76 (s, 2H), 6.30 (t, J = 6.8 Hz, 1H), 7.22-7.34 (m, 7H), 7.39 (dd, J = 7.7, 7.7 Hz, 1H), 7.41 (s, 1H), 7.47-7.53 (m, 3H), 7.61 (d, J = 7.8 Hz, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 47.98 (CH₂), 53.75 (CH₂), 123.96 (CH, q, $J_{FC} = 3.8$ Hz), 124.08 (C, q, $J_{FC} = 272$ Hz), 124.12 (C, q, $J_{FC} = 273$ Hz), 124.37 (CH, q,

 $J_{\rm FC} = 3.8$ Hz), 124.64 (CH, q, $J_{\rm FC} = 3.8$ Hz), 126.46 (CH, q, $J_{\rm FC} = 3.8$ Hz), 127.19 (CH), 128.19 (CH), 128.53 (CH), 128.89 (CH), 129.05 (CH), 130.84 (CH), 130.92 (C, q, $J_{\rm FC} = 32$ Hz), 130.96 (CH), 130.99 (C, q, $J_{\rm FC} = 32$ Hz), 133.17 (CH), 139.59 (C), 140.00 (C), 141.05 (C), 142.31 (C); ¹⁹F NMR (376 MHz, CDCl₃) δ (ppm) -62.62, -62.69; IR (neat) 3030, 2836, 1608, 1589, 1495, 1331, 1168, 1125, 1074 cm⁻¹; MS (EI) *m/z* 435 (M⁺, 24), 344 (18), 132 (33), 91 (100%); HRMS (EI) *m/z* 435.1418 (calcd for C₂₄H₁₉F₆N 435.1422).

4f: (2 mmol scale, 234 mg, 35%); $R_f = 0.2$ (hexane-ether = 1 : 4); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.44 (bs, 1H), 2.31 (s, 3H), 2.37 (s, 3H), 3.35 (d, J = 6.8 Hz, 2H), 3.74 (s, 2H), 6.14 (t, J = 6.8 Hz, 1H), 7.02-7.07 (m, 4H), 7.12-7.15 (m, 4H), 7.19-7.30 (m, 5H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 21.13 (CH₃), 21.31 (CH₃), 48.24 (CH₂), 53.59 (CH₂), 126.73 (CH), 126.92 (CH), 127.41 (CH), 128.20 (CH), 128.40 (CH), 128.85 (CH), 129.72 (CH), 136.83 (C), 136.85 (C), 136.99 (C), 139.69 (C), 140.41 (C), 143.38 (C); IR (neat) 3316, 3024, 2919, 1512, 1453, 1111 cm⁻¹; MS (EI) *m/z* 327 (M⁺, 26), 236 (27), 195 (42), 119 (69), 91 (100%); HRMS (EI) *m/z* 327.1985 (calcd for C₂₄H₂₅N 327.1987).

4g: (2.9 mmol scale, 365 mg, 35%); $R_f = 0.2$ (hexane-ether = 1 : 4); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.48 (bs, 1H), 3.35 (d, J = 7.0 Hz, 2H), 3.75 (s, 2H), 3.79 (s, 3H), 3.84 (s, 3H), 6.07 (t, J = 7.0 Hz, 1H), 6.80 (d-like, J = 8.9 Hz, 2H), 6.87 (d-like, J = 8.8 Hz 2H), 7.07 (d-like, J = 8.8 Hz, 2H), 7.17 (d-like, J = 8.9 Hz, 2H), 7.21-7.31 (m, 5H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 48.26 (CH₂), 53.62 (CH₂), 55.31 (CH₃), 55.33 (CH₃), 113.51 (CH), 113.53 (CH), 125.76 (CH), 126.94 (CH), 128.22 (CH), 128.42 (CH), 128.68 (CH), 130.99 (CH), 132.20 (C), 135.32 (C), 140.41 (C), 142.69 (C), 158.77 (C), 159.01 (C); IR (neat) 3328, 2953, 2834, 1606, 1511, 1246, 1173, 1035 cm⁻¹; MS (EI) *m*/*z* 359 (M⁺, 100), 268 (52%); HRMS (EI) *m*/*z* 359.1878 (calcd for C₂₄H₂₅NO₂ 359.1887).

4h: (3.1 mmol scale, 515 mg, 50%); Rf = 0.4 (hexane-ether = 1 : 4); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.49 (bs, 1H), 3.17-3.21 (m, 2H), 3.73 (s, 2H), 6.41 (t, *J* = 6.8 Hz,

-67-

1H), 7.14-7.16 (m, 1H), 7.17-7.29 (m, 12H), 7.41-7.44 (m, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 48.27 (CH₂), 53.59 (CH₂), 126.35 (CH), 126.78 (CH), 126.91 (CH), 127.40 (CH), 128.08 (CH), 128.34 (CH), 128.37 (CH), 128.82 (CH), 129.00 (CH), 129.72 (CH), 131.49 (CH), 133.73 (C), 138.22 (C), 139.99 (C), 140.25 (C), 140.35 (C); IR (neat) 3026, 2835, 1597, 1495, 1445, 1362, 1125, 1050 cm⁻¹; MS (EI) *m/z* 335 (M⁺, 7.5), 333 (M⁺, 18), 298 (42), 91 (100%); HRMS (EI) *m/z* 333.1282, 335.1245 (calcd for C₂₂H₂₀CIN 333.1284, 335.1255).

4i: (3.5 mmol scale, 531 mg, 45%); $R_f = 0.5$ (hexane-AcOEt = 1 : 4); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.59 (bs, 1H), 3.33 (d, J = 6.8 Hz, 2H), 3.75 (s, 2H), 6.21 (t, J = 6.8 Hz, 1H), 7.08 (d-like, J = 8.4 Hz, 2H), 7.19-7.33 (m, 12H). Selected NOEs are between δ 3.33 (CH₂-CH=) and δ 7.08 (2'-*H* of 4-chlorophenyl).; ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 48.01 (CH₂), 53.59 (CH₂), 127.09 (CH), 127.47 (CH), 127.55 (CH), 128.21 (CH), 128.31 (CH), 128.49 (CH), 131.21 (CH), 133.25 (C), 138.08 (C), 140.14 (C), 141.82 (C), 142.55 (C); IR (neat) 3316, 3027, 2833, 1597, 1489, 1445, 1090, 1015 cm⁻¹; MS (EI) *m/z* 335 (M⁺, 16), 333 (M⁺, 16), 242 (41), 132 (49), 91 (100%); HRMS (EI) *m/z* 333.1277, 335.1266 (calcd for C₂₂H₂₀ClN 333.1284, 335.1255).

4j: (2.4 mmol scale, 805 mg, 41%); $R_f = 0.3$ (hexane-AcOEt = 1 : 1); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.50 (bs, 1H), 3.34 (d, J = 6.8 Hz, 2H), 3.74 (s, 2H), 6.18 (t, J = 6.8 Hz, 1H), 7.11-7.16 (m, 4H), 7.20-7.37 (m, 10H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 48.14 (CH₂), 53.63 (CH₂), 127.02 (CH), 127.51 (CH), 128.18 (CH), 128.32 (CH), 128.34 (CH), 128.45 (CH), 128.73 (CH), 129.74 (CH), 133.15 (C), 139.18 (C), 140.24 (C), 140.73 (C), 142.46 (C); IR (neat) 3323, 3027, 2836, 1599, 1488, 1453, 1442, 1092, 1012 cm⁻¹; MS (EI) *m/z* 335 (M⁺, 23), 333 (M⁺, 65), 242 (55), 132 (69), 91 (100%); HRMS (EI) *m/z* 333.1279, 335.1266 (calcd for C₂₂H₂₀ClN 333.1284, 335.1255).

4k: (2.6 mmol scale, 432 mg, 48%); $R_f = 0.2$ (hexane-AcOEt = 1 : 4); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.54 (bs, 1H), 3.31 (d, J = 7.0 Hz, 2H), 3.76 (s, 2H), 6.30 (t, J = 7.0

-68-
Hz, 1H), 7.16-7.18 (m, 2H), 7.23-7.32 (m, 10H), 8.18 (d-like, J = 8.8 Hz, 2H). Selected NOEs are between δ 6.30 (CH=CPh) and δ 7.16-7.18 (2-*H* of CH=CPh).; ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 47.78 (CH₂), 53.53 (CH₂), 123.48 (CH), 127.12 (CH), 127.41 (CH), 127.88 (CH), 128.13 (CH), 128.45 (CH), 129.69 (CH), 130.72 (CH), 139.88 (C), 140.94 (C), 141.75 (C), 146.64 (C), 147.02 (C); IR (neat) 3026, 2838, 1598, 1516, 1346, 1107 cm⁻¹; MS (EI) *m/z* 344 (M⁺, 31), 253 (22), 132 (35), 91 (100%); HRMS (EI) *m/z* 344.1521 (calcd for C₂₂H₂₀N₂O₂ 344.1525).

Typical experimental procedure for 5 (Table 4-2, entry 2). To a solution of 1,1-diethyl 2-hydrogen ethenetricarboxylate (1) (prepared from 1,1-diethyl 2-*tert*-butyl ethenetricarboxylate (136 mg, 0.5 mmol) upon treatment with CF₃CO₂H (2 mL))¹⁷ in DMF (0.7 mL) were added *N*-benzyl 3,3-diphenyl-2-propen-1-amine (**4a**) (150 mg, 0.5 mmol) in DMF (0.7 mL), Et₃N (0.07 mL, 51 mg, 0.5 mmol), HOBt (1-hydroxybenzotriazole) (135 mg, 1 mmol), and EDCI (1-[3-(dimethylamino)propyl]-3-ethylcarbodiimide hydrochloride) (100 mg, 0.5 mmol) at 0 °C. The reaction mixture was stirred for 1 h at 0 °C, and was allowed to warm to room temperature and stirred for 20 h. The reaction mixture was diluted with CH₂Cl₂. The organic phase was washed with saturated aqueous NaHCO₃ solution, 2 M aqueous citric acid, saturated aqueous NaHCO₃ and water, dried (Na₂SO₄), and evaporated *in vacuo*. The residue was purified by column chromatography over silica gel eluting with hexane-Et₂O to give **5a-***trans* (129 mg, 52%).

5a-trans: $R_f = 0.4$ (hexane-ether = 1 : 4); colorless crystals; mp 159.5-160 °C (hexane-AcOEt); ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.25 (t, J = 7.1 Hz, 3H), 1.38 (t, J = 7.1 Hz, 3H), 2.60 (dd, J = 9.6, 9.0 Hz, 1H), 2.96 (dddd, J = 13.5, 9.6, 8.0, 6.3 Hz, 1H), 3.24 (dd, J = 9.0, 8.0 Hz, 1H), 3.48 (d, J = 13.5 Hz, 1H), 3.78 (d, J = 15.0 Hz, 1H), 4.16 (qd, J = 10.8, 7.1 Hz, 1H), 4.29-4.48 (m, 4H), 4.90 (d, J = 15.0 Hz, 1H), 6.90 (d, J = 6.8 Hz, 2H), 6.98 (d, J = 7.6 Hz, 1H), 7.18-7.32 (m, 10H), 7.53 (dd, J = 8.0, 1.0 Hz, 1H). Selected NOEs are between δ 2.96 (C3a-H) and δ 3.24 (C3-HH), and between δ 2.60 (C3-*H*H) and δ 3.48 (C9a-*H*), and between δ 2.60 (C3-*H*H) and δ 6.90 (*o*-H of C4-*Ph*). Atom numbering is shown in Table 4-3.; ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 14.00 (CH₃), 14.13 (CH₃), 36.37 (CH), 43.81 (CH), 46.15 (CH₂), 46.91 (CH), 47.32 (CH₂), 60.32 (C), 62.03 (CH₂), 62.51 (CH₂), 126.99 (CH), 127.01 (CH), 127.39 (CH), 127.92 (CH), 128.41 (CH), 128.50 (CH), 128.61 (CH), 130.25 (CH), 130.52 (CH), 131.45 (CH), 134.64 (CH), 136.80 (C), 138.80 (C), 141.18 (C), 168.44 (C), 170.96 (C), 171.86 (C). Selected HMBC correlations are between δ 2.96 (C9a-*H*) and δ 36.37 (*C*3a), between δ 2.60 (C3-*H*H), 2.96 (C3a-*H*), 6.98 (C5-*H*) and δ 46.91 (*C*4), and between δ 2.96 (C3a-*H*), 3.48 (C9a-*H*) and δ 60.32 (*C*9).; IR (KBr) 2977, 1735, 1691, 1496, 1441, 1249, 1025 cm⁻¹; MS (EI) *m/z* 497 (M⁺, 75), 394 (45), 91 (100%); HRMS (EI) *m/z* 497.2201 (calcd for C₃₁H₃₁NO₅ 497.2202).

5a-*cis*: (1 mmol scale, benzene, 80 °C, 247 mg, 48%); $R_f = 0.2$ (hexane-ether = 1 : 8); pale yellow viscous oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.15 (t, J = 7.0 Hz, 3H), 1.36 (t, J = 7.2 Hz, 3H), 3.00 (dd, J = 9.6, 7.4 Hz, 1H), 3.14 (dd, J = 9.6, 8.2 Hz, 1H), 3.35 (dddd, J = 10.0, 8.2, 7.4, 5.1 Hz, 1H), 4.09 (m, 1H), 4.10 (d, J = 15.2 Hz, 1H), 4.20 (d, J = 10.0 Hz, 1H), 4.18-4.27 (m, 1H), 4.29 (d, J = 5.1 Hz, 1H), 4.31 (d, J = 15.2 Hz, 1H), 4.33-4.46 (m, 2H), 6.87 (d, J = 7.8 Hz, 1H), 6.93-6.95 (m, 2H), 7.06-7.08 (m, 2H), 7.13-7.33 (m, 8H), 8.09 (dd, J = 8.0, 1.0 Hz, 1H). Selected NOEs are between δ 3.35 (C3a-*H*) and δ 4.20 (C9a-*H*), 4.29 (C4-*H*) (overlapped), and between δ 4.29 (C4-*H*) and δ 6.87 (C5-*H*), 7.06-7.08 (*o*-H of C4-*Ph*).; ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 13.80 (CH₃), 14.05 (CH₃), 36.38 (CH), 45.08 (CH), 46.40 (CH₂), 47.58 (CH), 48.49 (CH₂), 59.24 (C), 61.94 (CH₂), 62.39 (CH₂), 126.66 (CH), 127.19 (CH), 127.31 (CH), 127.57 (CH), 127.76 (CH), 128.58 (CH), 128.66 (CH), 129.84 (CH), 130.94 (CH), 132.54 (C), 135.93 (C), 138.18 (C), 139.90 (C), 168.70 (C), 169.71 (C), 172.70 (C). Selected HMBC correlations are between δ 3.00 (C3-*H*H), 3.52 (C3-H*H*), 4.20 (C9a-*H*), 4.29 (C4-*H*) and δ 36.38 (C3a), between δ 3.00 (C3-*H*H), 3.52 (C3-H*H*), 7.06-7.08 (*o*-H of C4-*Ph*), 4.29 (C4-*H*) and δ 36.38 (C3a), between δ 3.00 (C3-*H*H), 3.52 (C3-H*H*), 7.06-7.08 (*o*-H of C4-*Ph*) and δ 36.38 (C3a), between δ 3.00 (C3-*H*H), 3.52 (C3-H*H*), 7.06-7.08 (*o*-H of C4-*Ph*) and δ 36.38 (C3a), between δ 3.00 (C3-*H*H), 3.52 (C3-H*H*), 4.20 (C9a-*H*), 4.29 (C4-*H*) and δ 36.38 (C3a), between δ 3.00 (C3-*H*H), 3.52 (C3-H*H*), 7.06-7.08 (*o*-H of C4-*Ph*) and δ 45.08 (C4), and between δ 4.20 (C9a-*H*) and δ 59.24 (C9).; IR (KBr) 2980, 1729, 1694, 1495,

1445, 1239, 1029 cm⁻¹; MS (EI) *m/z* 497 (M⁺, 46), 423 (6.4), 350 (13), 276 (22), 91 (100%); HRMS (EI) *m/z* 497.2195 (calcd for C₃₁H₃₁NO₅ 497.2202).

5b-*cis*: (1 mmol scale, benzene, 80 °C, 277 mg, 55%); $R_f = 0.2$ (hexane-ether = 1 : 1); pale yellow crystals; mp 47.5-48.0 °C; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 0.669-0.825 (m, 2H), 0.959-1.11 (m, 3H), 1.14 (t, J = 7.1 Hz, 3H), 1.22-1.26 (m, 1H), 1.33 (t, J = 7.2 Hz, 3H), 1.36-1.43 (m, 2H), 1.51-1.61 (m, 3H), 2.72 (dd, *J* = 13.6, 7.3 Hz, 1H), 2.95 (dd, *J* = 13.6, 7.5 Hz, 1H), 3.05 (dd, J = 9.3, 7.5 Hz, 1H), 3.26 (dd, J = 9.3, 8.2 Hz, 1H), 3.36 (dddd, J = 9.6, 8.2, 7.5, 1H)4.9 Hz, 1H), 4.01-4.09 (m, 1H), 4.12 (d, J = 9.6 Hz, 1H), 4.18-4.26 (m, 1H), 4.31 (d, J = 4.9 Hz, 1H), 4.31-4.43 (m, 2H), 6.92 (d, J = 7.6 Hz, 1H), 7.15-7.18 (m, 3H), 7.24-7.33 (m, 2H), 7.36-7.40 (m, 2H), 8.04 (dd, J = 8.0, 1.2 Hz, 1H). Selected NOEs are between δ 3.36 (C3a-H) and δ 4.12 (C9a-H), 4.31 (C4-H), and between δ 4.31 (C4-H) and δ 6.92 (C5-H).; ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 13.80 (CH₃), 14.05 (CH₃), 25.75 (CH₂), 26.38 (CH₂), 30.34 (CH₂), 30.61 (CH₂), 35.55 (CH), 36.45 (CH), 45.09 (CH), 47.73 (CH), 48.92 (CH2), 49.83 (CH2), 59.13 (C), 61.81 (CH2), 62.35 (CH2), 126.64 (CH), 127.23 (CH), 127.48 (CH), 127.67 (CH), 128.73 (CH), 129.86 (CH), 130.97 (CH), 132.66 (C), 138.20 (C), 140.11 (C), 168.56 (C), 169.81 (C), 172.60 (C). Selected HMBC correlations are between δ 3.05 (C3-HH), 3.26 (C3-HH), 4.12 (C9a-H), 4.31 (C4-*H*) and δ 36.45 (C3a), between δ 3.05 (C3-*H*H), 6.92 (C5-*H*) and δ 45.09 (C4), and between δ 4.12 (C9a-H) and δ 59.13 (C9).; IR (KBr) 2924, 1730, 1695, 1448, 1238, 1038 cm⁻¹; MS (EI) m/z 503 (M⁺, 5.3), 205 (6.9), 86 (100%); HRMS (EI) m/z 503.2672 (calcd for C₃₁H₃₇NO₅ 503.2672).

5*c*-*trans*: (0.5 mmol scale, DMF, r.t., 141 mg, 53%); $R_f = 0.2$ (hexane-ether = 1 : 8); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.27 (t, J = 7.1 Hz, 3H), 1.39 (t, J = 7.1 Hz, 3H), 2.58 (dd, J = 9.7, 9.1 Hz, 1H), 2.94 (dddd, J = 13.5, 9.7, 7.7, 6.1 Hz, 1H), 3.24 (dd, J = 9.1, 7.7 Hz, 1H), 3.37 (d, J = 13.5 Hz, 1H), 3.83 (d, J = 15.0 Hz, 1H), 4.17 (dq, J = 10.7, 7.1 Hz, 1H), 4.30-4.50 (m, 4H), 4.88 (d, J = 15.0 Hz, 1H), 6.86 (dd, J = 8.6, 5.3 Hz, 2H), 6.93-7.00 (m, 4H),

7.21-7.32 (m, 6H). Selected NOEs are between δ 2.94 (C3a-*H*) and δ 3.24 (C3-H*H*) and between δ 2.58 (C3-*H*H) and δ 3.37 (C9a-*H*), and between δ 2.58 (C3-*H*H) and δ 6.86 (*o*-H of C4-*Ar*).; ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 13.98 (CH₃), 14.12 (CH₃), 36.42 (CH), 43.48 (CH), 45.57 (CH), 46.17 (CH₂), 47.13 (CH₂), 60.14 (C), 62.29 (CH₂), 62.85 (CH₂), 115.54 (CH, d, *J*_{FC} = 21 Hz), 116.28 (CH, d, *J*_{FC} = 21 Hz), 116.92 (CH, d, *J*_{FC} = 23 Hz), 127.50 (CH), 127.91 (CH), 128.66 (CH), 131.54 (CH, d, *J*_{FC} = 8.4 Hz), 132.74 (CH, d, *J*_{FC} = 7.7 Hz), 134.47 (C, d, *J*_{FC} = 3.1 Hz), 136.43 (C, d, *J*_{FC} = 7.7 Hz), 136.55 (C), 136.57 (C), 161.25 (C, d, *J*_{FC} = 246 Hz), 161.85 (C, d, *J*_{FC} = 247 Hz), 167.79 (C), 170.48 (C), 171.38 (C). Selected HMBC correlations are between δ 2.58 (C3-*H*H), 3.24 (C3-H*H*) and δ 36.42 (C3a), between δ 6.86 (*o*-H of C4-*Ph*) and δ 45.57 (*C*4), and between δ 2.94 (C3a-*H*), 3.37 (C9a-*H*) and δ 60.14 (C9).; ¹⁹F NMR (376 MHz, CDCl₃) δ (ppm) -114.56 (ddd, *J* = 10.3, 6.9, 6.9 Hz), -115.29 (m); IR (KBr) 2979, 1735, 1691, 1507, 1495, 1260, 1161, 1028 cm⁻¹; MS (EI) *m*/z 533 (M⁺, 18), 430 (12), 57 (100%); HRMS (EI) *m*/z 533.2013 (calcd for C₃₁H₂₉F₂NO₅ 533.2014).

5c-*cis*: (0.54 mmol scale, toluene, 110 °C, 282 mg, 98%); $R_f = 0.5$ (hexane-ether = 1 : 4); pale yellow viscous oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.17 (t, J = 7.0 Hz, 3H), 1.37 (t, J = 7.1 Hz, 3H), 2.93 (dd, J = 9.6, 7.0 Hz, 1H), 3.13 (dd, J = 9.6, 8.3 Hz, 1H), 3.33 (dddd, J = 9.8, 8.3, 7.0, 5.1 Hz, 1H), 4.03-4.11 (m, 1H), 4.18-4.28 (m, 5H), 4.34-4.50 (m, 2H), 6.79 (dd, J = 8.5, 6.0 Hz, 1H), 6.88 (ddd, J = 8.5, 8.2, 2.7 Hz, 1H), 6.92-6.94 (m, 2H), 7.00-7.02 (m, 4H), 7.19-7.22 (m, 3H), 8.01 (dd, $J_{FH} = 11.0$, $J_{HH} = 2.7$ Hz, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 13.80 (CH₃), 14.05 (CH₃), 36.13 (CH), 43.78 (CH), 46.36 (CH₂), 47.36 (CH), 48.08 (CH₂), 58.77 (C), 62.18 (CH₂), 62.67 (CH₂), 114.79 (CH, d, $J_{FC} = 21$ Hz), 115.66 (CH, d, $J_{FC} = 21$ Hz), 118.07 (CH, d, $J_{FC} = 25$ Hz), 127.49 (CH), 127.56 (CH), 128.65 (CH), 128.69 (CH, d, $J_{FC} = 9.2$ Hz), 131.27 (CH, d, $J_{FC} = 8.4$ Hz), 133.71 (C, d, $J_{FC} = 3.1$ Hz), 134.29 (C, d, $J_{FC} = 8.4$ Hz), 135.14 (C, d, $J_{FC} = 3.1$ Hz), 135.62 (C), 161.52 (C, d, $J_{FC} = 245$ Hz), 161.92 (C, d, $J_{FC} = 246$ Hz), 168.15 (C), 168.95 (C), 172.25 (C); ¹⁹F NMR (376 MHz, CDCl₃) δ (ppm) -115.06 ~ -114.93 (m, 2F); ¹H

NMR (400 MHz, C₆D₆) δ (ppm) 0.787 (t, J = 7.0 Hz, 3H), 1.16 (t, J = 7.2 Hz, 3H), 2.62 (dd, J = 9.4, 8.4 Hz, 1H), 2.69 (dd, J = 9.4, 7.0 Hz, 1H), 2.85 (dddd, J = 9.6, 8.4, 7.0, 4.9 Hz, 1H), 3.70 (dq, J = 10.7, 7.0 Hz, 1H), 3.94 (d, J = 15.0 Hz, 1H), 4.02 (dq, J = 10.7, 7.0 Hz, 1H), 4.14 (d, J = 10.7, 7.0 Hz), 4.14 (d, J = 10.7, 7.0 Hz),15.0 Hz, 1H), 4.21 (d, J = 4.9 Hz, 1H), 4.38 (d, J = 9.6 Hz, 1H), 4.38-4.51 (m, 2H), 6.56-6.60 (m, 3H), 6.65-6.70 (m, 3H), 6.88-6.96 (m, 3H), 7.01-7.05 (m, 2H), 8.54 (dd, $J_{\text{FH}} = 11.1$, $J_{\text{HH}} = 2.7$ Hz, 1H). Selected NOEs are between δ 2.85 (C3a-H) and δ 4.38 (C9a-H), 4.21 (C4-H), and between δ 4.21 (C4-*H*) and δ 4.38 (C9a-*H*).; ¹³C NMR (100.6 MHz, C₆D₆) δ (ppm) 13.57 (CH₃), 13.98 (CH₃), 36.44 (CH), 44.09 (CH), 46.37 (CH₂), 47.54 (CH), 47.85 (CH₂), 59.30 (C), 62.36 (CH₂), 62.47 (CH₂), 114.86 (CH, d, *J*_{FC} = 21 Hz), 115.61 (CH, d, *J*_{FC} = 21 Hz), 118.68 (CH, d, *J*_{FC} = 25 Hz), 127.59 (CH), 127.72 (CH), 128.78 (CH), 129.04 (CH, d, *J*_{FC} = 7.7 Hz), 131.45 (CH, d, *J*_{FC} = 7.7 Hz), 134.45 (C, d, $J_{FC} = 3.1$ Hz), 135.30 (C, d, $J_{FC} = 8.4$ Hz), 135.58 (C, d, $J_{FC} = 3.1$ Hz), 136.41 (C), 161.97 (C, d, $J_{FC} = 244$ Hz), 162.14 (C, d, $J_{FC} = 245$ Hz), 168.29 (C), 169.31 (C), 171.83 (C). Selected HMBC correlations are between δ 2.69 (C3-HH), 4.38 (C9a-H), 4.21 (C4-H) and δ 36.44 (C3a), between δ 2.69 (C3-HH) and δ 44.09 (C4), and between δ 2.85 (C3a-H), 4.38 (C9a-H), 8.54 (C8-H) and δ 59.30 (C9).; ¹⁹F NMR (376 MHz, C₆D₆) δ (ppm) -114.89 (m, 1F), -115.20 (m, 1F); IR (neat) 2982, 1732, 1699, 1604, 1511, 1445, 1161, 1039 cm⁻¹; MS (EI) *m/z* 533 (M⁺, 52), 312 (29), 267 (25), 253 (23), 91 (100%); HRMS (EI) *m/z* 533.2019 (calcd for C₃₁H₂₉NO₅ 533.2014).

5d-*trans*: (1 mmol scale, DMF, r.t., 526 mg, 93%); $R_f = 0.6$ (hexane-ether = 1 : 4); colorless crystals; mp 130-131 °C (hexane-AcOEt); ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.26 (t, J = 7.1 Hz, 3H), 1.39 (t, J = 7.1 Hz, 3H), 2.58 (dd, J = 9.7, 9.1 Hz, 1H), 2.93 (dddd, J = 13.5, 9.7, 7.7, 6.1 Hz, 1H), 3.24 (dd, J = 9.1, 7.7 Hz, 1H), 3.34 (d, J = 13.5 Hz, 1H), 3.80 (d, J = 14.9 Hz, 1H), 4.14-4.22 (m, 1H), 4.29-4.50 (m, 3H), 4.33 (d, J = 6.1 Hz, 1H), 4.89 (d, J = 14.9 Hz, 1H), 6.83 (d, J = 8.4 Hz, 2H), 6.89 (d, J = 8.4 Hz, 1H), 7.18 (dd, J = 8.4, 2.1 Hz, 1H), 7.22-7.32 (m, 7H), 7.52 (d, J = 2.1 Hz, 1H). Selected NOEs are between δ 2.93 (C3a-*H*) and δ 3.24 (C3-H*H*), 4.33 (C4-*H*)

and between δ 2.58 (C3-*H*H) and δ 3.34 (C9a-*H*), 6.83 (*o*-H of C4-*Ar*).; ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 13.95 (CH₃), 14.09 (CH₃), 36.21 (CH), 43.56 (CH), 45.90 (CH), 46.21 (CH₂), 47.07 (CH₂), 60.06 (C), 62.29 (CH2), 62.86 (CH₂), 127.51 (CH), 127.92 (CH), 128.66 (CH), 128.83 (CH), 128.91 (CH), 130.49 (CH), 131.39 (CH), 132.46 (CH), 132.94 (C), 133.28 (C), 136.32 (C), 136.58 (C), 136.90 (C), 139.07 (C), 167.73 (C), 170.33 (C), 171.17 (C). Selected HMBC correlations are between δ 2.58 (C3-*H*H), 3.24 (C3-H*H*), 3.34 (C9a-*H*) and δ 36.21 (*C*3a), between δ 6.83 (*o*-H of C4-*Ph*) and δ 45.90 (*C*4), and between δ 2.93 (C3a-*H*), 3.34 (C9a-*H*) and δ 60.06 (*C*9).; IR (KBr) 2979, 1734, 1686, 1596, 1488, 1442, 1364, 1253, 1186, 1013 cm⁻¹; MS (EI) *m*/*z* 567 (M⁺, 50), 565 (72), 464 (26), 462 (31), 91 (100%); HRMS (EI) *m*/*z* 565.1407, 567.1392 (calcd for C₃₁H₂₉Cl₂NO₅ 565.1423, 567.1393).

5e-*trans*: (1.02 mmol scale, THF, r.t., 644 mg, 99%); $R_f = 0.6$ (hexane-ether = 1 : 4); colorless crystals; mp 134-135 °C; 1H NMR (400 MHz, CDCl₃) δ (ppm) 1.26 (t, J = 7.1 Hz, 3H), 1.37 (t, J = 7.1 Hz, 3H), 2.53 (dd, J = 9.7, 9.2 Hz, 1H), 3.01 (dddd, J = 13.7, 9.7, 7.8, 5.9 Hz, 1H), 3.31 (dd, J = 9.2, 7.8 Hz, 1H), 3.37 (d, J = 13.7 Hz, 1H), 3.84 (d, J = 15.0 Hz, 1H), 4.18 (dq, J = 10.7, 7.1 Hz, 1H), 4.30-4.48 (m, 3H), 4.51 (d, J = 5.9 Hz, 1H), 4.88 (d, J = 15.0 Hz, 1H), 7.05 (d, J = 7.8 Hz, 1H), 7.18-7.32 (m, 7H), 7.43 (dd, J = 7.8, 7.7 Hz, 1H), 7.53-7.58 (m, 2H), 7.71 (d, J = 8.4 Hz, 1H). Selected NOEs are between δ 3.01 (C3a-H) and δ 3.31 (C3-HH) and between δ 2.53 (C3-HH) and δ 3.37 (C9a-H), 7.05 (6-H of C4-Ar).; ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 13.95 (CH₃), 14.00 (CH₃), 36.17 (CH), 43.37 (CH), 46.20 (CH₂), 46.84 (CH), 46.89 (CH₂), 60.24 (C), 62.50 (CH₂), 62.95 (CH₂), 123.63 (C, q, $J_{FC} = 272$ Hz), 123.85 (C, q, $J_{FC} = 273$ Hz), 124.12 (CH, q, $J_{FC} = 3.1$ Hz), 124.46 (CH, q, $J_{FC} = 3.8$ Hz), 126.59 (CH, q, $J_{FC} = 3.8$ Hz), 127.57 (CH), 127.93 (CH), 128.08 (CH, q, $J_{FC} = 3.8$ Hz), 128.68 (CH), 129.46 (CH), 130.89 (C, q, $J_{FC} = 33$ Hz), 131.12 (C, q, $J_{FC} = 3.8$ Hz), 131.61 (CH), 133.46 (CH), 136.41 (C), 138.46 (C), 138.77 (C), 141.07 (C), 167.53 (C), 170.05 (C), 170.97 (C). Selected HMBC correlations are between δ 2.53 (C3-HH) and δ 3.617 (C3a), between δ 7.05 (6-H of C4-Ar) and δ 46.84 (C4), and

between δ 3.01 (C3a-*H*), 3.37 (C9a-*H*), 7.71 (C8-*H*) and δ 60.24 (*C*9).; ¹⁹F NMR (376 MHz, CDCl₃) δ (ppm) -62.70, -62.89; IR (KBr) 3062, 2984, 1739, 1704, 1620, 1607, 1415, 1330, 1253, 1166, 1132 cm⁻¹; MS (EI) *m/z* 633 (M⁺, 84), 530 (29), 149 (14), 118 (16), 91 (100%); HRMS (EI) *m/z* 633.1951 (calcd for C₃₃H₂₉F₆NO₅ 633.1950); Anal. Calcd for C₃₃H₂₉F₆NO₅: C, 62.56; H, 4.61; N, 2.21. Found: C, 62.60; H, 4.66; N, 2.21.

5f-*cis*: (1 mmol scale, DMF, 110 °C, 292 mg, 55%); $R_f = 0.7$ (hexane-ether = 1 : 4); pale yellow viscous oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.16 (t, J = 7.0 Hz, 3H), 1.36 (t, J = 7.2 Hz, 3H), 2.33 (s, 3H), 2.39 (s, 3H), 3.00 (d, *J* = 9.6, 7.6 Hz, 1H), 3.12 (dd, *J* = 9.6, 8.3 Hz, 1H), 3.30 (dddd, J = 9.8, 8.3, 7.6, 5.1 Hz, 1H), 4.06 (dq, J = 10.7, 7.0 Hz, 1H), 4.11 (d, J = 15.2 Hz, 1H), 4.18 = 9.8 Hz, 1H), 4.22 (d, J = 5.1 Hz, 1H), 4.25 (dq, J = 10.7, 7.0 Hz, 1H), 4.35 (d, J = 15.2 Hz, 1H), 4.35-4.46 (m, 2H), 6.78 (d, J = 7.8 Hz, 1H), 6.92-6.97 (m, 5H), 7.11 (d, J = 7.8 Hz, 2H), 7.16-7.19 (m, 3H), 7.94 (d, J = 1.0 Hz, 1H). Selected NOEs are between δ 3.30 (C3a-H) and δ 4.18 (C9a-H), 4.22 (C4-H), 3.12 (C3-HH).; ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 13.78 (CH₃), 14.04 (CH₃), 21.02 (CH₃), 21.42 (CH₃), 36.36 (CH), 44.19 (CH), 46.29 (CH₂), 47.68 (CH), 48.40 (CH₂), 59.12 (C), 61.82 (CH₂), 62.32 (CH₂), 127.26 (CH), 127.33 (CH), 127.51 (CH), 128.47 (CH), 128.50 (CH), 129.26 (CH), 129.62 (CH), 131.21 (CH), 132.18 (C), 135.23 (C), 135.88 (C), 136.06 (C), 136.68 (C), 136.81 (C), 168.74 (C), 169.65 (C), 172.83 (C). Selected HMBC correlations are between δ 3.00 (C3-HH), 3.12 (C3-HH), 4.18 (C9a-H) and δ 36.36 (C3a), between δ 6.78 (C5-H) and δ 44.19 (C4), and between δ 4.18 (C9a-H), 7.94 (C8-H) and δ 59.12 (C9).; IR (KBr) 2980, 1730, 1696, 1495, 1444, 1237, 1041 cm⁻¹; MS (EI) *m/z* 525 (M⁺, 15), 304 (13), 84 (100%); HRMS (EI) *m/z* 525.2523 (calcd for C₃₃H₃₅NO₅ 525.2515).

5h-*trans*: (0.73 mmol scale, DMF, r.t., 126 mg, 33%); $R_f = 0.6$ (hexane-AcOEt = 1 : 1); colorless crystals; mp 183.0-183.9 °C; 1H NMR (400 MHz, CDCl₃) δ (ppm) 1.25 (t, J = 7.1 Hz, 3H), 1.38 (t, J = 7.1 Hz, 3H), 2.58 (dd, J = 9.9, 9.4 Hz, 1H), 3.03 (dddd, J = 13.5, 9.9, 7.8, 6.1 Hz, 1H), 3.43 (d, J = 13.5 Hz, 1H), 3.48 (dd, J = 9.4, 7.8 Hz, 1H), 3.79 (d, J = 15.2 Hz, 1H), 4.17 (dq, J = 10.7,

7.1 Hz, 1H), 4.29-4.49 (m, 3H), 4.91 (d, J = 15.2 Hz, 1H), 4.96 (d, J = 6.1 Hz, 1H), 6.69 (dd, J = 7.4, 2.0 Hz, 1H), 6.91 (dd, J = 7.8, 1.2 Hz, 1H), 7.11-7.33 (m, 9H), 7.39 (dd, J = 7.8, 1.6 Hz, 1H), 7.53 (dd, J = 8.0, 1.2 Hz, 1H). Selected NOEs are between δ 3.03 (C3a-*H*) and δ 3.48 (C3-H*H*), 4.96 (C4-*H*) and between δ 3.43 (C9a-*H*) and δ 2.58 (C3-*H*H), 6.69 (6-H of C4-*Ar*).; ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 14.02 (CH₃), 14.15 (CH₃), 36.20 (CH), 43.13 (CH), 43.83 (CH), 46.11 (CH₂), 47.53 (CH₂), 60.25 (C), 62.10 (CH₂), 62.59 (CH₂), 127.22 (CH), 127.31 (CH), 127.41 (CH), 127.89 (CH), 128.29 (CH), 128.57 (CH), 128.63 (CH), 129.23 (CH), 130.60 (CH), 131.42 (CH), 132.74 (CH), 134.85 (C), 134.99 (C), 136.82 (C), 138.47 (C), 139.00 (C), 168.30 (C), 171.02 (C), 171.67 (C). Selected HMBC correlations are between δ 2.58 (C3-*H*H), 3.48 (C3-*HH*), 3.43 (C9a-*H*), 4.96 (C4-*H*) and δ 36.20 (*C*3a), between δ 6.69 (6-H of C4-*Ar*) and δ 43.13 (*C*4), and between δ 3.43 (C9a-*H*), 7.53 (C8-*H*) and δ 60.25 (*C*9).; IR (KBr) 2976, 1747, 1724, 1702, 1433, 1250, 1044 cm⁻¹; MS (EI) *m/z* 533 (M⁺, 9.4), 531 (M⁺, 24), 428 (24), 202 (21), 91 (100%); HRMS (EI) *m/z* 531.1807, 533.1794 (calcd for C₃₁H₃₀CINO₅ 531.1813, 533.1783).

5h-*cis*: (1 mmol scale, toluene, 110 °C, 445 mg, 84% (**5h**-*cis*:**5h**-*trans*=56:28), **5h**-*cis* was partially isolated by removal of 5h-*trans* by crystallization of the mixture and the subsequent column chromatography of the filtrate. 116 mg, 22%); Rf = 0.7 (hexane-AcOEt = 1 : 1); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.18 (t, *J* = 7.1 Hz, 3H), 1.34 (t, *J* = 7.1 Hz, 3H), 2.89 (dd, *J* = 9.8, 6.4 Hz, 1H), 3.24 (dd, *J* = 9.8, 7.9 Hz, 1H), 3.43 (dddd, *J* = 9.6, 7.9, 6.4, 6.1 Hz, 1H), 3.94 (d, *J* = 15.0 Hz, 1H), 4.10-4.18 (m, 1H), 4.16 (d, *J* = 9.6 Hz, 1H), 4.20-4.28 (m, 1H), 4.32-4.45 (m, 2H), 4.33 (d, *J* = 15.0 Hz, 1H), 4.93 (d, *J* = 6.1 Hz, 1H), 6.75 (d, *J* = 7.8 Hz, 1H), 6.85 (dd, *J* = 7.8, 1.8 Hz, 1H), 6.90-6.93 (m, 2H), 7.12-7.21 (m, 6H), 7.30-7.34 (m, 1H), 7.42 (dd, *J* = 7.8, 1.6 Hz, 1H), 8.11 (dd, *J* = 8.0, 1.2 Hz, 1H). Selected NOEs are between δ 3.43 (C3a-*H*) and δ 4.16 (C9a-*H*), 4.93 (C4-*H*), 3.24 (C3-H*H*) and between δ 4.16 (C9a-*H*) and δ 4.93 (C4-*H*).; ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 13.89 (CH₃), 14.05 (CH₃), 34.38 (CH), 40.88 (CH), 46.36 (CH₂), 47.62 (CH), 48.59 (CH₂), 58.96 (C), 61.94 (CH₂), 62.54 (CH₂), 126.75 (CH),

126.80 (CH), 127.33 (CH), 127.60 (CH), 127.80 (CH), 127.92 (CH), 128.28 (CH), 128.59 (CH), 129.88 (CH), 130.48 (CH), 130.75 (CH), 132.62 (C), 135.13 (C), 135.87 (C), 137.31 (C), 138.35 (C), 168.64 (C), 169.82 (C), 172.36 (C). Selected HMBC correlations are between δ 2.89 (C3-*H*H), 3.24 (C3-H*H*), 4.16 (C9a-*H*), 4.93 (C4-*H*) and δ 34.38 (*C*3a), between δ 6.75 (C5-*H*) and δ 40.88 (C4), and between δ 4.16 (C9a-*H*), 8.11 (C8-*H*) and δ 58.96 (*C*9).; IR (neat) 2981, 1730, 1697, 1443, 1237, 1039 cm⁻¹; MS (EI) *m*/*z* 533 (M⁺, 22), 531 (M⁺, 56), 384 (28), 91 (100%); HRMS (EI) *m*/*z* 531.1805, 533.1780 (calcd for C₃₁H₃₀CINO₅ 531.1813, 533.1783);

5i-trans: (0.80 mmol scale, DMF, r.t., 306 mg, 72%); $R_f = 0.6$ (ether); colorless crystals; mp 154-156 °C (hexane-AcOEt); ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.25 (t, J = 7.1 Hz, 3H), 1.38 (t, J = 7.1 Hz, 3H), 2.59 (dd, J = 9.7, 9.1 Hz, 1H), 2.96 (dddd, J = 13.7, 9.7, 7.7, 6.1 Hz, 1H), 3.24 (dd, J = 9.1, 7.7 Hz, 1H), 3.38 (d, J = 13.7 Hz, 1H), 3.82 (d, J = 15.0 Hz, 1H), 4.15 (dg, J = 10.7),7.1 Hz, 1H), 4.28-4.48 (m, 4H), 4.89 (d, J = 15.0 Hz, 1H), 6.84 (d, J = 8.4 Hz, 2H), 6.94 (dd, J = 15.0 Hz, 1H), 6.84 (d, J7.8, 1.4 Hz, 1H), 7.19-7.32 (m, 9H), 7.53 (dd, J = 7.9, 1.3 Hz, 1H). Selected NOEs are between δ 2.96 (C3a-H) and δ 3.24 (C3-HH), between δ 3.38 (C9a-H) and δ 2.59 (C3-HH), 6.84 (o-H of C4-Ar), and between δ 2.59 (C3-HH) and δ 6.84 (o-H of C4-Ar).; ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 13.99 (CH₃), 14.13 (CH₃), 36.19 (CH), 43.79 (CH), 46.19 (CH₂), 46.31 (CH), 47.20 (CH₂), 60.21 (C), 62.10 (CH₂), 62.60 (CH₂), 127.25 (CH), 127.47 (CH), 127.94 (CH), 128.54 (CH), 128.66 (CH), 128.71 (CH), 130.65 (CH), 131.31 (CH), 131.50 (CH), 133.02 (C), 134.60 (C), 136.66 (C), 138.30 (C), 139.66 (C), 168.30 (C), 170.91 (C), 171.64 (C). Selected HMBC correlations are between δ 2.59 (C3-HH), 3.24 (C3-HH), 3.38 (C9a-H) and δ 36.19 (C3a), between δ 6.84 (o-H of C4-Ar) and δ 46.31 (C4), and between δ 3.38 (C9a-H), 7.53 (C8-H), 2.96 (C3a-H) and δ 60.21 (C9).; IR (KBr) 2979, 1735, 1693, 1490, 1256 cm⁻¹; MS (EI) m/z 533 (M⁺, 15), 531 (M⁺, 39), 428 (21), 91 (100%); HRMS (EI) *m/z* 531.1803, 533.1790 (calcd for C₃₁H₃₀ClNO₅ 531.1813, 533.1783).

-77-

5i-cis: (0.80 mmol scale, toluene, 110 °C, 166 mg, 39%); $R_f = 0.6$ (ether); colorless crystals; mp 157-159 °C (hexane-AcOEt); ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.15 (t, J = 7.1 Hz, 3H), 1.36 $(t, J = 7.1 \text{ Hz}, 3\text{H}), 2.95 \text{ (dd}, J = 9.6, 7.4 \text{ Hz}, 1\text{H}), 3.11 \text{ (dd}, J = 9.6, 8.2 \text{ Hz}, 1\text{H}), 3.31 \text{ (dddd}, J = 9.6, 8.2 \text{ Hz}, 1\text{Hz}), 3.31 \text{ (dddd}, J = 9.6, 8.2 \text{ Hz}, 1\text{Hz}), 3.31 \text{ (dddd}, J = 9.6, 8.2 \text{ Hz}, 1\text{Hz}), 3.31 \text{ (dddd}, J = 9.6, 8.2 \text{ Hz}, 1\text{Hz}), 3.31 \text{ (dddd}, J = 9.6, 8.2 \text{ Hz}, 1\text{Hz}), 3.31 \text{ (dddd}, J = 9.6, 8.2 \text{ Hz}), 3.31 \text{ (dddd}, J = 9.6, 8.2 \text{ Hz}), 3.31 \text{ (dddd}, J = 9.6, 8.2 \text{ Hz}), 3.31 \text{ (dddd}, J = 9.6, 8.2 \text{ Hz}), 3.31 \text{ (dddd}, J = 9.6, 8.2 \text{ Hz}), 3.31 \text{ (ddddd}, J = 9.6, 8.2 \text{ Hz}), 3.31 \text{ (ddddd}, J = 9.6, 8.2 \text$ 9.6, 8.2, 7.4, 5.1 Hz, 1H), 4.06 (dq, J = 10.7, 7.1 Hz, 1H), 4.11 (d, J = 14.9 Hz, 1H), 4.17 (d, J = 9.6 Hz, 1H), 4.23 (dq, J = 10.7, 7.1 Hz, 1H), 4.27 (d, J = 5.1 Hz, 1H), 4.31 (d, J = 14.9 Hz, 1H), 4.33-4.47 (m, 2H), 6.83 (d, J = 7.8 Hz, 1H), 6.93-6.95 (m, 2H), 7.00 (d-like, J = 8.4 Hz, 2H), 7.15-7.21 (m, 4H), 7.25-7.34 (m, 3H), 8.09 (dd, J = 8.1, 1.1 Hz, 1H). Selected NOEs are between δ 3.11 (C3a-H) and δ 4.17 (C9a-H), 4.27 (C4-H), 3.11 (C3-HH) and between δ 4.17 (C9a-H) and δ 4.27 (C4-H).; ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 13.82 (CH₃), 14.08 (CH₃), 36.26 (CH), 44.45 (CH), 46.40 (CH₂), 47.41 (CH), 48.30 (CH₂), 59.10 (C), 62.04 (CH₂), 62.50 (CH₂), 126.92 (CH), 127.44 (CH), 127.60 (CH), 127.92 (CH), 128.66 (CH), 128.84 (CH), 131.07 (CH), 131.17 (CH), 132.39 (C), 133.09 (C), 135.78 (C), 137.62 (C), 138.38 (C), 168.67 (C), 169.66 (C), 172.60 (C). Selected HMBC correlations are between δ 2.95 (C3-HH), 3.11 (C3-HH), 4.17 (C9a-H) and δ 36.26 (C3a), between δ 6.83 (C5-H) and δ 44.45 (C4), and between δ 4.17 (C9a-H), 8.09 (C8-H) and δ 59.10 (C9).; IR (KBr) 2978, 2935, 1731, 1685, 1492, 1450, 1292, 1231, 1094, 1036 cm⁻¹; MS (EI) m/z 533 (M⁺, 45), 531 (M⁺, 23), 430 (23), 149 (34), 91 (82), 57 (100%); HRMS (EI) m/z 531.1809 (calcd for C₃₁H₃₀ClNO₅ 531.1813).

5*j*-*trans*: (0.43 mmol scale, DMF, r.t., 229 mg, 86%); $R_f = 0.7$ (hexane-AcOEt = 1 : 1); colorless crystals; mp 182.0-182.5 °C (hexane-AcOEt); 1H NMR (400 MHz, CDCl₃) δ (ppm) 1.27 (t, J = 7.1 Hz, 3H), 1.39 (t, J = 7.1 Hz, 3H), 2.59 (dd, J = 9.7, 9.0 Hz, 1H), 2.92 (dddd, J = 13.5, 9.7, 7.8, 6.3 Hz, 1H), 3.24 (dd, J = 9.0, 7.8 Hz, 1H), 3.44 (d, J = 13.5 Hz, 1H), 3.77 (d, J = 15.0 Hz, 1H), 4.19 (dq, J = 10.7, 7.1 Hz, 1H), 4.29-4.50 (m, 4H), 4.90 (d, J = 15.0 Hz, 1H), 6.88-6.90 (m, 2H), 6.92 (d, J = 8.4 Hz, 1H), 7.18 (dd, J = 8.4, 2.1 Hz, 1H), 7.21-7.32 (m, 8H), 7.51 (d, J = 2.1 Hz, 1H). Selected NOEs are between δ 2.92 (C3a-*H*) and δ 3.24 (C3-HH), between δ 3.44 (C9a-*H*) and δ 2.59 (C3-*H*H), 6.88-6.90 (*o*-H of C4-*Ar*), and between δ 2.59 (C3-*H*H) and δ 6.88-6.90 (*o*-H of C4-*Ar*).

C4-*Ar*).; ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 13.96 (CH₃), 14.09 (CH₃), 36.35 (CH), 43.56 (CH), 46.14 (CH₂), 46.45 (CH), 47.18 (CH₂), 60.14 (C), 62.27 (CH₂), 62.80 (CH₂), 127.23 (CH), 127.44 (CH), 127.89 (CH), 128.63 (CH), 128.79 (CH), 130.14 (CH), 130.32 (CH), 132.61 (CH), 132.65 (C), 136.27 (C), 136.65 (C), 137.38 (C), 140.55 (C), 167.88 (C), 170.41 (C), 171.44 (C). Selected HMBC correlations are between δ 2.59 (C3-*H*H), 3.24 (C3-H*H*), 3.44 (C9a-*H*) and δ 36.35 (C3a), between δ 6.88-6.90 (*o*-H of C4-*Ar*) and δ 46.45 (C4), and between δ 3.44 (C9a-*H*), 7.51 (C8-*H*), 2.92 (C3a-*H*) and δ 60.14 (C9).; IR (KBr) 2979, 2902, 1741, 1699, 1596, 1492, 1434, 1362, 1250, 1199, 1115, 1026 cm⁻¹; MS (EI) *m*/*z* 533 (M⁺, 19), 531 (M⁺, 48), 428 (25), 191 (26), 91 (100%); HRMS (EI) *m*/*z* 531.1803, 533.1790 (calcd for C₃₁H₃₀CINO₅ 531.1813, 533.1783).

5j-cis: (1 mmol scale, toluene, 110 °C, 241 mg, 78% (5j-cis:5j-trans=31:47), 5j-cis was partially isolated by column chromatography. 84 mg, 16%); $R_f = 0.62$ (benzene-ether = 1 : 1); pale yellow oil; ¹H NMR (400 MHz, CDCl3) δ (ppm) 1.17 (t, J = 7.0 Hz, 3H), 1.37 (t, J = 7.2 Hz, 3H), 2.95 (dd, J = 9.8, 7.0 Hz, 1H), 3.16 (dd, J = 9.8, 8.3 Hz, 1H), 3.31 (dddd, J = 9.8, 8.3, 7.0, 5.1 Hz, 1H),4.09 (dq, J = 10.7, 7.0 Hz, 1H), 4.17-4.29 (m, 5H), 4.35-4.49 (m, 2H), 6.80 (dd, J = 8.3, 0.9 Hz)1H), 6.91-6.93 (m, 2H), 7.03-7.06 (m, 2H), 7.00-7.02 (m, 4H), 7.13 (dd, J = 8.3, 2.3 Hz, 1H), 7.16-7.23 (m, 3H), 7.25-7.34 (m, 3H), 8.26 (d, J = 2.3 Hz, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 13.80 (CH₃), 14.04 (CH₃), 35.88 (CH), 44.66 (CH), 46.36 (CH₂), 47.55 (CH), 48.29 (CH₂), 58.84 (C), 62.16 (CH₂), 62.66 (CH₂), 127.41 (CH), 127.44 (CH), 127.51 (CH), 127.92 (CH), 128.63 (CH), 128.74 (CH), 128.80 (CH), 129.71 (CH), 130.76 (CH), 132.69 (C), 134.14 (C), 135.65 (C), 136.81 (C), 139.12 (C), 168.08 (C), 169.02 (C), 172.28 (C); ¹H NMR (400 MHz, C_6D_6 δ (ppm) 0.779 (t, J = 7.1 Hz, 3H), 1.16 (t, J = 7.2 Hz, 3H), 2.69 (dd, J = 9.4, 8.2 Hz, 1H), 2.76 (dd, J = 9.4, 7.0 Hz, 1H), 2.92 (dddd, J = 9.6, 8.2, 7.0, 4.7 Hz, 1H), 3.70 (dq, J = 10.7, 7.1 Hz, 1H), 3.94 (d, J = 15.1 Hz, 1H), 3.99 (dq, J = 10.7, 7.1 Hz, 1H), 4.07 (d, J = 15.1 Hz, 1H), 4.28 (d, J = 4.7 Hz, 1H), 4.39 (d, J = 9.6 Hz, 1H), 4.37-4.53 (m, 2H), 6.66 (dd, J = 8.4, 0.6 Hz, 1H), 6.80-6.82 (m, 2H), 6.89 (d, J = 7.2 Hz, 2H), 6.93 (dd, J = 8.4, 2.3 Hz, 1H), 6.93-7.06 (m, 6H), 8.83 (d, J = 2.3 Hz, 1H). Selected NOEs are between $\delta 2.92$ (C3a-*H*) and $\delta 4.39$ (C9a-*H*), 4.28 (C4-*H*), and between $\delta 4.28$ (C4-*H*) and $\delta 4.39$ (C9a-*H*).; ¹³C NMR (100.6 MHz, C₆D₆) δ (ppm) 13.57 (CH₃), 13.98 (CH₃), 36.20 (CH), 45.08 (CH), 46.36 (CH₂), 47.68 (CH), 48.07 (CH₂), 59.33 (C), 62.35 (CH₂), 62.46 (CH₂), 127.32 (CH), 127.51 (CH), 127.76 (CH), 128.06 (CH), 128.76 (CH), 128.83 (CH), 129.21 (CH), 129.91 (CH), 131.54 (CH), 132.97 (C), 135.12 (C), 136.49 (C), 137.56 (C), 139.73 (C) 168.26 (C), 169.43 (C), 171.79 (C). Selected HMBC correlations are between $\delta 2.69$ (C3-*H*H), 2.76 (C3-H*H*), 4.39 (C9a-*H*), 4.28 (C4-*H*) and $\delta 36.20$ (C3a), between $\delta 2.69$ (C3-*H*H), 2.76 (C3-H*H*) and $\delta 45.08$ (C4), and between $\delta 2.92$ (C3a-*H*), 4.39 (C9a-*H*), 8.83 (C8-*H*) and $\delta 59.33$ (C9).; IR (neat) 2982, 1731, 1693, 1596, 1495, 1475, 1445, 1365, 1240, 1173, 1101, 1038 cm⁻¹; MS (EI) *m/z* 533 (M⁺, 30), 531 (M⁺, 78), 310 (28), 91 (100%); HRMS (EI) *m/z* 531.1808, 533.1808 (calcd for C₃₁H₃₀ClNO₅ 531.1813, 533.1783).

5k-trans: (1 mmol scale, toluene, 110 °C, 292 mg, 55%); $R_f = 0.5$ (hexane-AcOEt = 1 : 1); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.26 (t, J = 7.1 Hz, 3H), 1.40 (t, J = 7.1 Hz, 3H), 2.53 (dd, J = 9.7, 9.2 Hz, 1H), 3.07 (dddd, J = 13.7, 9.7, 8.0, 6.4 Hz, 1H), 3.30 (dd, J = 9.2, 8.0 Hz, 1H), 3.34 (d, J = 13.7 Hz, 1H), 3.84 (d, J = 14.9 Hz, 1H), 4.16 (dq, J = 10.6, 7.1 Hz, 1H), 4.30-4.49 (m, 3H), 4.51 (d, J = 6.4 Hz, 1H), 4.87 (d, J = 14.9 Hz, 1H), 6.91 (dd, J = 7.7, 1.1 Hz, 1H), 7.10 (d, J = 8.7 Hz, 2H), 7.22-7.35 (m, 7H), 7.57 (dd, J = 8.0, 1.2 Hz, 1H), 8.14 (d, J = 8.7 Hz, 2H). Selected NOEs are between δ 3.07 (C3a-*H*) and δ 3.30 (C3-H*H*), 4.51 (C4-*H*) and between δ 2.53 (C3-*H*H) and δ 3.34 (C9a-*H*), 7.10 (*o*-H of C4-*Ar*).; ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 13.98 (CH₃), 14.15 (CH₃), 35.99 (CH), 43.87 (CH), 46.21 (CH₂), 46.74 (CH), 47.04 (CH₂), 60.09 (C), 62.19 (CH₂), 62.75 (CH₂), 123.73 (CH), 127.57 (CH), 127.72 (CH), 127.96 (CH), 128.69 (CH), 128.80 (CH), 130.94 (CH), 131.07 (CH), 131.17 (CH), 134.69 (C), 136.46 (C), 137.31 (C), 146.96 (C), 148.66 (C), 168.10 (C), 170.82 (C), 171.26 (C). Selected HMBC correlations are between δ 2.53 (C3-*H*H), and δ 46.74 (C4), and between δ 3.07 (C3a-*H*), 3.34 (C9a-*H*), 7.10 (*o*-H of C4-*Ar*), 3.34 (C9a-*H*), 3.30 (C3-H*H*) and δ 35.99 (C3a), between δ 6.91 (C5-*H*), 7.10 (*o*-H of C4-*Ar*) and δ 46.74 (C4), and between δ 3.07 (C3a-*H*), 3.34 (C9a-*H*),

7.57 (C8-*H*) and δ 60.09 (*C*9).; IR (KBr) 2981, 1730, 1697, 1604, 1521, 1348, 1257, 1110, 1051, 1026 cm⁻¹; MS (FAB) *m/z* 565 ([M+Na]⁺), 543 ([M+H]⁺); HRMS (FAB) *m/z* 565.1956 (calcd for C₃₁H₃₀N₂O₇Na [M+Na]⁺ 565.1951), 543.2137 (calcd for C₃₁H₃₁N₂O₇ [M+H]⁺ 543.2131).

Typical experimental procedure for 8 (Table 4-4, entry 2). To a solution of monomethyl fumarate (**6a**) (299 mg, 1 mmol) and N-benzyl 3,3-diphenyl-2-propen-1-amine (**4a**) (299 mg, 1 mmol) in toluene (1.6 mL) were added, Et₃N (0.14 mL, 101 mg, 1 mmol), HOBt (270 mg, 2 mmol), and EDCI (199 mg, 1 mmol) at 0 °C. The reaction mixture was stirred for 5 min. at 0 °C, and then heated at 110 °C and stirred for 20 h. The reaction mixture was diluted with CHCl₂. The organic phase was washed with saturated aqueous NaHCO₃ solution, 2 M aqueous citric acid, saturated aqueous NaHCO₃ and water, dried (Na₂SO₄), and evaporated *in vacuo*. The residue was purified by column chromatography over silica gel eluting with hexane-Et₂O to give **8a** (194 mg, 47%).

8a: $R_f = 0.5$ (hexane-ether = 1 : 4); colorless crystals; mp 164.5-165.5 °C; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 2.60 (dddd, J = 13.7, 9.8, 6.8, 5.5 Hz, 1H), 2.74 (dd, J = 9.8, 8.7 Hz, 1H), 3.21 (dd, J = 8.7, 6.8 Hz, 1H), 3.25 (dd, J = 13.7, 11.5 Hz, 1H), 3.92 (d, J = 14.8 Hz, 1H), 3.93 (s, 3H), 4.02 (d, J = 11.5 Hz, 1H), 4.37 (d, J = 5.5 Hz, 1H), 4.74 (d, J = 14.8 Hz, 1H), 6.97-6.99 (m, 2H), 7.02 (dd, J = 7.7, 1.3 Hz, 1H), 7.15-7.31 (m, 10H), 7.41 (d, J = 7.8 Hz, 1H). Selected NOEs are between δ 2.60 (C3a-*H*) and δ 3.21 (C3-H*H*), 4.37 (C4-*H*), 4.02 (C9-*H*), and between δ 3.25 (C9a-*H*), 2.74 (C3-*H*H) and δ 6.97-6.99 (*o*-H of C4-*Ph*). Atom numbering is shown in Table 4-4.; ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 40.34 (CH), 40.41 (CH), 46.34 (CH₂), 46.69 (CH), 46.88 (CH), 47.69 (CH₂), 52.64 (CH₃), 126.97 (CH), 127.57 (CH), 127.63 (CH), 127.65 (CH), 127.79 (CH), 128.08 (CH), 128.47 (CH), 128.74 (CH), 130.36 (CH), 131.65 (CH), 133.92 (C), 136.53 (C), 138.45 (C), 140.84 (C), 173.40 (C), 173.90 (C). Selected HMBC correlations are between δ 4.02 (C9-*H*) and δ 40.34, 40.41 (C3a, C9a), between δ 2.74 (C3-*H*H), 3.21 (C3-H*H*), 2.60 (C3a-*H*), 4.02 (C9-*H*) and δ 46.69 (C4), and between δ 3.25 (C9a-*H*) and δ 46.88 (C9); IR (KBr)

3024, 1735, 1685, 1494, 1430, 1309, 1205, 1161 cm⁻¹; MS (EI) *m/z* 411 (M⁺, 47), 351 (23), 205 (22), 118 (24), 91 (100%); HRMS (EI) *m/z* 411.1845 (calcd for C₂₇H₂₅NO₃ 411.1834); Anal. Calcd for C₂₇H₂₅NO₃: C, 78.81; H, 6.12; N, 3.40. Found: C, 78.89; H, 6.22; N, 3.41.

7b: (1 mmol scale, THF, r.t., 188 mg, 89%); $R_f = 0.6$ (hexane-ether = 1 : 1); colorless oil; ¹H NMR (400 MHz, CDCl₃) (2 rotamers, ratio 1.9:1) δ (ppm) 3.99 (d, J = 6.6 Hz, 2H×0.66, major rotamer), 4.18 (d, J = 7.0 Hz, 2H×0.34, minor rotamer), 4.47 (s, 2H×0.34), 4.66 (s, 2H×0.66), 5.90 (t, J = 6.6 Hz, 2H×0.66), 6.07 (t, J = 7.0 Hz, 2H×0.34), 6.73 (dq, J = 13.6, 1.7 Hz, 1H×0.66), 6.77-6.99 (m, 1H+1H×0.34), 7.06-7.41 (m, 15H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 45.32 (CH₂), 46.54 (CH₂), 49.76 (CH₂), 51.03 (CH₂), 122.59 (C, q, $J_{FC} = 270$ Hz), 122.61 (C, q, $J_{FC} = 270$ Hz), 122.86 (CH), 122.97 (CH), 126.67 (CH), 127.41 (CH), 127.55 (CH), 127.80 (CH), 128.02 (CH), 128.09 (CH), 128.12 (CH), 128.20 (CH), 128.22 (CH), 128.38 (CH), 128.43 (CH), 128.51 (CH), 128.67 (CH), 128.75 (C), 136.65 (C), 138.19 (C), 138.75 (C), 140.98 (C), 141.39 (C), 145.68 (C), 163.55 (C), 163.66 (C); ¹⁹F NMR (376 MHz, CDCl₃) δ (ppm) -64.72 (d, $J_{FH} = 5.7$ Hz, 3F×0.34); IR (neat) 3060, 1681, 1633, 1495, 1445, 1303, 1266, 1132 cm⁻¹; MS (EI) *m/z* 421 (M⁺, 18), 330 (29), 191 (100%); HRMS (EI) *m/z* 421.1656 (calcd for C₂₆H₂₂F₃NO 421.1653).

8c: (1 mmol scale, toluene, 110 °C, 237 mg, 48%); $R_f = 0.5$ (hexane-ether = 1 : 4); pale yellow crystals; mp 84-85 °C; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 2.68-2.72 (m, 1H), 3.13-3.26 (m, 3H), 3.81 (d, J = 14.8 Hz, 1H), 4.41 (d, J = 4.1 Hz, 1H), 4.92 (d, J = 14.8 Hz, 1H), 6.83-6.85 (m, 2H), 7.10 (dd, J = 7.5, 1.9 Hz, 1H), 7.15-7.17 (m, 2H), 7.22-7.39 (m, 7H), 7.91 (d, J = 7.6 Hz, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 36.74 (CH, q, $J_{FC} = 1.5$ Hz), 40.39 (CH), 45.95 (CH₂), 46.45 (CH), 46.80 (CH₂), 56.47 (C, septet, $J_{FC} = 26$ Hz), 123.59 (C, q, $J_{FC} = 288$ Hz), 125.01 (C, q, $J_{FC} = 287$ Hz), 127.41 (CH), 127.51 (CH), 127.66 (CH), 127.89 (CH), 128.73 (CH), 128.77 (CH), 129.90 (CH), 130.05 (CH), 130.97 (CH, septet, $J_{FC} = 3.8$ Hz), 132.12 (CH), 136.34 (C), 139.51 (C), 140.85 (C), 167.96 (C); ¹⁹F NMR (376 MHz, CDCl₃) δ (ppm) -61.15 (q, $J_{FH} = 6.9$ Hz, 3F), -64.20 (q,

6.9 Hz, 3F); ¹H NMR (400 MHz, C₆D₆) δ (ppm) 2.31 (dd, J = 9.5, 8.6 Hz, 1H), 2.65 (dd, J = 8.6, 7.4 Hz, 1H), 2.75 (dddd, J = 13.9, 9.5, 7.4, 5.7 Hz, 1H), 3.12 (d, J = 13.9 Hz, 1H), 3.14 (d, J = 15.0 Hz, 1H), 3.73 (d, J = 5.7 Hz, 1H), 4.82 (d, J = 15.0 Hz, 1H), 6.64-6.66 (m, 2H), 6.72 (dd, J = 7.4, 2.0 Hz, 1H). 6.91-7.10 (m, 9H), 7.98 (d, J = 7.2 Hz, 1H). Selected NOEs are between δ 2.75 (C3a-*H*), 2.65 (C3-H*H*) and δ 3.73 (C4-*H*), between δ 2.31 (C3-*H*H) and δ 3.12 (C9a-*H*) and between δ 2.31 (C3-*H*H), 3.12 (C9a-*H*) and δ 6.64-6.66 (*o*-H of C4-*Ph*).; ¹³C NMR (100.6 MHz, C₆D₆) δ (ppm) 36.90 (CH, q, $J_{FC} = 2.3$ Hz), 40.66 (CH), 45.47 (CH2), 46.28 (CH), 46.47 (CH₂), 56.93 (C, septet, $J_{FC} = 26$ Hz), 124.88 (C, q, $J_{FC} = 289$ Hz), 125.85 (C, q, $J_{FC} = 289$ Hz), 127.33 (CH), 127.62 (CH), 127.64 (CH), 128.00 (CH), 128.76 (CH), 128.81 (CH), 129.82 (CH), 130.23 (CH), 131.27 (CH, septet, $J_{FC} = 3.8$ Hz), 132.29 (CH), 137.36 (C), 140.15 (C), 141.32 (C), 166.98 (C). Selected HMBC correlations are between δ 2.65 (C3-H*H*), 3.73 (C4-*H*) and δ 40.66 (C9a), and between δ 3.12 (C9a-*H*) and δ 56.93 (C9).; IR (KBr) 3030, 2888, 1711, 1496, 1258, 1200, 1161 cm⁻¹; MS (EI) *m/z* 489 (M⁺, 61), 91 (100%); HRMS (EI) *m/z* 489.1527 (calcd for C₂₇H₂₁F₆NO 489.1527).

8d: (1 mmol scale, THF, 60 °C, 388 mg, 80%); $R_f = 0.6$ (hexane-ether = 1 : 4); colorless crystals; mp 202-203 °C; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 2.56 (dddd, J = 13.8, 9.9, 7.0, 5.5 Hz, 1H), 2.72 (dd, J = 9.9, 9.2 Hz, 1H), 3.15 (dd, J = 13.8, 11.6 Hz, 1H), 3.21 (dd, J = 9.2, 7.0 Hz, 1H), 3.94 (s, 3H), 3.95 (d, J = 14.8 Hz, 1H), 3.96 (d, J = 11.6 Hz, 1H), 4.31 (d, J = 5.5 Hz, 1H), 4.71 (d, J = 14.8 Hz, 1H), 6.90 (d-like, J = 8.4 Hz, 2H), 6.92 (d, J = 8.4 Hz, 1H), 7.14-7.17 (m, 3H), 7.24-7.32 (m, 5H), 7.39-7.40 (m, 1H). Selected NOEs are between δ 2.56 (C3a-*H*) and δ 3.21 (C3-H*H*), 4.31 (C4-*H*), 3.96 (C9-*H*), between δ 3.15 (C9a-*H*) and δ 2.72 (C3-*H*H), and between δ 3.15 (C9a-*H*), 2.72 (C3-*H*H) and δ 6.90 (*o*-H of C4-*Ar*).; ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 39.99 (CH), 40.16 (CH), 45.63 (CH), 46.40 (CH2), 46.54 (CH), 47.43 (CH2), 52.89 (CH₃), 127.70 (CH), 127.73 (CH), 128.08 (CH), 128.27 (CH), 128.79 (CH), 131.49 (CH), 132.69 (CH), 133.22 (C), 133.51 (C), 135.63 (C), 136.30 (C), 136.56 (C), 138.70 (C), 172.76 (C), 173.17 (C).

Selected HMBC correlations are between δ 2.72 (C3-*H*H), 3.21 (C3-H*H*), 3.15 (C9a-*H*), 3.96 (C9-*H*) and δ 40.16 (C3a), between δ 3.21 (C3-H*H*), 3.15 (C9a-*H*) and δ 45.63 (C4), and between δ 3.15 (C9a-*H*) and δ 46.54 (C9).; IR (KBr) 2946, 2912, 1740, 1698, 1560, 1485, 1436, 1273, 1118, 1014 cm⁻¹; MS (EI) *m/z* 481 (M⁺, 69), 479 (M⁺, 100), 419 (27), 118 (34%); HRMS (EI) *m/z* 479.1049, 481.1028 (calcd for C₂₇H₂₃Cl₂NO₃ 479.1055, 481.1025); Anal. Calcd for C₂₇H₂₃Cl₂NO₃: C, 67.51; H, 4.83; N, 2.92. Found: C, 67.51; H, 4.84; N, 2.95.

7e: (1 mmol scale, THF, r.t., 376 mg, 77%); $R_f = 0.8$ (hexane-ether = 1 : 4); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) (2 rotamers, ratio 1.3:1) *δ* (ppm) 4.00 (d, J = 6.4 Hz, 2H×0.57, major rotamer), 4.16 (d, J = 6.8 Hz, 2H×0.43, minor rotamer), 4.49 (s, 2H×0.43), 4.66 (s, 2H×0.57), 5.87 (t, J = 6.4 Hz, 1H×0.57), 6.04 (t, J = 6.8 Hz, 1H×0.43), 6.72 (dq, $J_{HH} = 15.2$, $J_{FH} = 1.8$ Hz, 1H×0.57), 6.79-6.90 (m, 1H), 6.96-7.36 (m, 13H+1H×0.43); ¹³C NMR (100.6 MHz, CDCl₃) *δ* (ppm) 45.31 (CH₂), 46.42 (CH₂), 49.78 (CH₂), 51.28 (CH₂), 122.46 (C, q, $J_{FC} = 270$ Hz), 122.49 (C, q, $J_{FC} = 270$ Hz), 124.07 (CH), 124.17 (CH), 126.57 (CH), 127.78 (CH), 127.96 (CH, q, $J_{FC} = 6.1$ Hz), 128.05 (CH), 128.30 (CH), 128.34 (CH), 128.51 (CH), 128.55 (CH), 128.67 (CH), 128.69 (CH), 128.92 (CH), 128.96 (CH), 129.91 (CH, q, $J_{FC} = 35$ Hz), 130.10 (CH, q, $J_{FC} = 35$ Hz), 130.93 (CH), 133.66 (C), 133.76 (C), 134.11 (C), 134.41 (C), 135.67 (C), 135.99 (C), 136.46 (C), 136.59 (C), 133.76 (C), 134.11 (C), 143.14 (C), 163.36 (C), 163.57 (C); 19F NMR (376 MHz, CDCl3) *δ* (ppm) -64.67 (d, $J_{FH} = 5.7$ Hz, 3F×0.57), -64.90 (d, $J_{FH} = 4.6$ Hz, 3F×0.43); IR (neat) 3064, 1681, 1638, 1493, 1303, 1134, 1091, 1014 cm⁻¹; MS (EI) *m/z* 491 (M⁺, 13), 489 (M⁺, 20), 400 (29), 398 (41), 261 (70), 259 (100%); HRMS (EI) *m/z* 489.0880, 491.0849 (caled for C₂₆H₂₀Cl₂F₃NO 489.0874, 491.0845).

8e: (0.5 mmol scale, toluene in a closed vessel, 160 °C, 94 mg, 38%); $R_f = 0.7$ (hexane-AcOEt = 1 : 1); colorless crystals; mp 198.5-199.0 °C; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 2.56 (dddd, J = 14.2, 9.7, 6.3, 4.5 Hz, 1H), 2.62 (dd, J = 14.2, 9.7 Hz, 1H), 2.93 (dd, J = 9.7, 9.2 Hz, 1H), 3.22 (dd, J = 9.2, 6.3 Hz, 1H), 4.05 (dq, J = 9.7, JFC = 9.2 Hz, 1H), 4.05 (d, J = 14.8 Hz, 1H), 4.27 (d,

J = 4.5 Hz, 1H), 4.73 (d, J = 14.8 Hz, 1H), 6.88 (d-like, J = 8.4 Hz, 2H), 7.01 (d, J = 8.4 Hz, 1H), 7.18-7.20 (m, 2H), 7.25-7.33 (m, 6H), 7.63 (bs, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 37.47 (CH), 42.04 (CH), 43.30 (CH, q, $J_{FC} = 27$ Hz), 45.49 (CH), 46.63 (CH₂), 46.78 (CH₂), 126.59 (C, q, J_{FC} = 282 Hz), 128.12 (CH), 128.76 (CH), 128.81 (CH), 128.85 (CH), 130.02 (CH, q, $J_{FC} = 3.1$ Hz), 131.22 (CH), 132.09 (C, q, J \neg FC = 1.6 Hz), 132.42 (CH), 133.41 (C), 133.56 (C), 136.30 (C), 136.98 (C), 138.00 (C), 171.85 (C); ¹⁹F NMR (376 MHz, CDCl₃) δ (ppm) -65.57 (d, $J_{\rm FH} = 9.2$ Hz); ¹H NMR (400 MHz, C₆D₆) δ (ppm) 1.57 (dddd, J = 14.5, 10.5, 6.4, 4.9 Hz, 1H), 2.23 (dd, J = 10.5, 9.0 Hz, 1H), 2.30 (dd, J = 14.5, 10.0 Hz, 1H), 2.49 (dd, J = 9.0, 6.4 Hz, 1H), 3.30 (d, J = 4.9 Hz, 1H), 3.38 (d, J = 14.9 Hz, 1H), 3.74 (dq, J = 10.0, JFH = 8.9 Hz, 1H), 4.81 (d, *J* = 14.9 Hz, 1H), 6.34 (d, *J* = 8.3 Hz, 1H), 6.43 (d, *J* = 8.5 Hz, 2H), 6.94 (dd, *J* = 8.3, 2.1 Hz, 1H), 6.98 (d-like, J = 8.5 Hz, 2H), 7.05-7.16 (m, 5H), 7.58 (s, 1H). Selected NOEs are between δ 1.57 (C3a-H) and δ 2.49 (C3-HH), 3.30 (C4-H), 3.74 (C9-H), between δ 2.49 (C3-HH) and δ 3.30 (C4-*H*), and between δ 2.30 (C9a-*H*) and δ 6.43 (*o*-H of C4-*Ar*).; ¹³C NMR (100.6 MHz, C₆D₆) δ (ppm) 37.26 (CH), 41.52 (CH), 43.38 (CH, q, $J_{FC} = 27$ Hz), 45.23 (CH), 45.82 (CH₂), 46.50 (CH₂), 127.25 (C, q, J_{FC} = 282 Hz), 127.89 (CH), 128.33 (CH). 128.49 (CH), 128.74 (CH), 128.91 (CH), 130.49 (CH, q, J_{FC} = 3.1 Hz), 131.31 (CH). 132.38 (CH), 132.68 (C, q, J_{FC} = 1.6 Hz), 133.37 (C), 133.49 (C), 137.30 (C), 137.49 (C), 138.27 (C), 170.91 (C). Selected HMBC correlations are between δ 2.23 (C3-HH), 2.49 (C3-HH), 2.30 (C9a-H), 3.30 (C4-H) and δ 41.52 (C3a), between δ 2.23 (C3-HH), 2.30 (C9a-H), 1.57 (C3a-H) and δ 45.23 (C4), and between C 2.30 (C9a-H) and C 43.38 (C9).; IR (KBr) 2912, 1696, 1491, 1243, 1165, 1097 cm⁻¹; MS (EI) m/z 491 (M⁺, 59), 489 (M⁺, 88), 91 (100%); HRMS (EI) m/z 489.0880, 491.0862 (calcd for C₂₆H₂₀Cl₂F₃NO 489.0874, 491.0845).

8f: (1 mmol scale, benzene, 80 °C, 298 mg, 53%); $R_f = 0.3$ (hexane-ether = 1 : 4); pale yellow viscous oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 2.68 (dd, J = 9.4, 9.1 Hz, 1H), 3.04 (d, J = 13.9 Hz, 1H), 3.15 (dddd, J = 13.9, 9.4, 7.2, 5.3 Hz, 1H), 3.25 (dd, J = 9.1, 7.2, 1H), 3.84 (d, J = 14.8 Hz,

-85-

1H), 4.36 (d, J = 5.3 Hz, 1H), 4.88 (d, J = 14.8 Hz, 1H), 6.76 (d, J = 8.6 Hz, 2H), 7.03 (d, J = 8.6 Hz, 1H), 7.14-7.16 (m, 2H), 7.24-7.34 (m, 6H), 7.89 (s, 1H). Selected NOEs are between δ 3.15 (C3a-*H*) and δ 3.25 (C3-H*H*), and between δ 3.15 (C3a-*H*) and δ 4.36 (C4-*H*), between δ 2.68 (C3-*H*) and δ 3.04 (C9a-*H*), and between δ 2.68 (C3-*H*H), 3.04 (C9a-*H*) and δ 6.76 (*o*-H of C4-*Ar*).; ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 36.56 (CH, q, $J_{FC} = 2.3$ Hz), 40.08 (CH), 45.38 (CH), 45.72 (CH), 46.85 (CH₂), 56.37 (C, septet, $J_{FC} = 26$ Hz), 123.75 (C, q, $J_{FC} = 288$ Hz), 124.70 (C, q, $J_{FC} = 288$ Hz), 127.76 (CH), 127.88 (CH), 128.81 (CH), 128.99 (C), 129.05 (CH), 130.50 (CH), 130.95 (CH, septet, $J_{FC} = 3.7$ Hz), 131.17 (CH), 133.09 (CH), 133.67 (C), 133.74 (C), 136.08 (C), 137.39 (C), 138.94 (C), 167.24 (C). Selected HMBC correlations are between δ 2.68 (C3-*HH*), 3.25 (C3-*HH*), 3.04 (C9a-*H*), 4.36 (C4-*H*) and δ 36.56 (*C*3a), between δ 3.25 (C3-*HH*), 4.36 (C4-*H*) and δ 36.56 (*C*3a), between δ 3.25 (C3-*HH*), 4.36 (C4-*H*), 3.15 (C3a-*H*) and δ 40.08 (*C*9a), and between δ 3.04 (C9a-*H*) and δ 56.37 (*C*9).; ¹⁹F NMR (376 MHz, CDCl₃) δ (ppm) -61.06 (q, $J_{FF} = 6.9$ Hz), -64.31 (q, $J_{FF} = 6.9$ Hz); IR (KBr) 3031, 2888, 1711, 1596, 1491, 1259, 1203, 1163, 1015 cm⁻¹; MS (EI) *m*/*z* 559 (M⁺, 43), 557 (M⁺, 63), 91 (100%); HRMS (EI) *m*/*z* 557.0755, 559.0730 (calcd for C₂₇H₁₉Cl₂F₆NO 557.0748, 559.0718).

7a,c,d,f are unstable and decompose to give complex mixtures gradually. They are freshly prepared and used immediately in Table 4-5. For 7a, 1H and ¹³C NMR and mass spectra and for 7d, ¹H and ¹³C NMR spectra were measured. For 7c and 7f, copy of ¹H NMR are shown in Figure 4-1, 4-2.

7a: (1 mmol scale, THF, r.t., 318 mg, 77%); $R_f = 0.3$ (hexane-ether = 1 : 1); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) (2 rotamers, ratio 1.5:1) δ (ppm) 3.72 (s, 3H×0.4, minor rotamer), 3.74 (s, 3H×0.6, major rotamer), 4.01 (d, J = 6.4 Hz, 2H×0.6), 4.17 (d, J = 7.0 Hz, 2H×0.4), 4.48 (s, 2H×0.4), 4.62 (s, 2H×0.6), 5.91 (t, J = 6.4 Hz, 1H×0.6), 6.08 (t, J = 7.0 Hz, 1H×0.4), 6.89 (d, J = 15.2 Hz, 1H×0.6), 6.94 (d, J = 15.2 Hz, 1H×0.4), 6.97 (dd, J = 7.5, 1.9 Hz, 2H×0.4), 7.05-7.13 (m, 4H+2H×0.6), 7.20-7.40 (m, 9H), 7.30 (d, J = 15.2 Hz, 1H×0.6), 7.44 (d, J = 15.2 Hz, 1H×0.4); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 44.83 (CH₂), 46.37 (CH₂), 49.19 (CH₂), 50.83 (CH₂), 51.96 (CH₃), 52.02 (CH₃), 123.11 (CH), 123.16 (CH), 126.67 (CH), 127.23 (CH), 127.34 (CH), 127.39 (CH), 127.46

(CH), 127.69 (CH), 127.80 (CH), 127.83 (CH), 128.02 (CH), 128.15 (CH), 128.22 (CH), 128.34
(CH), 128.41 (CH), 128.47 (CH), 128.70 (CH), 129.56 (CH), 129.61 (CH), 131.29 (CH), 131.54
(CH), 133.86 (CH), 133.89 (CH), 135.99 (C), 136.69 (C), 138.20 (C), 138.61 (C), 140.89 (C), 141.28
(C), 145.21 (C), 145.38 (C), 164.49 (C), 164.57 (C), 165.80 (C), 165.89 (C); MS (EI) m/z 411 (M⁺, 28), 298 (18), 191 (100%); HRMS (EI) *m/z* 411.1845 (calcd for C₂₇H₂₅NO₃ 411.1834).

7c: (1 mmol scale, THF, r.t., 149 mg, 39%); $R_f = 0.6$ (hexane-ether = 1 : 1); pale yellow oil.





7d: (1 mmol scale, THF, r.t., 415 mg, 86%); $R_f = 0.7$ (hexane-ether = 1 : 4); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) (2 rotamers, ratio 1:1) δ (ppm) 3.76 (s, 3H×0.5), 3.77 (s, 3H×0.5), 4.00 (d, J = 6.4 Hz, 2H×0.5), 4.13 (d, J = 6.8 Hz, 2H×0.5), 4.51 (s, 2H×0.5), 4.63 (s, 2H×0.5), 5.87 (t, J = 6.4 Hz, 1H×0.5), 6.03 (t, J = 6.8 Hz, 1H×0.5), 6.87 (d, J = 15.2 Hz, 1H), 6.92 (d, J = 15.2 Hz,

1H), 6.95-7.05 (m, 4H+2H×0.5), 7.09-7.12 (m, 2H×0.5), 7.18-7.35 (m, 5H+2H×0.5+1H×0.5), 7.37 (d-like, J = 8.4 Hz, 2H×0.5), 7.42 (d, J = 15.2 Hz, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 45.11 (CH₂), 46.52 (CH₂), 49.54 (CH₂), 51.36 (CH₂), 52.15 (CH₃), 124.45 (CH), 126.74 (CH), 127.72 (CH), 128.00 (CH), 128.34 (CH), 128.37 (CH), 128.53 (CH), 128.58 (CH), 128.68 (CH), 128.93 (CH), 130.99 (CH), 131.03 (CH), 131.61 (CH), 131.88 (CH), 133.68 (C), 133.73 (C), 133.76 (CH), 134.09 (C), 134.27 (C), 135.90 (C), 136.22 (C), 136.65 (C), 136.69 (C), 139.05 (C), 139.51 (C), 142.85 (C), 143.14 (C), 164.64 (C), 164.82 (C), 165.89 (C), 165.96 (C).

7f: (1 mmol scale, THF, r.t., 320 mg, 57%); Rf = 0.7 (hexane-ether = 1 : 4); pale yellow oil.





Transformation of 7 to 8 (Table 4-5, entry 8): To a solution of **7d** (415 mg, 0.86 mmol, freshly prepared under the conditions of Table 4-4, entry 9) in THF (2 mL) was added Et₃N (0.12 mL, 87 mg, 0.86 mmol). The mixture was stirred at 60 °C for 20 h. The reaction mixture was

concentrated under reduced pressure. The residue was purified by column chromatography over silica gel eluting with hexane-Et₂O to give **8d** (365 mg, 88%).

References

¹ (a) E. Ciganek, Org. React. 1984, 32, 1. (b) W. R. Roush, In Comprehensive Organic Synthesis,
B. M. Trost, I. Fleming, Eds.; Pergamon, 1991, Vol. 5, 513. (c) C. O. Kappe, S. S. Murphree, A.
Padwa, Tetrahedron 1997, 53, 14179. (d) G. Brieger, J. N. Bennett. Chem. Rev. 1980, 80, 63. (e) F.
Fringuelli, A. Taticchi, The Diels-Alder Reaction - Selected Practical Methods, John Wiley & Sons: Chichester, 2002.

² (a) L. H. Klemm, T. M. McGuire, J. Heterocyclic Chem. 1972, 9, 1215. (b) L. H. Klemm, T. M. McGuire, K. W. Gopinath, J. Org. Chem. 1976, 41, 2571. (c) W. Oppolzer, R. Achini, E. Pfenninger, H. P. Weber, Helv. Chim. Acta 1976, 59, 1186. (d) S. Sun, I. J. Turchi, D. Xu, W. V. Murray, J. Org. Chem. 2000, 65, 2555. (e) R. Pedrosa, C. Andrés, J. Nieto, J. Org. Chem. 2002, 67, 782. (f) T. Ozawa, T. Kurahashi, S. Matsubara, Org. Lett. 2011, 13, 5390. (g) L. S. Kocsis, E. Benedetti, K. M. Brummond, Org. Lett. 2012, 14, 4430. (h) E. Benedetti, A. B. E. Veliz, M. Charpenay, L. S. Kocsis, K. M. Brummond, Org. Lett. 2013, 15, 2578. (i) H. J. H. J. Mun, E. Y. Seong, K.-H. Ahn, E. J. Kang, J. Org. Chem. 2018, 83, 1196. (j) D. I. Saavedra, B. D. Rencher, D.-H. Kwon, S. J. Smith, D. H. Ess, M. B. Andrus, J. Org. Chem. 2018, 83, 2018. (k) M. T. Cox, J. Chem. Soc., Chem. Commun. 1975, 903.

³ S. Yamazaki, H. Sugiura, S. Ohashi, K. Ishizuka, R. Saimu, Y. Mikata, A. Ogawa, J. Org. Chem.
2016, 81, 10863.

⁴ T. Wagner-Jauregg, *Synthesis* **1980**, 769.

⁵ (a) M. J. Batchelor, J. M. Mellor, *Tetrahedron Lett.* 1985, 26, 5109. (b) J. Becher, H. C. Nielsen, J. P. Jacobsen, O. Simonsen, H. Clausen, *J. Org. Chem.* 1988, 53, 1862. (c) T.-C. Wu, K. N. Houk, *Tetrahedron Lett.* 1985, 26, 2293. (d) W. V. Murray, S. Sun, I. J. Turchi, F. K. Brown, A. D.

Gauthier, J. Org. Chem. 1999, 64, 5930. (e) T. N. Cayzer, M. N. Paddon-Row, D. Moran, A. D.
Payne, M. S. Sherburn, P. Turner, J. Org. Chem. 2005, 70, 5561. (f) A. Benallou, H. E. A. E.
Abdallaoui, H. Garmes, I. J. I. A. S. 2014, 8, 685.

- ⁶ G. A. Pinna, G. Cignarella, G. Loriga, G. Murineddu, J.-M. Mussinu, S. Ruiu, P. Faddad, W. Fratta, *Bioorg. Med. Chem.* **2002**, 10, 1929.
- ⁷ H. Sugiura, S. Yamazaki, K. Go, A. Ogawa *Eur. J. Org. Chem.* doi/abs/10.1002/ejoc.201801508
 ⁸ (a) M. Zheng, F. Wu, K. Chen, S. Zhu, *Org. Lett.* 2016, 18, 3554. (b) A. Diment, E. Ritchie, W. C. Taylor, *Aust. J. Chem.* 1969, 22, 1721. (c) B. R. Stranix, G. D. Darling, *J. Org. Chem.* 1997, 62, 9001. (d) M. C. Carreño, J. Mahugo, A. Urbano, *Tetrahedron Lett.* 1997, 38, 3047.
- ⁹ R. B. Woodward, R. Hoffmann, *The Conservation of Orbital Symmetry*; Verlag Chemie: New York, **1970**.
- ¹⁰ A. Sugimoto, J. Yamano, M. Yasueda, S. Yoneda, *J. Chem. Soc., Perkin Trans. 1* 1988, 2579.
 ¹¹ C. Hansch, A. Leo, R. W. Taft, *Chem. Rev.* 1991, 91, 165.
- ¹² S. Yamazaki, K. Ohmitsu, K. Ohi, T. Otsubo, K. Moriyama, Org. Lett. 2005, 7, 759.
- ¹³ (a) J. Rehbein, S. Leick, M. Hiersemann, J. Org. Chem. 2009, 74, 1531. (b) S. Balduzzi, M. A.
- Brook, M. J. McGlinchey, Organometallics 2005, 24, 2617. (c) S. E. Denmark, M. A. Harmata, K.
- S. White, J. Org. Chem. 1987, 52, 4031. (d) N. A. LeBel, N. Balasubramanian, J. Am. Chem. Soc. 1989, 111, 3363.
- ¹⁴ S. G. Jarboe, P. Beak, Org. Lett. 2000, 2, 357.
- ¹⁵ H.-P. Wu, R. Aumann, R. Fröhlich, E. Wegelius, Organometallics 2001, 20, 2183.
- ¹⁶ J. Limberger, T. S. Claudino, A. L. Monteiro, *RSC Adv.* **2014**, 4, 45558.
- ¹⁷ T. Y. Cowie, M. Veguillas, R. L. Rae, M. Rougé, J. M. Żurek, A. W. Prentice, M. J. Paterson,
- M. W. P. Bebbington, J. Org. Chem. 2017, 82, 6656.

Chapter 5

Sequential Intramolecular Diels-Alder Reaction of 3-Heteroaryl-2-propenylamides of Ethenetricarboxylate

5-1 Introduction

The development of the synthetic strategies of linearly fused heterocyclic ring systems has attracted significant attention because of their widespread occurrence in biologically active compounds. ¹ The intramolecular Diels-Alder (IMDA) reaction of vinylfurans and the heteroaromatic analogues such as vinyl pyrroles, thiophenes,^{2,3} and imidazoles,⁴ has been used to construct the linearly fused tricyclic heterocycles.⁵

In chapters 3 and 4, reactions of highly electron-deficient alkenic carboxylic acid, 1,1-diethyl 2-hydrogen ethenetricarboxylate 1, with *E*- or *Z*-cinnamylamines and 3,3-diary-2-propenylamines under the amide formation conditions gave tricyclic compounds in sequential amide formation/IMDA reaction/rearomatization process (Scheme 5-1). Reaction of 1 with *E*-cinnamylamines bearing electron-withdrawing groups under the amide formation conditions gave *trans*-fused tetrahydrobenz[*f*]isoindolines. The reaction of 1 with *Z*-cinnamylamines on heating at 80-110 °C gave [4 + 2] cycloadducts, *cis*-fused tricyclic compounds as the major products. Reaction of 1 with 3,3-diaryl-2-propen-1-amines gave *cis*- and *trans*-fused tricyclic compounds selectively, depending on the substituents on the benzene ring, reaction temperature, and solvent. The reaction mechanism of the formation of *cis*- and *trans*-fused rings was discussed.

In order to extend the stereoselective construction of *cis*- and *trans*-fused linearly fused tricyclic compounds by IMDA reaction of styrenes (vinyl benzene) to the reaction of vinyl heteroarenes, the reaction of **1** with 3-heteroaryl-2-propenylamines has been studied in this work. Substituted regioisomers of furan and thiophene as examples of five-membered and electron-rich

heteroarenes and pyridine as an example of six-membered and electron-poor heteroarenes have been examined.⁶ Lower aromaticity of the heterocycles than benzene may effect the reaction.⁷ It is also desirable to develop new efficient reactions using furans as renewable resources.⁸ The origin of stereoselectivity of the fused rings has been examined by the DFT calculations.



Scheme 5-1. The reaction of 3-aryl-2-propenylamines and ethenetricarboxylate.

5-2 Results and Discussion

Reactions of 1,1-diethyl 2-hydrogen ethenetricarboxylate 1 and furan substrates in the presence of EDCI/HOBt/Et₃N were examined. Reaction of 1 and *E*-3-(2-furyl)-2-propenylamines **2a-b** in the presence of EDCI/HOBt/Et₃N in THF at room temperature gave a complex mixture and the possible [2 + 2] cycloadducts were not detected. The reaction of 1 and **2a-b** in benzene, DME, DMF and toluene at 80-110 °C gave [4 + 2] cycloadducts, furo[2,3-f] isoindoles **3a-b** as the major products (Table 5-1). The products **3a-b** are unstable and decomposed gradually at room temperature. Treatment of crude **3a-b** with 1 M HCl in ether gave the furan-reproduced products **4a-b** (Table 5-2). The *cis*-fused stereochemistry of **4a-b** was determined by observed NOEs between C7a-*H* and C4a-*H*. The reaction of **2a** in ClCH₂CH₂Cl at 80-110 °C gave **4a** directly,

probably because of generation of HCl *in situ* under the reaction conditions.⁹ The stereochemistry of unstable **3a-b** was deduced from their crude NMR and the proposed mechanism shown below.



Table 5-1. Reactions of 1,1-diethyl ethenetricarboxylate 1 and (*E*)-3-(2-furyl)-2-propenylamines 2.

^a A complex mixture. ^b The reaction in DMF at room temperature gave a complex mixture.

Table 5-2. Transformation of 3 to 4.

$\begin{array}{c c} & EtO_2C & CO_2Et \\ & H & H \\ & H & O \\ & H & O \\ & & 3a-b \end{array} \qquad \begin{array}{c} 1M \text{ HCl/ether} & Ph & O \\ & H & O \\ & & H & O \\ & & H & O \\ & & & H \\ & & & & & \\ & & & & & \\ & & & &$									
Entry	3	Х	HC1	Solvent	Temp.	Product	Yield (%)		
1	3 a	Н	1 eq	ClCH ₂ CH ₂ Cl	80 °C	4a	78		
2	3b	Br	0.1 eq	ClCH ₂ CH ₂ Cl	80 °C	4b	74		
3	3b	Br	0.1 eq	Benzene	80 °C	4b	65		

The reaction of **1** and *E*-3-(2-furyl)-2-propenylamine **5a** in THF at room temperature gave [4 + 2] cycloadduct, furo[2, 3-f]isoindole **6a** as the major products (Table 5-3). The product **6a** is unstable and decomposed gradually at room temperature. The reaction of **1** and *E*-3-(2-furyl)-2-propenylamines **5a-b** in ClCH₂CH₂Cl, DME, PhCF₃, benzene, and toluene at 80-110 °C gave *trans*-fused furan-reproduced products **7a-b** directly as the major products. The *trans*-fused stereochemistry of **7a-b** was determined by observed NOEs between C4a-*H* and C5-H^{α}H^{β}, between C5-H^{α}H^{β} and C4-H^{α}H^{β}, and between C4-H^{α}H^{β} and C7a-*H*. The stereochemistry of unstable **6a** was deduced from the crude NMR and the proposed mechanism shown below.

Although the initial cycloadducts **3** and **6** are unstable, they were isolated as crude forms. On the other hand, the corresponding initial cycloadducts for styrenes were not detected under the reaction conditions.¹⁰ This is probably because the aromaticity strength of heteroarenes is comparatively lower than that of benzene. A few initial adducts of IMDA reactions of vinyl heteroarenes were isolated and characterized.^{3, 4}

EtO ₂ C HO ₂ C	∠CO₂Et +	HN- Ŕ	5a : X = CH ₂ Ph 5b : X = CH ₂ Cyc	Et ₃ N HOBt EDCI clohexyl	tO_2C CO_3 H_{a} H_{a}	2Et D^{1} 2^{2} 3 $$	EtO ₂ C O 7 H 8 6 7 H 8 6 7 H 8 7a 7a 4a $H^{\beta} H^{\alpha} H^{\beta}$ 7a 7a 7a 4a 7a 7a 7a 7a 7a 7a 7a 7	$\begin{array}{c} CO_2Et\\ \hline 8a \\ 3a \\ 3a \\ 3a \\ 4 \end{array}$
	Entry	5	R	Solvent	Temp.	product	Yield (%)	
	1	5 a	CH ₂ Ph	THF	r.t.	6a	ca. 59 ^a	
	2	5a	CH ₂ Ph	CH ₂ ClCH ₂ Cl	80 °C	7a	78	
	3	5a	CH ₂ Ph	DME	80 °C	7a	52	
	4	5a	CH ₂ Ph	PhCF ₃	80 °C	7a	65	
	5	5a	CH ₂ Ph	Benzene	80 °C	7a	70	
	6	5a	CH ₂ Ph	Toluene	110 °C	7a	72	
	7	5b	CH ₂ Cyclohexyl	Benzene	80 °C	7b	48	
	8	5b	CH ₂ Cyclohexyl	Toluene	110 °C	7b	39	

Table 5-3. Reactions of 1,1-diethyl ethenetricarboxylate 1 and (*E*)-3-(3-furyl)-2-propenylamines 5.

^a Product **6a** is unstable and decomposes to give complex mixtures gradually.

Next, the reaction of **1** and other 3-heteroaryl-2-propenylamines **8**, **9**, and **10** in the presence of the amide condensation reagents was examined (Scheme 5-2).



Scheme 5-2 3-Heteroaryl-2-propenylamines 8, 9, and 10.

The reactions of 3-(2-thiophenyl)-2-propen-1-amines **8a,b** and 3-(-3-thiophenyl) -2-propen-1-amine **9** only gave complex mixtures under the various reaction conditions, probably

because the initial [4 + 2] cycloadducts, further intermediates or thiophene reproduced products for thiophene derivatives are unstable under the reaction conditions.

Among them, the reaction of **1** and **8b** at room temperature gave ketone derivative **11** as an isolable product in 45% yield (Scheme 5-3). The similar C8-oxidized products may be formed for other substrates such as **2**, **5**, and **8a**, probably by oxidation with adventitious oxygen in situ. Addition of a radical scavenger such as TEMPO or the reaction under air was also attempted. However, the reproduced results and isolation of the oxidized products for the other substrates could not be achieved. The stereochemistry of **11** was determined as *cis*-fused by observed NOEs (in C₆D₆) between C7a-*H* and C4a-*H*, similar to products **4a-b** from 3-(2-furyl)-2propen-1-amines **2a-b**.



Scheme 5-3. The reaction of 1 and 8b.

Reaction of **1** and 3-(3-pyridinyl)-2-propen-1-amine **10a** with EDCI/HOBt/Et₃N at room temperature, 60 °C and 80 °C gave HOBt-incorporated 3,4-*trans*-pyrrolidine **12** as a single diastereomer in 61% yield selectively (Scheme 5-4). On the other hand, the reaction of 3-(4-pyridinyl)-2-propen-1-amine **10b** only gave complex mixtures

Since the nitrogen is more electronegative than carbon, the pyridine ring is electron-deficient.¹¹ The reaction of **10a** proceeded similarly to the styrene derivative with electron-withdrawing *m*-nitro group. The stereochemistry of **12** was deduced as shown in Scheme 5-4, similarly to the proposed mechanism for formation of HOBt adduct from *m*-nitrocinnamylamide.^{10a} The O-C bond formation and C-C bond formation from the

intermediate amide occurred concertedly to lead to a cyclized product **12**. Intermolecular HOBt nucleophilic attack from outside may lead to 3,4-*trans* cyclized product **12** by steric reason.



Scheme 5-4. The reaction of 1 and 10a.

Understanding the detailed mechanism of the cycloadditions is important to find the factor to control the selectivity. In order to explain the observed *cis*- and *trans*-fused selectivity, and examine the applicability of the reaction models¹⁰ on the selectivity to heteroaryl ring systems, the reaction mechanism was investigated using B3LYP/6-31G*,^{12,13} ω B97X-D¹⁴ and M06-2X¹⁵ calculations including the PCM¹⁶ solvent effect (solvent=THF). TS geometry was characterized by vibrational analysis, which checked whether the obtained geometry has single imaginary frequencies (v[‡]). From TSs, reaction paths were traced by the intrinsic reaction coordinate (IRC) method¹⁷ to obtain the energy-minimum geometries. ΔE (sum of electronic and zero-point energies) were refined by single-point calculations of RB3LYP/6-311+G(d,p) SCRF = (PCM, solvent = THF), RwB97X-D and RM06-2X/6-311+G(d,p) SCRF = (PCM, solvent = THF), respectively.

The *cis* and *trans*-fused stereoselectivity for reaction of 3-(2- and 3-furyl)-2-propenylamides in the [4 + 2] cycloaddition path has been examined. For 2-furan, the stepwise path to *trans*-fused [4 + 2] cycloadduct and the concerted path to *cis*-fused [4 + 2] cycloadduct were calculated by B3LYP functional (Scheme 5-5). On the other hand, the concerted paths to both *trans*- and *cis*-fused [4 + 2] cycloadducts were obtained by ω B97X-D and M06-2X. Although the calculation method dependency is seen, the activation energies TS2 for the paths leading to *trans* cycloadduct **3M**-*trans* are lower than those of TS3 for the paths leading to *cis*-[4 + 2] cycloadduct **3M**-*cis*, but *trans*-[4 + 2] cycloadduct **3M**-*trans* is less stable than *cis*-[4 + 2] cycloadduct **3M**-*cis* in each calculation methods. The stability of **3M**-*cis* may be partially attributed to **3a**,**4a**-*cis* (1,3-diequatorial-like) conformation of the cyclohexene ring. At higher temperature, the reverse reaction may occur and the reaction leads to the more stable [4 + 2] cycloadduct **3M**-*cis*.

For 3-furan, both concerted paths lead to *cis* and *trans* adducts by B3LYP, ω B97X-D and M06-2X functionals (Scheme 5-6). Similarly, the activation energy TS4 for the path leading to *trans* cycloadduct **6M**-*trans* are lower than TS5 for the path leading to *cis*-[4 + 2] cycloadduct **6M**-*cis*, but *trans*-[4 + 2] cycloadduct **6M**-*trans* is less stable than *cis*-[4 + 2] cycloadduct **6M**-*cis*.

The relative stability of **3M**-*trans* to the precursor **AM1** ($\Delta E^{\circ} = +1.59$, [-14.16], -13.80 kcal/mol) is lower than that of **6M**-*trans* to **CM1** ($\Delta E^{\circ} = -4.31$, [-17.77], -17.87 kcal/mol) within each calculation method. The bond length C4-C3a in **3M**-*trans* is longer than C8-C8a in **6M**-*trans* within each calculation method. The longer C4-C3a bond possibly arises from the larger zwitter-ionic character of **3M**-*trans* compared to **6M**-*trans* (Scheme 5-7). 2-Furylmethyl cation is 9.4 kcal/mol¹⁸ more stable than 3-furylmethyl cation, possibly because of more effective delocalization of positive charge by contributions from resonance structures. Therefore, the reverse reaction of **6M**-*trans* (\rightarrow **CM1**) may be less facile than that of **3M**-*trans* may give the final stable furan-reproduced product **7M**-*trans* by the stepwise protonation-deprotonation (1,3-H⁺ shift) under the reaction conditions.



Scheme 5-5. [4 + 2] Cycloaddition path for (2-furyl)-2-propenylamide. ΔE 's (sum of electronic and zero-point energies) by RB3LYP/6-311+G(d,p) SCRF = (PCM, solvent = THF) // RB3LYP/6-31G* SCRF = (PCM, solvent = THF), RwB97XD/6-311+G(d,p) SCRF = (PCM, solvent = THF) // RwB97XD /6-31G* SCRF = (PCM, solvent = THF) in square brackets [], and RM062X/6-311+G(d,p) SCRF = (PCM, solvent = THF) // RM062X/6-31G* SCRF = (PCM, solvent = THF) with underline, relative to AM1 are shown.



Scheme 5-6. [4 + 2] Cycloaddition path for (3-furyl)-2-propenylamide. ΔE 's (sum of electronic and zero-point energies) by RB3LYP/6-311+G(d,p) SCRF = (PCM, solvent = THF) // RB3LYP/6-31G* SCRF = (PCM, solvent = THF), RwB97XD/6-311+G(d,p) SCRF = (PCM, solvent = THF) // RwB97XD /6-31G* SCRF = (PCM, solvent = THF) in square brackets [], and RM062X/6-311+G(d,p) SCRF = (PCM, solvent = THF) // RM062X/6-31G* SCRF = (PCM, solvent = THF) with underline, relative to CM1 are shown.



Scheme 5-7. Bond lengths C4-C3a (**3M***-trans*) and C8-C8a (**6M***-trans*) and resonance structures of 2- and 3-furylmthyl cations.

5-3 Conclusion

In summary, sequential reaction of 1,1-diethyl 2-hydrogen ethenetricarboxylate with 3-heteroarylpropenylamines has been examined. Reaction of 1,1-diethyl 2-hydrogen ethenetricarboxylate with *E*-3-(2-furyl)-2-propenylamines under the amide formation conditions gave *cis*-fused tricyclic compounds on heating, via the amide formation, [4 + 2] cycloaddition, and H-shift reactions. On the other hand, the reaction with *E*-3-(3-furyl)-2-propenylamines gave *trans*-fused tricyclic compounds. The origin of observed stereoselectivity of the fused rings has been examined by the DFT calculations. The difference between 2-furyl and 3-furyl derivatives in the stereoselectivity of the products may arise from the delocalization of positive charge of 2-furylmethyl and 3-furylmethyl cations. Reaction of 1,1-diethyl 2-hydrogen ethenetricarboxylate with 3-(3-pyridinyl)-2-propen-1-amine under the amide formation conditions gave HOBt-incorporated 3,4-*trans*-pyrrolidine selectively. The scope and limitation under the reaction

conditions have been described. The results are useful for the study on the effects of various vinyl heteroarenes in the sequential reactions.

5-4 Experimental Section

General Procedures

¹H Chemical shifts are reported in ppm relative to Me₄Si. ¹³C Chemical shifts are reported in ppm relative to CDCl₃ (77.1 ppm). ¹³C mutiplicities were determined by DEPT and HSQC. Mass spectra were recorded at an ionizing voltage of 70 eV by EI. Mass analyzer type used for EI is double-focusing. All reactions were carried out under a nitrogen atmosphere. Column chromatography was performed on silica gel (75-150 μm).

Ethenetricarboxylate **1** was prepared according to the literature.¹⁹ *E*-3-heteroaryl-2- propenylamines **2a-b**, **5a-b**, were prepared from the corresponding *E*-3-heteroaryl-2-propenals and amines by reductive amination in methanol according to the literature procedure.²⁰ *E*-3-(2-furyl)-2-propenal, ²¹ *E*-3-(5-bromo-2-furyl)-2-propenal,²¹ *E*-3-(3-furyl)-2-propenal, ²² *E*-3-(3-thienyl)-2-propenal, ²³ *E*-3-(5-bromo-2-thienyl)-2-propenal, ²⁴ *E*-3-(4-pyridinyl)-2-propenal, ²⁵ and *E*-3-(3-pyridinyl)-2-propenal^{25, 26} were prepared by the reaction of the corresponding heteroarylaldehydes and formylmethylenetriphenylphosphorane according to the literature procedure. *E*-3-(2-thienyl)-2-propenal was prepared according to the literature.²⁷

2a: (8.9 mmol scale, 1.43 g, 75%); $R_f = 0.4$ (hexane-ether = 1 : 1); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.81 (bs, 1H), 3.38 (dd, J = 6.2, 1.3 Hz, 2H), 3.81 (s, 2H), 6.19 (d, J = 3.3 Hz, 1H), 6.23 (dt, J = 15.8, 6.2 Hz, 1H), 6.33-6.39 (m, 2H), 7.22-7.27 (m, 1H), 7.29-7.33 (m, 5H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 50.74 (CH₂), 53.19 (CH₂), 107.22 (CH), 111.20 (CH), 119.90 (CH), 127.03 (CH), 127.14 (CH), 128.22 (CH), 128.46 (CH), 140.12 (C), 141.76 (CH), 152.73 (C); IR (neat) 3311, 3027, 2824, 1494, 1453, 1361, 1254, 1151, 1012 cm⁻¹; MS (EI) m/z 213 (M⁺, 44), 122 (54), 91 (100%); HRMS (EI) m/z 213.1137 (calcd for C₁₄H₁₅NO 213.1154).

2b: (4.0 mmol scale, 640 mg, 54%); $R_f = 0.3$ (hexane-ether = 1 : 1); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.57 (bs, 1H), 3.38 (d, J = 4.7 Hz, 1H), 3.81 (s, 2H), 6.13 (d, J = 3.3 Hz, 1H), 6.20-6.30 (m, 3H), 7.22-7.28 (m, 1H), 7.30-7.33 (m, 4H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 50.58 (CH₂), 53.26 (CH₂), 109.41 (CH), 112.94 (CH), 118.81 (CH), 121.14 (C), 127.06 (CH), 128.12 (CH), 128.19 (CH), 128.44 (CH), 128.48 (C), 140.22 (C), 154.79 (C); IR (neat) 3313, 3027, 2816, 1488, 1453, 1128, 1012 cm⁻¹; MS (EI) m/z 293 (M⁺, 2.7), 291 (M⁺, 2.7), 212 (47), 91 (100%); HRMS (EI) m/z 291.0263, 293.0241 (calcd for C₁₄H₁₄BrNO 291.0259, 293.0238).

5a: (1.6 mmol scale, 222 mg, 65%); $R_f = 0.3$ (hexane-ether = 1 : 4); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.52 (bs, 1H), 3.37 (dd, J = 6.4, 1.4 Hz, 2H), 3.81 (s, 2H), 6.03 (dt, J = 15.8, 6.4 Hz, 1H), 6.39 (bd, J = 15.8 Hz, 1H), 6.52 (dd, J = 1.2, 0.6 Hz, 1H), 7.23-7.28 (m, 1H), 7.30-7.36 (m, 5H), 7.38 (bs, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 51.18 (CH₂), 53.39 (CH₂), 107.64 (CH), 121.17 (CH), 124.04 (C), 127.03 (CH), 128.13 (CH), 128.22 (CH), 128.47 (CH), 140.11 (CH), 140.30 (C), 143.50 (CH); IR (neat) 3327, 3026, 2815, 1495, 1453, 1159, 1072, 1024 cm⁻¹; MS (EI) m/z 213 (M⁺, 32), 184 (17), 122 (25), 91 (100%); HRMS (EI) m/z 213.1172 (calcd for C₁₄H₁₅NO 213.1154).

5b: (2.3 mmol scale, 230 mg, 43%); $R_f = 0.3$ (hexane-ether = 1 : 4); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 0.865-0.965 (m, 2H), 1.10-1.30 (m, 3H), 1.41-1.52 (m, 1H), 1.65-1.77 (m, 5H), 2.46 (d, J = 6.6 Hz, 2H), 3.33 (dd, J = 6.4, 1.4 Hz, 2H), 6.02 (dt, J = 15.8, 6.4 Hz, 1H), 6.37 (d, J = 15.8 Hz, 1H), 6.52 (d, J = 1.6 Hz, 1H), 7.34 (d, J = 1.6 Hz, 1H), 7.38 (bs, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 26.11 (CH₂), 26.72 (CH₂), 31.52 (CH₂), 38.13 (CH), 52.09 (CH₂), 56.34 (CH₂), 107.67 (CH), 120.80 (CH), 124.10 (C), 128.52 (CH), 140.01 (CH), 143.43 (CH); IR (neat) 3329, 2922, 1508, 1448, 1161, 1025 cm⁻¹; MS (EI) m/z 219 (M⁺, 24), 136 (40), 107 (100%); HRMS (EI) m/z 219.1621 (calcd for C₁₄H₂₁NO 219.1623).

8a: (2.25 mmol scale, 396 mg, 77%); $R_f = 0.2$ (hexane-ether = 1 : 1); colorless oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.50 (bs, 1H), 3.39 (dd, J = 6.3, 1.5 Hz, 2H), 3.82 (s, 2H), 6.15 (dt, J = 15.6, 6.3 Hz, 1H), 6.67 (d, J = 15.6 Hz, 1H), 6.91 (d, J = 3.6 Hz, 1H), 6.94 (dd, J = 5.0, 3.6 Hz,

1H), 7.12 (d, J = 5.0 Hz, 1H), 7.23-7.29 (m, 1H), 7.31-7.35 (m, 4H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 50.96 (CH₂), 53.37 (CH₂), 123.94 (CH), 124.59 (CH), 125.20 (CH), 127.06 (CH), 127.34 (CH), 128.24 (CH), 128.32 (CH), 128.50 (CH), 140.25 (C), 142.37 (C); IR (neat) 3308, 3026, 2818, 1495, 1453, 1203, 1117 cm⁻¹; MS (EI) m/z 229 (M⁺, 100), 138 (44), 132 (47%); HRMS (EI) m/z 229.0922 (calcd for C₁₄H₁₅NS 229.0925).

8b: (5.2 mmol scale, 1.29 g, 72%); $R_f = 0.2$ (hexane-EtOAc = 1 : 1); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.56 (bs, 1H), 3.35 (dd, J = 6.3, 1.6 Hz, 2H), 3.80 (s, 2H), 6.04 (dt, J = 15.6, 6.3 Hz, 1H), 6.54 (dtd, J = 15.6, 1.6, 0.5 Hz, 1H), 6.63 (d, J = 3.8 Hz, 1H), 6.87 (d, J = 3.8 Hz, 1H), 7.23-7.29 (m, 1H), 7.30-7.35 (m, 4H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 50.77 (CH₂), 53.34 (CH₂), 110.59 (C), 124.04 (CH), 125.40 (CH), 127.08 (CH), 128.20 (CH), 128.49 (CH), 128.81 (CH), 130.17 (CH), 140.03 (C), 143.95 (C); IR (neat) 3311, 3026, 2819, 1643, 1494, 1453, 1436, 1361, 1197, 1118, 1051, 1028 cm⁻¹; MS (EI) m/z 309 (M⁺, 20), 307 (M⁺, 20), 228 (62), 132 (44), 91 (100%); HRMS (EI) m/z 307.0025, 309.0012 (calcd for C₁₄H₁₄BrNS 307.0030, 309.0010).

9: (0.44 mmol scale, 63 mg, 66%); $R_f = 0.2$ (hexane-EtOAc = 1 : 4); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.56 (bs, 1H), 3.39 (dd, J = 6.4, 1.4 Hz, 2H), 3.81 (s, 2H), 6.16 (dt, J = 15.8, 6.4 Hz, 1H), 6.54 (d, J = 15.8 Hz, 1H), 7.09 (bs, 1H), 7.19 (dd, J = 5.1, 1.4 Hz, 1H), 7.21-7.29 (m, 2H), 7.30-7.35 (m, 4H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 51.15 (CH₂), 53.33 (CH₂), 121.59 (CH), 125.02 (CH), 125.69 (CH), 125.99 (CH), 127.01 (CH), 128.21 (CH), 128.32 (CH), 128.46 (CH), 139.77 (C), 140.22 (C); IR (neat) 3313, 3025, 2817, 1494, 1453, 1118 cm⁻¹; MS (EI) m/z 229 (M⁺, 36), 132 (56), 91 (100%); HRMS (EI) m/z 229.0925 (calcd for C₁₄H₁₅NS 229.0925).

10a: (2.9 mmol scale, 193 mg, 29%); $R_f = 0.2$ (ether-MeOH = 1 : 1); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.89 (bs, 1H), 3.46 (dd, J = 6.1, 1.0 Hz, 2H), 3.84 (s, 2H), 6.38 (dt, J = 16.0, 6.1 Hz, 1H), 6.53 (d, J = 16.0 Hz, 1H), 7.21 (dd, J = 7.9, 4.8 Hz, 1H), 7.24-7.29 (m, 1H), 7.31-7.36 (m, 4H), 7.67 (ddd, J = 7.9, 2.0, 1.4 Hz, 1H), 8.44 (dd, J = 4.7, 1.4 Hz, 1H), 8.57 (d, J = 16.0 Hz, 1H), 7.97 (ddd, J = 7.9, 2.0, 1.4 Hz, 1H), 8.44 (dd, J = 4.7, 1.4 Hz, 1H), 8.57 (d, J = 16.0 Hz, 1H), 8.44 (dd, J = 4.7, 1.4 Hz, 1H), 8.57 (d, J = 16.0 Hz, 1H), 8.57 (d, J = 16.0 Hz, 1H), 8.44 (dd, J = 4.7, 1.4 Hz, 1H), 8.57 (d, J = 16.0 Hz,
2.0 Hz, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 51.05 (CH₂), 53.42 (CH₂), 123.43 (CH), 127.09 (CH), 127.68 (CH), 128.19 (CH), 128.48 (CH), 130.94 (CH), 132.69 (CH), 132.72 (C), 140.02 (C), 148.22 (CH), 148.42 (CH); IR (neat) 3296, 3027, 2822, 1651, 1568, 1453, 1415, 1122, 1025 cm⁻¹; MS (EI) m/z 224 (M⁺, 5.7), 132 (20), 118 (23), 91 (100%); HRMS (EI) m/z 224.1319 (calcd for C₁₅H₁₆N₂ 224.1313).

10b: (1.6 mmol scale, 112 mg, 31%); $R_f = 0.4$ (ether-MeOH = 1 : 1); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.76 (bs, 1H), 3.47 (d, J = 4.7 Hz, 2H), 3.84 (s, 2H), 6.51 (d, J = 15.9 Hz, 1H), 6.52 (dt, J = 15.9, 4.7 Hz, 1H), 7.21-7.22 (m, 2H), 7.24-7.35 (m, 5H), 8.51-8.52 (m, 2H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 50.85 (CH₂), 53.51 (CH₂), 120.85 (CH), 127.15 (CH), 128.20 (CH), 128.53 (CH), 128.82 (CH), 133.73 (CH), 140.02 (C), 144.52 (C), 150.13 (CH); IR (neat) 3295, 3026, 2923, 1650, 1597, 1550, 1494, 1453, 1415 cm⁻¹; MS (EI) m/z 224 (M⁺, 17), 132 (25), 91 (100%); HRMS (EI) m/z 224.1315 (calcd for C₁₅H₁₆N₂ 224.1313).

Typical experimental procedure for preparation of 3, 4, 6, 7, 11, 12 (Table 5-1, entry 3). To a solution of 1,1-diethyl 2-hydrogen ethenetricarboxylate (1) (prepared from 1,1-diethyl 2-*tert*-butyl ethenetricarboxylate (272 mg, 1 mmol) upon treatment with CF_3CO_2H (4 mL)) in benzene (0.7 mL) were added **2a** (213 mg, 1 mmol) in benzene (0.7 mL), Et₃N (0.14 mL, 102 mg, 1 mmol), HOBt (1-hydroxybenzotriazole) (270 mg, 2 mmol), and EDCI (1-[3-(dimethyl-amino)propyl]- 3-ethylcarbodiimide hydrochloride) (199 mg, 1.04 mmol) at 0 °C. The reaction mixture was stirred for 1 h at 0 °C, and was allowed to warm to 80 °C and stirred for 20 h. The reaction mixture was diluted with CH_2Cl_2 . The organic phase was washed with saturated aqueous NaHCO₃ solution, 2 M aqueous citric acid, saturated aqueous NaHCO₃ and water, dried (Na₂SO₄), and evaporated *in vacuo*. The residue was purified by column chromatography over silica gel eluting with hexane-Et₂O to give **3a** (252 mg, ca. 61%).

3a,b and **6** are unstable and decompose to give complex mixtures gradually. **3a,b** are freshly prepared and used immediately in Table 5-2.

3a: $R_f = 0.5$ (hexane-ether = 1 : 8); colorless oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.18 (t, J = 7.1 Hz, 3H), 1.35 (t, J = 7.1 Hz, 3H), 2.98-3.03 (m, 1H), 3.22-3.24 (m, 2H), 3.37-3.42 (m, 1H), 4.01 (d, J = 14.7 Hz, 1H), 4.01-4.21 (m, 3H), 4.28-4.53 (m, 2H), 4.92 (d, J = 14.7 Hz, 1H), 4.97-4.98 (m, 1H), 5.32-5.33 (m, 1H), 6.45 (dd, J = 2.9, 2.3 Hz, 1H), 7.25-7.35 (m, 5H). Selected NOEs are between δ 3.22-3.24 (C5a-*H*,8a-*H* (overlapped)) and δ 3.37-3.42 (C8-H*H*), 4.01-4.21 (C3a-*H* (overlapped)), 4.97-4.98 (C9-*H*) and between δ 4.97-4.98 (C9-*H*) and δ 2.98-3.03 (C8-*H*H), 3.37-3.42 (C8-H*H*). Atom numbering is shown in Table 5-1.; ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 13.92 (CH₃), 14.10 (CH₃), 31.09 (CH), 47.16 (CH₂), 48.16 (CH), 49.92 (CH), 51.76 (CH₂), 58.32 (C), 61.30 (CH₂), 62.42 (CH₂), 95.51 (CH), 104.53 (CH), 127.51 (CH), 128.48 (CH), 128.59 (CH), 136.61 (C), 144.49 (CH), 155.79 (C), 167.70 (C), 170.24 (C), 172.19 (C). Selected HMBC correlations are between δ 2.98-3.03 (C8-*H*H), 3.22-3.24 (C5a-*H*,8a-*H* (overlapped)) and δ 51.76 (C8), and between δ 3.37-3.42 (C8-H*H*) and δ 4.97-4.98 (C9-*H*) and δ 51.76 (C8), and between δ 3.37-3.42 (C8-H*H*) and δ 4.97-4.98 (C9-*H*) and δ 2.98-3.03 (C8-*H*H), 128.59 (CH), 136.61 (C), 144.49 (CH), 155.79 (C), 167.70 (C), 170.24 (C), 172.19 (C). Selected HMBC correlations are between δ 2.98-3.03 (C8-*H*H), 3.22-3.24 (C5a-*H*,8a-*H* (overlapped)) and δ 51.76 (C8), and between δ 3.37-3.42 (C8-H*H*) and δ 48.16 (C5a).

3b: (Table 5-1, entry 9, 1.0 mmol scale, 361 mg, ca. 73%); $R_f = 0.5$ (hexane-ether = 1 : 8); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.21 (t, J = 7.1 Hz, 3H), 1.34 (t, J = 7.1 Hz, 3H), 2.97-3.04 (m, 1H), 3.17-3.25 (m, 2H), 3.37-3.41 (m, 1H), 4.01 (d, J = 14.7 Hz, 1H), 4.01-4.11 (m, 2H), 4.20 (dq, J = 10.7, 7.1 Hz, 1H), 4.27-4.44 (m, 2H), 4.92 (d, J = 14.7 Hz, 1H), 5.05-5.06 (m, 1H), 5.35 (dd, J = 2.5, 1.0 Hz, 1H), 7.24-7.34 (m, 5H). Selected NOEs are between δ 3.17-3.25 (C5a-*H*,8a-*H* (overlapped)) and δ 3.37-3.41 (C8-H*H*), 4.01-4.11 (C3a-*H* (overlapped)), 5.05-5.06 (C9-*H*) and between δ 5.05-5.06 (C9-*H*) and δ 2.97-3.04 (C8-*H*H), 3.37-3.41 (C8-H*H*).; ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 13.90 (CH₃), 14.05 (CH₃), 30.55 (CH), 47.14 (CH₂), 47.78 (CH), 51.18 (CH), 51.44 (CH₂), 58.33 (C), 61.37 (CH₂), 62.53 (CH₂), 97.12 (CH), 105.00 (CH), 127.50 (CH), 128.42 (CH), 128.55 (CH), 129.04 (C), 136.42 (C), 154.89 (C), 167.39 (C), 169.75 (C), 171.76 (C).

Transformation of 3a to 4a (Table 5-2, entry 1): To a solution of **3a** (252 mg, 0.62 mmol) in $ClCH_2CH_2Cl$ (1.0 mL) was added 1M HCl/Ether (0.62 mL, 0.62 mmol) and H₂O (9 mg, 0.5 mmol). The mixture was stirred at 80 °C for 20 h. The reaction mixture was concentrated under reduced pressure. The residue was purified by column chromatography over silica gel eluting with hexane-Et₂O to give **4a** (139 mg, 78%).

4a: $R_f = 0.7$ (hexane-ether = 1 : 8); colorless oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.21 (t, J = 7.1 Hz, 3H), 1.33 (t, J = 7.1 Hz, 3H), 2.23 (dd, J = 17.2, 8.2 Hz, 1H), 2.76 (dd, J = 17.2, 8.2 Hz, 1H), 2.86 (d, J = 9.6 Hz, 1H), 3.18 (dddd, J = 8.2, 8.2, 6.3, 4.5 Hz, 1H), 3.51 (dd, J = 9.6, 4.5 Hz, 1H), 3.69 (d, J = 6.3 Hz, 1H), 4.05-4.21 (m, 2H), 4.20 (d, J = 14.7 Hz, 1H), 4.28-4.50 (m, 2H), 4.60 (d, J = 14.7 Hz, 1H), 6.73 (d, J = 2.0 Hz, 1H), 7.16-7.19 (m, 2H), 7.26 (d, J = 2.0 Hz, 1H), 7.27-7.34 (m, 3H). Selected NOEs are between δ 3.18 (C7a-*H*) and δ 3.69 (C4a-*H*), 3.51 (C7-H*H*), 2.76 (C8-H*H*) and between δ 3.69 (C4a-*H*) and δ 3.51 (C7-H*H*). Atom numbering is shown in Table 5-1.; ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 13.94 (CH₃), 14.09 (CH₃), 24.68 (CH₂), 31.33 (CH), 46.89 (CH₂), 47.18 (CH), 51.57 (CH₂), 53.99 (C), 61.71 (CH₂), 62.00 (CH₂), 111.93 (CH), 113.08 (C), 127.82 (CH), 128.51 (CH), 128.81 (CH), 136.14 (C), 140.86 (CH), 149.82 (C), 168.74 (C), 169.81 (C), 171.00 (C). Selected HMBC correlations are between δ 2.23 (C8-*H*H), 2.76 (C8-H*H*), 2.86 (C7-*H*H), 3.69 (C4a-*H*) and δ 31.33 (C7a), δ 2.86 (C7-*H*H), 3.69 (C4a-*H*) and δ 31.33 (C7a), δ 2.86 (C7-*H*H), 3.69 (C4a-*H*) and δ 3.19 (C4a), 1143, 1366, 1243, 1147, 1109, 1038 cm⁻¹; MS (EI) m/z 411 (M⁺, 10), 322 (14), 248 (33), 204 (42), 84 (100%); HRMS (EI) m/z 411.1691 (calcd for C₂₃H₂₅NO₆ 411.1662).

4b: (Table 5-2, entry 2, 0.39 mmol scale, 143 mg, 74%); $R_f = 0.7$ (hexane-ether = 1 : 4); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.23 (t, J = 7.1 Hz, 3H), 1.32 (t, J = 7.1 Hz, 3H), 2.19 (dd, J = 17.4, 8.2 Hz, 1H), 2.74 (dd, J = 17.4, 8.2 Hz, 1H), 2.86 (d, J = 9.8 Hz, 1H), 3.18 (dddd, J = 8.2, 8.2, 6.1, 4.4 Hz, 1H), 3.51 (dd, J = 9.8, 4.4 Hz, 1H), 3.67 (d, J = 6.1 Hz, 1H), 4.03-4.46 (m, 4H), 4.17 (d, J = 14.7 Hz, 1H), 4.62 (d, J = 14.7 Hz, 1H), 6.67 (s, 1H), 7.17-7.19 (m, 2H), 7.28-7.35 (m, 3H). Selected NOEs are between δ 3.18 (C7a-H) and δ 3.67 (C4a-H), 3.51

(C7-H*H*), 2.74 (C8-H*H*) and between δ 3.67 (C4a-*H*) and δ 3.51 (C7-H*H*).; ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 13.94 (CH₃), 14.07 (CH₃), 24.47 (CH₂), 31.06 (CH), 46.89 (CH₂), 46.96 (CH), 51.47 (CH₂), 53.79 (C), 61.93 (CH₂), 62.26 (CH₂), 113.39 (CH), 115.80 (C), 120.51 (C), 127.90 (CH), 128.50 (CH), 128.86 (CH), 135.98 (C), 151.48 (C), 168.28 (C), 169.25 (C), 170.73 (C). Selected HMBC correlations are between δ 2.19 (C8-*H*H), 2.74 (C8-H*H*), 2.86 (C7-*H*H), 3.67 (C4a-*H*) and δ 31.06 (C7a), δ 2.86 (C7-*H*H), 3.51 (C7-H*H*), 3.67 (C4a-*H*) and δ 24.47 (C8), and between δ 3.67 (C4a-*H*) and δ 53.79 (C4).; IR (neat) 2981, 1738, 1697, 1516, 1443, 1366, 1293, 1246, 1194, 1045, 1029 cm⁻¹; MS (EI) m/z 491 (M⁺, 40), 489 (39), 418 (19), 416 (19), 372 (38), 370 (36), 91 (100%); HRMS (EI) m/z 489.0797, 491.0782 (calcd for C₂₃H₂₄BrNO₆ 489.0787, 491.0767).

6a: (0.93 mmol scale, 227 mg, ca. 59%); $R_f = 0.5$ (ether = 1 : 8); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.29 (t, J = 7.1 Hz, 3H), 1.35 (t, J = 7.2 Hz, 3H), 2.60-2.69 (m, 1H), 2.83 (d, J = 12.3 Hz, 1H), 3.09 (dd, J = 10.2, 9.1 Hz, 1H), 3.44 (dd, J = 9.1, 7.7 Hz, 1H), 4.21-4.41 (m, 5H), 4.53 (d, J = 14.8 Hz, 1H), 5.31 (dd, J = 3.9, 2.9 Hz, 1H), 5.68-5.70 (m, 2H), 6.81 (d-like, J = 2.5 Hz, 1H), 7.21-7.35 (m, 5H). Selected NOEs are between δ 2.60-2.69 (C4a-*H*) and δ 5.68-5.70 (C4-*H* (overlapped)), 5.31 (C8a-*H*), 3.44 (C5-H*H*), and between δ 3.09 (C5-*H*H) and δ 5.68-5.70 (C4-*H* (overlapped)). Atom numbering is shown in Table 5-3.; ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 14.09 (CH₃), 14.13 (CH₃), 36.46 (CH), 46.82 (CH₂), 48.20 (CH₂), 52.51 (CH), 56.38 (C), 62.17 (CH₂), 62.21 (CH₂), 84.48 (CH), 104.38 (CH), 112.49 (CH), 127.60 (CH), 128.16 (CH), 128.74 (CH), 136.64 (C), 143.30 (C), 154.35 (CH), 167.43 (C), 169.48 (C), 170.45 (C). Selected HMBC correlations are between δ 3.09 (C5-*H*H), 3.44 (C5-H*H*), and δ 36.46 (*C*4a), δ 3.44 (C5-H*H*) and δ 52.51 (*C*7a), δ 3.44 (C5-H*H*), 2.83 (C7a-*H*), and δ 112.49 (C4), and between δ 2.83 (C7a-*H*) and δ 56.38 (C8).

7a: (Table 5-2, entry 2, 1.0 mmol scale, 321 mg, 78%); $R_f = 0.5$ (hexane-ether = 1 : 8) ; pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.28 (t, J = 7.1 Hz, 3H), 1.36 (t, J = 7.1 H, 3H), 2.39 (dd, J = 14.8, 10.6 Hz, 1H), 2.59 (ddddd, J = 12.5, 10.6, 9.0, 7.4, 4.9 Hz, 1H), 2.66 (dd, J = 12.5, 10.6, 9.0, 7.4, 4.9 Hz, 1H), 2.66 (dd, J = 12.5, 10.6, 9.0, 7.4, 4.9 Hz, 1H), 2.66 (dd, J = 12.5, 10.6, 9.0, 7.4, 4.9 Hz, 1H), 2.66 (dd, J = 12.5, 10.6, 9.0, 7.4, 4.9 Hz, 1H), 2.66 (dd, J = 12.5, 10.6, 9.0, 7.4, 4.9 Hz, 1H), 2.66 (dd, J = 12.5, 10.6, 9.0, 7.4, 4.9 Hz, 1H), 2.66 (dd, J = 12.5, 10.6, 9.0, 7.4, 4.9 Hz, 1H), 2.66 (dd, J = 12.5, 10.6, 9.0, 7.4, 4.9 Hz, 1H), 2.66 (dd, J = 12.5, 10.6, 9.0, 7.4, 4.9 Hz, 1H), 2.66 (dd, J = 12.5, 10.6, 9.0, 7.4, 4.9 Hz, 1H), 2.66 (dd, J = 12.5, 10.6, 9.0, 7.4, 4.9 Hz, 1H), 2.66 (dd, J = 12.5, 10.6, 9.0, 7.4, 4.9 Hz, 1H), 2.66 (dd, J = 12.5, 10.6, 9.0, 7.4, 4.9 Hz, 1H), 2.66 (dd, J = 12.5, 10.6, 9.0, 7.4, 4.9 Hz, 1H), 2.66 (dd, J = 12.5, 10.6, 9.0, 7.4, 9.0, 7.4, 9.5 (dd, J = 12.5, 10.6, 9.5 (dd, J = 12.5, 10.5 (dd, J = 12.5,

14.8, 4.9 Hz, 1H), 3.03 (dd, J = 9.2, 9.0 Hz, 1H), 3.07 (d, J = 12.5 Hz, 1H), 3.39 (dd, J = 9.2, 7.4 Hz, 1H), 4.13-4.21 (m, 1H), 4.31-4.40 (m, 4H), 4.62 (d, J = 14.7 Hz, 1H), 6.22 (d, J = 2.0 Hz, 1H), 7.21-7.36 (m, 5H), 7.38 (d, J = 2.0 Hz, 1H). Selected NOEs are between δ 2.59 (C4a-*H*) and δ 3.39 (C5-H*H*) and between δ 3.03 (C5-*H*H), 3.07 (C7a-*H*) and δ 2.39 (C4-*H*H). Atom numbering is shown in Table 5-3.; ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 14.07 (CH₃), 14.09 (CH₃), 25.72 (CH₂), 34.10 (CH), 46.58 (CH₂), 49.72 (CH₂), 51.86 (CH), 57.09 (C), 61.97 (CH₂), 62.73 (CH₂), 110.31 (CH), 120.77 (C), 127.58 (CH), 128.20 (CH), 128.73 (CH), 136.63 (C), 143.65 (CH), 145.66 (C), 167.09 (C), 168.26 (C), 171.00 (C). Selected HMBC correlations are between δ 2.39 (C4-*H*H), 2.66 (C4-H*H*), 3.39 (C5-H*H*), 3.07 (C7a-*H*) and δ 34.10 (C4a), δ 3.39 (C5-H*H*), 2.59 (C4a-*H*), and δ 51.86 (C7a), and between δ 3.07 (C7a-*H*) and δ 57.09 (*C*).; IR (neat) 2982, 2937, 1739, 1497, 1443, 1367, 1256, 1193, 1148, 1039 cm⁻¹; MS (EI) m/z 411 (M⁺, 14), 293 (14), 119 (58), 84 (100%); HRMS (EI) m/z 411.1696 (calcd for C₂₃H₂₅NO₆ 411.1682).

7b: (Table 5-2, entry 7, 1.0 mmol scale, 200 mg, 48%); $R_f = 0.6$ (hexane-ether = 1 : 8) ; pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 0.919-1.01 (m, 2H), 1.13-1.23 (m, 3H), 1.26 (t, J = 7.1 Hz, 3H), 1.33 (t, J = 7.1 Hz, 3H), 1.60-1.68 (m, 3H), 1.71-1.74 (m, 3H), 2.44 (dd, J = 15.0, 11.0 Hz, 1H), 2.61 (ddddd, J = 12.7, 11.0, 9.4, 7.5, 4.9 Hz, 1H), 2.71 (dd, J = 15.0, 4.9 Hz, 1H), 3.04 (d, J = 12.7 Hz, 1H), 3.06 (dd, J = 13.6, 7.2 Hz, 1H), 3.16 (dd, J = 9.4, 9.1 Hz, 1H), 3.21 (dd, J = 13.6, 7.4 Hz, 1H), 3.53 (dd, J = 9.1, 7.5 Hz, 1H), 4.15 (dq, J = 10.7, 7.1 Hz, 1H), 4.26-4.41 (m, 3H), 6.24 (d, J = 2.0 Hz, 1H), 7.38 (d, J = 2.0 Hz, 1H). Selected NOEs are between δ 2.61 (C4a-H) and δ 3.53 (C5-HH), between δ 3.16 (C5-HH), 3.04 (C7a-H) and δ 2.44 (C4-HH).; ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 14.02 (CH₃), 14.07 (CH₃), 25.82 (CH₂), 25.85 (CH₂), 25.87 (CH₂), 26.45 (CH₂), 30.87 (CH₂), 30.90 (CH₂), 34.18 (CH), 36.24 (CH), 49.20 (CH₂), 51.31 (CH₂), 51.92 (CH), 57.03 (C), 61.85 (CH₂), 62.63 (CH₂), 110.30 (CH), 120.68 (C), 143.59 (CH), 145.88 (C), 167.07 (C), 168.24 (C), 171.07 (C). Selected HMBC correlations are between δ 2.44 (C4-HH), 2.71 (C4-HH), 3.53 (C5-HH), 3.04 (C7a-H) and δ 34.18 (C4a), δ 3.53 (C5-HH), 2.61 (C4a-H), and δ 51.92 (C7a), and between δ 3.04 (C7a-H) and δ 57.03 (C8); IR

(neat) 2929, 2851, 1747, 1699, 1498, 1446, 1367, 1263, 1193, 1036 cm⁻¹; MS (EI) m/z 417 (M⁺, 34), 335 (91), 59 (100%); HRMS (EI) m/z 417.2149 (calcd for C₂₃H₃₁NO₆ 417.2151).

11: (0.5 mmol scale, 118 mg, 45%); $R_f = 0.7$ (hexane-EtOAc = 1 : 1); red oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.22 (t, J = 7.1 Hz, 3H), 1.36 (t, J = 7.1 Hz, 3H), 3.46 (dd, J = 9.9, 5.9 Hz, 1H), 3.62 (dd, J = 6.5, 5.9 Hz, 1H), 3.70 (d, J = 9.9 Hz, 1H), 4.05-4.13 (m, 1H), 4.07 (d, J = 6.5Hz, 1H), 4.20-4.28 (m, 1H), 4.32-4.49 (m, 4H), 7.05-7.07 (m, 2H), 7.26-7.32 (m, 3H), 7.82 (s, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 13.83 (CH₃), 13.99 (CH₃), 43.34 (CH), 46.86 (CH₂), 48.03, 48.05 (CH₂, CH), 55.54 (C), 62.48 (CH₂), 63.03 (CH₂), 125.30 (C), 127.88 (CH), 127.99 (CH), 128.76 (CH), 134.64 (CH), 135.51 (C), 139.54 (C), 143.88 (C), 166.81 (C), 167.46 (C), 170.05 (C), 188.11 (C); ¹H NMR (400 MHz, C₆D₆) δ (ppm) 0.726 (t, J = 7.1 Hz, 3H), 1.10 (t, J =7.1 Hz, 3H), 2.71 (dd, J = 9.8, 5.9 Hz, 1H), 3.46 (dd, J = 6.8, 5.9 Hz, 1H), 3.51(d, J = 9.8 Hz, 1H), 3.60 (dq, J = 10.6, 7.1 Hz, 1H), 3.84 (dq, J = 10.6, 7.1 Hz, 1H), 3.88 (d, J = 14.7 Hz, 1H), 3.99 (d, J = 6.8 Hz, 1H), 4.10 (d, J = 14.7 Hz, 1H), 4.30-4.38 (m, 2H), 6.83-6.86 (m, 2H), 6.99-7.10 (m, 3H), 8.13 (s, 1H). Selected NOEs are between δ 3.46 (C7a-H) and δ 3.99 (C4a-H), 2.71 (C7-HH), and between δ 3.99 (C4a-H) and δ 2.71 (C7-HH). Atom numbering is shown in Scheme 5-3.; ¹³C NMR (100.6 MHz, C₆D₆) δ (ppm) 13.56 (CH₃), 13.95 (CH₃), 43.74 (CH), 46.66 (CH₂), 47.72 (CH₂), 48.43 (CH), 56.15 (C), 62.42 (CH₂), 62.73 (CH₂), 124.92 (C), 127.88 (CH), 128.11 (CH), 128.79 (CH), 135.06 (CH), 136.13 (C), 140.32 (C), 144.42 (C), 166.76 (C), 167.91 (C), 169.73 (C), 187.97 (C). Selected HMBC correlations are between δ 3.99 (C4a-H) and δ 56.15 (C4), δ 2.71 (C7-HH), 3.51 (C7-HH), 3.46 (C7a-H), 3.99 (C4a-H) and δ 187.97 (C8), and between δ 3.51 (C7-HH), 3.99 (C4a-H) and δ 43.74 (C7a).; IR (neat) 2981, 1738, 1699, 1667, 1409, 1258, 1229, 1195, 1095, 1016 cm⁻¹; MS (EI) m/z 521 (M⁺, 25), 519 (M⁺, 24), 402 (12), 401 (11), 119 (15), 118 (14), 91 (100%); HRMS (EI) m/z 519.0342, 521.0331 (calcd for C₂₃H₂₂BrNO₆S 519.0351, 521.0331).

12: (0.75 mmol scale, 256 mg, 61%); $R_f = 0.3$ (ether = 1 : 1); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.24 (t, J = 7.1 Hz, 3H), 1.27 (t, J = 7.1 Hz, 3H), 3.29 (dddd, J = 9.0, 7.0, 6.1, 5.5

Hz, 1H), 3.39 (dd, J = 7.0, 4.1 Hz, 1H), 3.60 (dd, J = 10.2, 9.0 Hz, 1H), 3.84 (d, J = 4.1 Hz, 1H), 3.87 (dd, J = 10.2, 5.5 Hz, 1H), 4.07-4.26 (m, 4H), 4.54 (d, J = 14.8 Hz, 1H), 4.63 (d, J = 14.8 Hz, 1H), 5.76 (d, J = 6.1 Hz, 1H), 7.14 (ddd, J = 8.2, 1.0, 1.0 Hz, 1H), 7.22-7.39 (m, 8H), 7.69 (ddd, J = 8.0, 2.0, 1.6 Hz, 1H), 7.90 (ddd, J = 8.3, 0.9, 0.9 Hz, 1H), 8.54 (dd, J = 4.9, 1.6 Hz, 1H), 8.60 (d, J = 2.0 Hz, 1H). Selected NOEs are between δ 3.39 (C3-*H*), 3.87 (C5-H*H*), 3.29 (C4-*H*) and δ 5.76 (C*H*(-3-Pyr)O) and between δ 3.29 (C4-*H*) and δ 3.84 (C*H*(CO₂Et)₂), 3.60 (C5-*H*H).; ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 13.96 (CH₃), 13.99 (CH₃), 39.25 (CH), 44.32 (CH), 46.38 (CH₂), 47.00 (CH₂), 51.24 (CH), 61.91 (CH₂), 62.07 (CH₂), 91.01 (CH), 108.36 (CH), 120.28 (CH), 123.62 (CH), 124.66 (CH), 127.45 (C), 127.80 (CH), 128.18 (CH), 128.32 (CH), 128.81 (CH), 131.60 (C), 135.27 (CH), 135.72 (C), 143.21 (C), 148.65 (CH), 150.96 (CH), 167.63 (C), 168.30 (C), 171.67 (C). Selected HMBC correlations are between δ 3.60 (C5-*HH*), 3.87 (C5-H*H*), 3.49 (C4), 46.39, 91 (100%); HRMS (EI) m/z 557.2264 (calcd for C₃₀H₃₁N₅O₆ 557.2274).

References

 ¹ For examples, see: (a) V. Benešová, Z. Samek, V. Herout, F. Šorm, *Collect. Czech. Chem. Commun.* **1969**, *34*, 582. (b)V. Benešová, V. Herout, F. Šorm, *Collect. Czech. Chem. Commun.* **1969**, *34*, 1810. (c) L. D. Scheei, V. B. Perone, R. L. Larkin, R. E. Kupel, *Biochemistry* **1963**, *2*, 1127. (d) F. Bohlmann, R. Zdero, M. Grenz, *Chem. Ber.* **1974**, *107*, 2730. (e) R. Kazlauskas, P. T. Murphy, R. J. Wells, J. J. Daly, P. Schönholzer, *Tetrahedron Lett.* **1978**, *19*, 4951. (f) S. M. Kupchan, R. J. Hemingway, D. Werner, A. Karim, J. Org. Chem. **1969**, *34*, 3903. (g) W. Herz, and G. Högenauer, *J. Org. Chem.* **1962**, *27*, 905. (h) X. Yu, Z. Che, H. Xu, *Chem. - Eur. J.* **2017**, *23*, 4467. (i) R. Achini, W. Oppolzer, E. Pfenninger, U.S. Pat. 4171369, **1975**. ² (a) K. Hayakawa, M. Yodo, S. Ohsuki, K. Kanematsu, J. Am. Chem. Soc. 1984, 106, 6735. (b)
Y. I. Horak, R. Z. Lytvyn, Y. V. Homza, V. P. Zaytsev, D. F. Mertsalov, M. N. Babkina, E. V. Nikitina, T. Lis, V. Kinzhybalo, V. S. Matiychuk, F. I. Zubkov, A. V. Varlamov, M. D. Obushak, *Tetrahedron Lett.* 2015, 56, 4499. (c) R. E. Patre, S. Gawas, S. Sen, P. S. Parameswaran, S. G. Tilve, *Tetrahedron Lett.* 2007, 48, 3517. (d) H. Kotsuki, A. Kawamura, M. Ochi, *Chem. Lett.* 1981, 917. (e) M. G. B. Drew, A. Jahans, L. M. Harwood, S. A. B. H. Apoux, *Eur. J. Org. Chem.* 2002, 3589. (f) S. Sun, W. V. Murray, J. Org. Chem. 1999, 64, 5941. (g) K. Lu, T. Luo, Z. Xiang, Z. You, R. Fathi, J. Chen, Z. Yang, J. Comb. Chem. 2005, 7, 958. (h) A. E. Bober, J. T. Proto, K. M. Brummond, Org. Lett. 2017, 19, 1500.

³ P. Kim, J. M. Tsuruda, M. M. Olmstead, S. Eisenberg, M. J. Kurth, *Tetrahedron Lett.* 2002, *43*, 3963.
⁴ Y. He, Y. Chen, H. Wu, C. J. Lovely, *Org. Lett.* 2003, *5*, 3623.

⁵ (a) E. Ciganek, Org. React. 1984, 32, 1. (b) W. R. Roush, In Comprehensive Organic Synthesis,
B. M. Trost, I. Fleming, Eds.; Pergamon, 1991, Vol. 5, 513. (c) C. O. Kappe, S. S. Murphree, A.
Padwa, Tetrahedron 1997, 53, 14179. (d) G. Brieger, J. N. Bennett. Chem. Rev. 1980, 80, 63. (e)
F. Fringuelli, A. Taticchi, The Diels-Alder Reaction - Selected Practical Methods, John Wiley & Sons: Chichester, 2002.

⁶ J. A. Joule, K. Mills, *Heterocyclic Chemistry*, 5th ed, John Wiley & Sons, 2013.

⁷ (a) B. Ya. Simkin, V. I. Minkin, M. N. Glukhovtsev, *Adv. Heterocycl. Chem.* 1993, *56*, 303. (b)
M. P. Carmody, M. J. Cook, R. D. Tack, *Tetrahedron* 1976, *32*, 1767. (c) Clive W. Bird, *Tetrahedron* 1996, *52*, 9945. (b) C. W. Bird, *Tetrahedron* 1992, *48*, 335. (d) G. P. Bean, *J. Org. Chem.* 1998, *63*, 2497. (e) M. K. Cyrański, P. v. R. Schleyer, T. M. Krygowski, H. Jiao, G. Hohlneicher, *Tetrahedron* 2003, *59*, 1657.

⁸ (a) G. W. Huber, S. Iborra, and A. Corma, *Chem. Rev.* 2006, **106**, 4044. (b) A. Corma, S. Iborra, and A. Velty, *Chem. Rev.* 2007, **107**, 2411. (c) X. Tong, Y. Ma, and Y. Li, *Appl. Catal. A: General* 2010, **385**, 1. (d) J. N. Chheda, G. W. Huber, and J. A. Dumesic, *Angew. Chem. Int. Ed.*

2007, **46**, 7164. (e) A. J. Ragauskas, C. K. Williams, B. H. Davison, G. Britovsek, J. Cairney, C. A. Eckert, W. J. Frederick Jr., J. P. Hallett, D. J. Leak, C. L. Liotta, J. R. Mielenz, R. Murphy, R. Templer, and T. Tschaplinski, *Science* 2006, **311**, 484. (f) P. W. Lichtenthaler, *Acc. Chem. Res.* 2002, **35**, 728. (g) R.-J. van Putten, J. C. van der Waal, E. de Jong, C. B. Rasrendra, H. J. Heeres, and J. G. de Vries, *Chem. Rev.* 2013, **113**, 1499. (h) B. R. Caes, R. E. Teixeira, K. G. Knapp, and R. T. Raines, *ACS Sustainable Chem. Eng.* **2015**, *3*, 2591. (i) P. D. de María N. Guajardo, *ChemSusChem* **2017**, *10*, 4123.

⁹ (a) J.-G. Ji, D.-Y. Zhang, Y.-H. Ye, Q.-Y. Xing, *Tetrahedron Lett.* **1998**, *39*, 6515. (b) W. A. Feld, D. G. Evans, *J. Chem. Eng. Data* **1983**, *28*, 138.

¹⁰ (a) S. Yamazaki, H. Sugiura, S. Ohashi, K. Ishizuka, R. Saimu, Y. Mikata, A. Ogawa, *J. Org. Chem.* 2016, 81, 10863. (b) H. Sugiura, S. Yamazaki, K. Go, A. Ogawa., *Eur. J. Org. Chem.* doi/abs/10.1002/ejoc.201801508.

¹¹ (a) K. Schofield, Aromatic Reactivity. *In Hetero-Aromatic Nitrogen Compounds*. 1967 Springer, Boston, MA. (b) H. H. Jaffe, H. L. Jones, *Adv. Heterocycl. Chem.* **1964**, *3*, 209.

¹² (a) A. D. Becke, J. Chem. Phys. 1993, 98, 5648. (b) C. Lee, W. Yang, R. G. Parr, Phys. Rev. B
1998, 37, 785.

¹³ Gaussian 09, Revision D.01, M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, B. Mennucci, G. A. Petersson, H. Nakatsuji, M. Caricato, X. Li, H. P. Hratchian, A. F. Izmaylov, J. Bloino, G. Zheng, J. L. Sonnenberg, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, J. A. Montgomery. Jr., J. E. Peralta, F. Ogliaro, M. Bearpark, J. J. Heyd, E. Brothers, K. N. Kudin, V. N. Staroverov, R. Kobayashi, J. Normand, K. Raghavachari, A. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, N. Rega, J. M. Millam, M. Klene, J. E. Knox, J. B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, R. L. Martin, K. Morokuma, V. G.

Zakrzewski, G. A. Voth, P. Salvador, J. J. Dannenberg, S. Dapprich, A. D. Daniels, Ö. Farkas, J.

- B. Foresman, J. V. Ortiz, J. Cioslowski, D. J. Fox, Gaussian, Inc., Wallingford CT, 2009.
- ¹⁴ J.-D. Chai, M. Head-Gordon, *Phys. Chem. Chem. Phys.* **2008**, *10*, 6615.
- ¹⁵ (a) Y. Zhao and D. G. Truhlar, *Theor. Chem. Acc.* 2008, **120**, 215-241. (b) Y. Zhao and D. G. Truhlar, *Acc. Chem. Res.* 2008, **41**, 157-167.
- ¹⁶ (a) E. Cancès, B. Mennucci, J. Tomasi, J. Chem. Phys. 1997, 107, 3032. (b) M. Cossi, V. Barone, B. Mennucci, J. Tomasi, Chem. Phys. Lett. 1998, 286, 253. (c) Mennucci, B.; Tomasi, J. J. Chem. Phys. 1997, 106, 5151.
- ¹⁷ (a) K. Fukui, J. Phys. Chem. 1970, 74, 4161. (b) C. Gonzalez, H. B. Schlegel, J. Phys. Chem.
 1989, 90, 2154.
- ¹⁸ P. P. Bera, M. Head-Gordon, and T. J. Lee, J. Chem. Phys. **2013**, 139, 174302.
- ¹⁹ S. Yamazaki, K. Ohnitsu, K. Ohi, T. Otsubo, K. Moriyama, Org. Lett. 2005, 7, 759.
- ²⁰ (a) H. Anan, A. Tanaka, R. Tsuzuki, M. Yokota, T. Yatsu, T. Fujikura, Chem. Pharm. Bull.
- **1996**, *44*, 1865. (b) F. A. Abdel-Magid, G. K. Carson, D. B. Harris, A. C. Maryanoff, D. R. Shah, *J. Org. Chem.* **1996**, *61*, 3849.
- ²¹ T. M. Cresp, M. V. Sargent, P. Vogel, J. Chem. Soc., Perkin Trans. 1 1974, 37.
- ²² P. Q. Le, T. S. Nguyen, and J. A. May, Org. Lett. 2012, 14, 6104.
- ²³ L. H. Kiemm, K. W. Gopinath, J. Heterocyclic Chem. 1965, 2, 225.
- ²⁴ (a) P. V. Bedworth, Y. Cai, A. Jen, and S. R. Marder, *J. Org. Chem.* **1996**, *61*, 2242. (b) US Patent. US 7728018, **2010**
- ²⁵ (a) I. Hagedorn, W. Hohler, Angev. Chem. Internat. Ed. 1975, 14, 486. (b) P. Valenta, N. A.
- Drucker, J. W. Bode and P. J. Walsh, Org. Lett. 2009, 11, 2117.
- ²⁶ W. Zeng, T. E. Ballard, C. Melander, *Tetrahedron Lett.* 2006, 47, 5923.
- ²⁷ H. Keskin, R. E. Miller, F. F. Nord, J. Org. Chem. 1951, 16, 199.

Chapter 6

Intramolecular Cyclization Reactions of β -Substituted Cinnamylamides of Ethenetricarboxylate

6-1 Introduction

The development of atom-economical and efficient synthetic reactions which proceed under mild conditions is the continuing aim of organic chemistry. In this regard, intramolecular reactions of highly electrophilic ethenetricarboxylate with high chemo- and stereoselectivity are particularly attractive in view of their wide applicability for various synthetic reactions.

The alkyl substituents at alkene positions for the esters or amides have been studied by Snider and Roush¹ and by Yamazaki group work.² Lewis acid-promoted reactions of these substrates (Scheme 6-1) gave halogen or OH-incorporated five-membered rings, intramolecular ene adducts and also six-membered rings from 2-alkyl or aryl substituted (R^1) propenyl ester or amides of ethenetricarboxylate. In addition, reaction of 1,1-diethyl 2-hydrogen ethenetricarboxylate 1 and 2-furylmethylamines in the presence of EDCI/HOBt/Et₃N at room temperature led directly to intramolecular Diels-Alder adducts, in chapter 2.³

The reaction of ethenetricarboxylates bearing aryl-substituted alkenyl groups as an extension of the alkene moiety was examined in chapters 3 and 4. In chapter 3, sequential intramolecular reactions of 1,1-diethyl 2-hydrogen ethenetricarboxylate 1 with *E*-cinnamylamines 2 under the amide formation conditions led to pyrrolidine products in one pot, via intramolecular [2 + 2], [4 + 2] (IMDA) and some other cyclizations.⁴ The types of the products depend on the substituents on benzene ring and the reaction conditions. Furthermore, the reaction of 1 with *Z*-cinnamylamines on heating gave [4 + 2] cycloadducts, *cis*-fused tricyclic compounds as the major products in chapter 4.⁶ Reaction of 1 with 3,3-diaryl-2-propen-1-amines gave *cis*- and *trans*-fused tricyclic compounds selectively, depending on the substituents on the benzene ring, reaction temperature and solvent.



Scheme 6-1. The reaction of the esters and amides of ethenetricarboxylates.

To extend the scope of [2 + 2] and [4 + 2] cycloadditions of the cinnamyl derivatives and also to gain an insight into the mechanisms determining the chemoselectivity of four-membered and six-membered-ring formations, and other five-membered ring cyclization, the reactions of β -alkyl (Me) and halogen substituted cinnamyl derivatives have been examined in this study. The effects on benzene ring to selectivity of [2 + 2] and [4 + 2] cycloadditions have also been examined.

6-2 Results and Discussion

First. reactions of 1,1-diethyl 2-hydrogen ethenetricarboxylate 1 and (E)-N-benzyl-3-phenyl-2-buten-1-amine 2a in the presence of EDCI/HOBt/Et₃N have been examined (Scheme 6-2). The reaction in THF gave cyclobutane-fused pyrrolidines 3a in 22% yield as the isolable major product. The reaction in DMF gave 3a in a better yield (49%). The relative configuration of 3a was determined as shown in Scheme 6-2 by NOESY experiment. The intermediate amide, [4 + 2] cycloadducts and possible ene adducts were not observed under the reaction conditions. The reaction of 1 and 2a at higher temperature (80-110 $^{\circ}$ C) in benzene and toluene gave a complex mixture. The reaction of 1 and (E)-N-benzyl-3-p-tolyl-2-buten-1-amine (2b) at room temperature gave a mixture possibly containing cyclobutane-fused pyrrolidine but the products could not be purified. The reaction of 1 and 2b at 110 °C in toluene and DMF gave a complex mixture.



Scheme 6-2. Reactions of 1,1-diethyl 2-hydrogen ethenetricarboxylate 1 and (*E*)-*N*-benzyl -3-aryl-2-buten-1-amine 2.

Next, the reaction of 1,1-diethyl 2-hydrogen ethenetricarboxylate 1 and cinnamylamines bearing electron-withdrawing groups in the presence of the amide condensation reagents was examined. Reaction of 1 and (*E*)-3-(4-(trifluoromethyl)phenyl)-2-buten-1-amine 2c with EDCI/HOBt/Et₃N at room temperature and 110 °C in THF and DMF gave tetrahydrobenz[*f*]isoindoline 4c-aq (axial methyl substituent) as the major product via [4 + 2] cycloaddition (Table 6-1). The reaction of **1** and **2c** at 110 °C in toluene gave a complex mixture. The stereochemistry of **4c-aq** was determined by NOEs.

The *trans*-fused tetrahydrobenz[f]isoindoline **4c-aq** may be formed via amide formation, IMDA reaction and H-transfer, similar to the reaction of *E*-cinnamylamines. Selective formation of the stereochemistry at the 4-methyl group may arise from the protonation from less hindered side (*cis* to adjacent H), similar to the reaction of 3,3-diaryl-2-propen-1-amines.

On the other hand, reaction of **1** and (*E*)-3-(4-nitrophenyl)-2-buten-1-amine **2d** with EDCI/HOBt/Et₃N at room temperature and 110 °C in DMF and toluene gave tetrahydrobenz[*f*]isoindoline **4d-eq** with the 4-methyl group *trans* to adjacent H as the major isolable product. Epimerization at C4 of the stereoisomer with 4-methyl *cis* to adjacent H initially formed (axial methyl substituent) would possibly provide the more stable **4d-eq** (equatrial methyl substituent). Related epimerization at C4 of tetrahydrobenz[*f*]isoindoline, using KOH in butanol or DMSO was reported by Oppolzer et al. ^{5,6}

EtO ₂ C Co HO ₂ C	O ₂ Et +	+ 2c	Y EDCI HOBt Et ₃ N	Ph Et O Ph		Et CF ₃ Ph	H H H H H H H H H H
	∫ Ph			4c-ax			4d-eq
	Entry	2	Y	Solvent	Temp.	Product	Yield(%)
	1	2c	4-CF ₃	THF	r.t.	4c-ax	60
	2	2c	4-CF ₃	DMF	r.t.	4c-ax	73
	4	2c	4-CF ₃	DMF	110 °C	4c-ax	58
	6	2d	4-NO ₂	DMF	r.t.	4d-eq	24
	7	2d	4-NO ₂	Toluene	110 °C	4d-eq	ca. 60 ^a

Table 6-1. Reaction of (*E*)-3-aryl-2-buten-1-amines bearing electronwithdrawing groups **2c,d**.

^a Including a small amount of possible stereoisomer.

Reaction of 1 and (*E*)-3-(3-nitrophenyl)-2-buten-1-amine 2e with EDCI/HOBt/Et₃N at room temperature and 110 °C gave HOBt-incorporated 3,4-*trans*-pyrrolidine 5 as a single diastereomer in 40-85% yields (Scheme 6-3), in analogy to the reaction of *m*-nitrocinnamylamines and also 3-(3-pyridinyl)-2-propenylamine. The stereochemistry of 5 was deduced as shown in Scheme 6-3, similar to the proposed mechanism for formation of those HOBt adducts.⁴



Scheme 6-3. Reaction of 1 and (*E*)-3-aryl-2-buten-1- amine 2e.

The reaction of **1** and (*E*)-3-phenyl-3-bromo-2-propene-1-amine (β -bromo *cis*-cinnamylamine), **6** was examined next. Reaction of **1** and **6** with EDCI/HOBt/Et₃N at room temperature in THF or DMF gave the corresponding amide **7** in 48-75% yields as the isolable product. The reaction of **1** and **6** on heating at 80 °C gave *cis*-fused tricyclic compound **8** in 48% yield (Table 6-2).

Treatment of **7** with Et₃N or 1M HCl/EtOAc in benzene or toluene heating at 80-110 °C gave *cis*-fused tricyclic compound **8** in 32-67% yields.

The stereochemistry of **8** was determined by observed NOEs. The observed *cis*-fused stereochemistry is in accord with that of *Z*-cinnmaylamine and C4 stereochemistry may arise from the protonation from less hindered side (*cis* to adjacent H), similar to the reactions of 3,3-diaryl-2-propen-1-amines.⁶

Et(O₂C O₂C 1	.CO₂Et + HN Ph	Br EDCI HOBt C Et ₃ N 6 16 h Ph		2Et	$\Delta \rightarrow N$ Ph	$ \begin{array}{c} EtO_2C \\ O \\ H \\ H \\ Br \\ 8 \end{array} $
-	Entry	Starting Materials	Reagents	Solvent	Temp.	Product	Yield (%)
-	1	1+6	EDCI/HOBt/Et ₃ N	THF	r.t.	7	75
	2	1+6	EDCI/HOBt/Et ₃ N	Benzene	80 °C	8	48
	3	7	Et ₃ N (1 equiv.)	Benzene	80 °C	8	67
	4	7	Et ₃ N (1 equiv.)	Toluene	110 °C	8	47
	5	7	1M HCl/EtOAc (1 equiv)	Benzene	80 °C	a	
	6	7		Benzene	80 °C	b	
	7	7		DMF	80 °C	c	

Table 6-2. Reactions of 1 and (*E*)-3-phenyl-3-bromo-2-propene-1-amine 7.

^a A mixture mainly containing **8**. ^b A mixture containing a small amount of **8**. ^c A complex mixture.

In chapter 4, reaction of ethenetricalbocylate and Z-cinnamylamines at heating in the presence of EDCI/HOBt/Et₃N led *cis*-fused tricyclic compounds, and amide intermediate **A**-*cis* was not detected at room temperature.⁶ The possible isolation of the amide 7 by lower reactivity than that of **A**-*cis* is probably due to both steric and electronic effects of Br group (Scheme 6-4).



Scheme 6-4. Steric effects in [2+2] cycloaddition.

6-3 Conclusion

In summary, intramolecular cycloaddition reactions of β -substituted cinnamylamides of ethenetricarboxylate have been studied. Reaction of ethenetricarboxylic acid 1,1-diester and (*E*)-3-aryl-2-buten-1-amines with EDCI/HOBt/Et₃N led to pyrrolidine products in one pot, via intramolecular [2 + 2], [4 + 2] (IMDA) cycloadditions and HOBt-incorporated cyclization. The types of the products depend on the substituents on benzene ring, similarly to the reaction of *E*-cinnamylamines. Reaction of ethenetricarboxylic acid 1,1-diester and (*E*)-3-phenyl-3-bromo-2-propene-1-amine with EDCI/HOBt/Et₃N at room temperature gave the corresponding amide as an isolable product. Heating the amide with HCl or Et₃N gave a *cis*-fused tricyclic compound via [4 + 2] cycloaddition/H-transfer. Further study of the reaction of diversely substituted cinnamylamines and the corresponding amides under various conditions is in due course.

6-4 Experimental Section

General Procedures

¹H Chemical shifts are reported in ppm relative to Me₄Si. ¹³C Chemical shifts are reported in ppm relative to CDCl₃ (77.1 ppm). ¹³C mutiplicities were determined by DEPT and HSQC. Peak assignments are made by 2D COSY, HSQC, NOESY, and HMBC spectra. Mass analyzer type used for EI is double-focusing in the HRMS measurements. Column chromatography was performed on silica gel (75-150 μm).

1,1-Diethyl 2-hydrogen ethenetricarboxylate 1 was prepared according to the literature.⁷

Arylpropenyl esters, ethyl (*E*)-3-phenylbut-2-enoate **Xa** (for **2a**), ethyl (*E*)-3-(4-methylphenyl)but-2-enoate **Xb** (for **2b**), ethyl (*E*)-3-(4-trifluoromethylphenyl)but-2-enoate **Xc** (for **2c**), ethyl (*E*)-3-(4-nitrophenyl)but-2-enoate **Xd** (for **2d**), ethyl (*E*)-3-(3-nitrophenyl)but-2-enoate **Xe** (for **2e**) and the corresponding alcohols, (*E*)-3-phenylbut-2-en-1-ol **Ya**, (*E*)-3-(4-methylphenyl)but-2-en-1-ol **Yb**, (*E*)-3-(4-trifluoromethylphenyl)but-2-en-1-ol **Yc**, (*E*)-3-(4-nitrophenyl)but-2-en-1-ol **Yd**, and (*E*)-3-(3-nitrophenyl)but-2-en-1-ol **Ye** were prepared according to the literature.⁸

Ethyl (*E*)-3-(4-methylphenyl)but-2-enoate (Xb): (20 mmol scale, 3.8 g, 94%) $R_f = 0.6$ (hexane-EtOAc = 4 : 1); colorless oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.31 (t, *J* = 7.1 Hz, 3H), 2.36 (s, 3H), 2.56 (s, 3H), 4.20 (q, *J* = 7.1 Hz, 2H), 6.13 (s, 1H), 7.17 (d *J* = 8.6 Hz, 2H), 7.38 (d, *J* = 8.2 Hz, 2H), ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 14.38 (CH₃), 17.81 (CH₃), 21.21, (CH₃), 116.27 (CH), 126.23 (CH), 129.21 (CH), 139.15 (C), 139.224 (C), 155.45 (C), 167.02 (C), IR (neat) 2979, 1712, 1443 cm⁻¹, MS (EI) *m/z* 204 (M⁺), HRMS (EI) *m/z* (M⁺) 204.1152 (calcd for C₁₃H₁₆O₂ 204.1150)

Ethyl (*E*)-3-(4-trifluoromethylphenyl)but-2-enoate (Xc): (20 mmol scale, 4.8 g, 92%) $R_f = 0.5$ (hexane-EtOAc = 4 : 1); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.33 (t, *J* = 7.1 Hz, 3H), 2.58 (t, 1H), 4.23 (q, *J* = 7.2 Hz, 2H), 6.15 (s, 1H), 7.56 (d, *J* = 8.2 Hz, 2H), 7.63 (d, *J* = 8.4 Hz, 2H) ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 14.30 (CH₃), 17.91 (CH₃), 60.14 (CH₂), 119.48 (CH), 126.68 (CH), 128.08 (CH), 130.49 (C), 145.83 (C), 153.80 (C), 166.44 (C), IR (neat) 2984, 1718, 1636, 1445, 1120 cm⁻¹, MS (EI) *m/z* 258 (M⁺), HRMS (EI) *m/z* (M⁺) 258.0867 (calcd for C₁₃H₁₃F₃O₂ 258.0868)

Ethyl (*E*)-3-(4-nitrophenyl)but-2-enoate (Xd): (20 mmol scale, 2.7 g, 57%) $R_f = 0.5$ (hexane-EtOAc = 4 : 1); colorless crystals; mp 71.5-72.5 °C (hexane-MeOH); ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.33 (t, *J* = 7.1 Hz, 3H), 2.59 (s, 3H), 4.24 (q, *J* = 7.2 Hz, 2H), 7.27-7.64 (m, 2H), 8.22-8.26 (m, 2H), ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 14.34 (CH₃), 17.97 (CH₃), 60.36 (CH), 120.19 (CH), 123.86 (CH), 147.93 (CH), 152.76 (C), 166.17 (C), IR (KBr) 3117, 2987, 1717, 1517, 1182 cm⁻¹, MS (EI) *m/z* 235 (M⁺), HRMS (EI) *m/z* (M⁺) 235.0846 (calcd for C₁₂H₁₃NO₄ 235.0845)

Ethyl (*E*)-3-(3-nitrophenyl)but-2-enoate (Xe): (20 mmol, 3.8 g, 80%) $R_f = 0.5$ (hexane-EtOAc = 4 : 1); colorless crystals; mp 42.5-43.0 °C; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.34 (t, *J* = 7.1 Hz, 3H), 2.61 (s, 3H), 4.25 (q, *J* = 7.2 Hz, 2H), 6.21 (s, 1H), 7.57 (t, *J* = 8.0 Hz, 1H), 7.80 (d, *J* = 10.5 Hz, 1H), 8.22 (d, *J* = 11.5 Hz, 1H), 8.33 (s, 1H), ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 14.35

(CH₃), 17.88 (CH₃), 60.33 (CH₂), 119.51 (CH), 121.35 (CH), 123.36 (CH), 129.66(CH), 132.24 (CH), 143.89 (C), 148.46 (C), 152.48 (C), 166.25 (C), IR (KBr) 2990, 1715, 1631, 1353, 1183 cm⁻¹, MS (EI) *m/z* 235 (M⁺), HRMS (EI) *m/z* (M⁺) 235.0847 (calcd for C₁₂H₁₃NO₄ 235.0845)

(*E*)-3-(4-Methylphenyl) but-2-en-1-ol (Yb): (1.2 mmol scale, 169 mg, 86%) colorless oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.92 (bs, 1H), 2.04 (s, 3H), 2.3 (s, 3H), 5.47 (d, *J* = 5.5 Hz, 2H), 5.94 (t, *J* = 7.4 Hz, 1H), 7.12 (d, *J* = 7.8 Hz, 2H), 7.29 (d, *J* = 8.2 Hz, 2H), ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 15.98 (CH₃), 21.08 (CH₃), 59.88 (CH₂), 125.64 (CH), 125.68 (CH), 128.67 (CH), 137.02 (C), 137.58 (C), 139.93 (C), IR (neat) 3341, 3024, 2921, 1647, 1512, 1444 cm⁻¹, MS (EI) *m/z* 162 (M⁺), HRMS (EI) *m/z* (M⁺) 162.1049 (calcd for C₁₁H₁₄O 162.1045)

(*E*)-3-(4-Trifluoromethylphenyl)but-2-en-1-ol (Yc): (12.8 mmol scale, 2713 mg, 98%) $R_f = 0.5$ (hexane-EtOAc = 1 : 1); colorless oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 2.11 (s, 3H), 2.21 (s, 1H), 6.4 (d, J = 6.4 Hz, 2H), 6.09 (t, J = 8.3 Hz, 1H), 7.48 (t, J = 8.0 Hz, 1H), 7.72 (d, J = 10.6 Hz, 1H), 8.09 (d, J = 10.0Hz, 1H), 8.23(s, 1H), ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 15.92 (CH₃), 59.77 (CH₂), 120.90 (CH) 121.96 (CH), 129.21 (CH), 129.32 (CH), 131.71 (CH), 135.23 (C), IR (neat) 3357, 2924, 1645, 1523, 1351, 1005 cm⁻¹, MS (EI) *m/z* 216 (M⁺), HRMS (EI) *m/z* (M⁺) 216.0760 (calcd for C₁₁H₁₁F₃O 216.0762) IR (neat) 3342, 3078, 2945, 1594, 1511, 1444 cm⁻¹, MS (EI) *m/z* 193 (M⁺), HRMS (EI) *m/z* (M⁺) 193.0736 (calcd for C₁₀H₁₁NO₃ 193.0739)

(E)-3-(4-Nitrophenyl)but-2-en-1-ol (Yd): (11 mmol scale, 2154 mg, 100%) pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm), 1.72 (s, 1H), 2.11 (s, 3H), 4.42 (d, J = 7.0, 2H), 6.11 (t, J = 7.7 Hz, 1H), 7.53-7.57 (m, 2H), 8.17-8.22 (m, 2H), ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 16.00 (CH₃), 59.98 (CH₂), 123.57 (CH), 123.72 (C), 123.82 (CH), 126.49 (CH), 130.51 (CH), 135.67 (C), 146.88 (C), 149.31 (C)

(*E*)-3-(3-Nitrophenyl)but-2-en-1-ol (Ye): (2.1 mmol scale, 406 mg, 94%) $R_f = 0.3$ (hexane-EtOAc = 1 : 1); brown oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.90 (s,1H), 2.75 (s, 3H), 4.38 (d, J = 7.4 Hz, 2H), 6.02 (t, J = 8.5 Hz, 1H), 7.48 (d, J = 8.2Hz, 2H), 7.56 (d, J = 8.2Hz, 2H), ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 16.01 (CH₃), 59.91 (CH₂), 125.29 (CH), 129.45

-123-

(CH), 136.54 (C), 146.41 (C), IR (neat) 3358, 2929, 1654, 1411, 1113 cm⁻¹, MS (EI) m/z 193 (M⁺), HRMS (EI) m/z (M⁺) 193.0739 (calcd for C₁₀H₁₁NO₃ 193.0739)

The β -substituted cinnnamylamines, (*E*)-3-aryl-2-buten-1-amine (**2a-e**), (*E*)-3-aryl-3-bromo-2propen-1-amine **6** were prepared from the corresponding alcohols, (*E*)-3-arylbut-2-en-1-ol(**Ya-e**), (*E*)-3-aryl-2-buten-1-amine. The corresponding bromides were prepared by reaction of the alcohols with PBr₃ in ether and used without further purification. The β -substituted cinnnamylamines **2a-e**, and **6** were prepared by reaction of benzylamine (2 equiv) with the corresponding bromids in ether according to the literature procedure.⁹

(*E*)-*N*-Benzyl-3-phenyl-2-buten-1-amine (2a): (2.8 mmol scale, 278 mg, 42%) $R_f = 0.2$ (hexane-EtOAc = 1 : 1); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.59 (bs, 1H), 2.03 (3, 3H), 3.46 (d, J = 7.4 Hz, 2H), 3.85 (s, 2H), 5.90 (t, J = 8.6 Hz, 1H), 7.21-7.41 (m, 10H), ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 16.17 (CH₃), 47.31 (CH₂), 53.63 (CH₂), 125.76 (CH), 126.50 (CH), 127.00 (CH), 127.06 (CH), 128.28 (CH), 128.50 (CH), 136.91 (C), 140.31 (C), 143.36(C), IR (neat) 3309, 3026, 1494, 1444 cm⁻¹, MS (EI) *m/z* 237 (M⁺), HRMS (EI) *m/z* (M⁺) 237.1510 (calcd for C₁₇H₁₉N 237.1517)

(*E*)-*N*-Benzyl-3-(4-methylphenyl)-2-buten-1-amine (2b): (12 mmol scale, 1.3 mg, 43%) $R_f = 0.2$ (hexane-EtOAc = 1 : 4); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.56 (s, 1H), 2.00 (s, 3H), 2.34 (s, 3H), 3.45 (d, J = 7.2 Hz, 2H), 3.84 (s, 2H), 5.88 (t, J = 8.8 Hz, 1H), 7.12, (d, J = 8.6 Hz, 2H), 7.24-7.34 (m 7H), ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 16.15 (CH₃), 21.11 (CH₃), 47.31 (CH₂), 53.63 (CH₂), 125.62 (CH), 127.25 (CH), 128.29 (CH), 128.49(CH), 128.96 (CH), 136.69 (C), 140.34 (C), 140.45 (C), IR (neat) 3025, 2918, 1512, 1494, 1452 cm⁻¹ MS (EI) *m/z* 251 (M⁺), HRMS (EI) *m/z* (M⁺) 251.1676 (calcd for C₁₈H₂₁N 251.1674)

(*E*)-*N*-Benzyl-3-(4-trifluoromethylphenyl)-2-buten-1-amine (2c): (4.2 mmol scale, 582 mg, 45%) $R_f = 0.2$ (hexane-EtOAc = 1 : 1); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.67 (bs, 1H), , 2.02 (s, 3H), 3.46 (d, J = 6.6 Hz, 2H), 3.80 (s, 2H), 5.95 (t, J = 7.8 Hz, 1H), 7.22-7.27 (m, 1H), 7.30-7.34 (m, 4H), 7.46 (d, J = 9.6 Hz, 2H), 7.55 (d, J = 8.8 Hz, 2H), ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 16.01 (CH₃), 25.19 (CH₂), 53.58 (CH₂), 125.20 (CH), 125.24 (CH), 127.05(CH), 128.19 (CH), 128.51 (CH), 135.75 (C), 140.13(C), 146.81 (C), IR (neat) 3315, 3028, 2919, 1615, 1453 cm⁻¹, MS (EI) *m/z* 305 (M⁺), HRMS (EI) *m/z* (M⁺) 305.1387 (calcd for C₁₈H₁₈F₃N 305.1391) (*E)-N*-Benzyl-3-(4-nitrophenyl)-2-buten-1-amine (2d): (6.8 mmol scale, 1.1 g, 57%) R_f = 0.2 (hexane-EtOAc = 1 : 4); blown oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.64 (s, 1H), 2.04 (s, 1H), 3.49 (d, *J* = 7.4 Hz, 2H), 3.85 (s, 2H), 6.05 (t, *J* = 7.2), 7.25-7.28 (m, 1H), 7.31-7.36 (m, 4H), 7.49 (d-like, *J* = 9.0 Hz, 2H), 8.14 (d-like, *J* = 9.0), ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 15.87, (CH₃), 47.29 (CH₂), 53.68 (CH₂), 123.53 (CH), 126.22 (CH), 127.07 (CH), 128.15(CH), 128.32 (CH), 128.44 (CH), 130.69 (CH), 134.89 (C), 139.98 (C), 146.50 (C), 149.61 (C), IR (neat) 3027, 2840, 1593, 1513, 1453, 1344 cm⁻¹, MS (EI) *m/z* 282 (M⁺), HRMS (EI) *m/z* (M⁺) 282.1363 (calcd for C₁₇H₁₈N₂O₂ 282.1368)

(*E*)-*N*-Benzyl-3-(3-nitrophenyl)-2-buten-1-amine (2e): (10.3 mmol scale, 1.3 g, 56%) $R_f = 0.2$ (hexane-EtOAc = 1 : 1); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.69 (s, 1H), 2.50 (s, 3H), 3.48 (d, J = 6.4 Hz, 2H), 3.85 (s, 2H), 6.01 (t, J = 7.2 Hz, 1H), 7.23-7.29 (m, 1H), 7.31-7.40 (m, 2H), 7.44 (t, J = 7.9 Hz, 1H), 7.68 (d, J = 10.6 Hz, 1H), 8.05 (d, J = 10.6 Hz, 1H), 8.21 (s, 1H), ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 15.88 (CH₃), 47.21 (CH₂), 77.26 (CH₂), 120.43 (CH), 121.58 (CH), 127.02 (CH), 128.12 (CH), 128.41 (CH), 129.04 (CH), 129.27 (CH), 134.57 (CH), 134.57 (C), 140.00 (C), 144.76 (C), 148.26 (C), IR (neat) 3329, 3027, 2917, 2864, 2526, 1646, 1453 cm⁻¹, MS (EI) *m*/*z* 282 (M⁺), HRMS (EI) *m*/*z* (M⁺) 282.1359 (calcd for C₁₇H₁₈N₂O₂ 282.1368) (*E*)-3-Phenyl-3-bromo-2-propen-1-amine (6): (1.8 mmol scale, 320 mg, 69%) $R_f = 0.2$ (hexane-EtOAc = 1 : 1); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.65 (bs, 1H), 3.23 (d, J = 7.3 Hz, 2H) 3.70 (s, 1H) 6.35 (t, J = 7.1 Hz, 1H), 7.20-7.33 (m, 10H), ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 48.33 (CH₂), 53.10 (CH₂), 123.11 (C), 127.12 (CH), 128.18 (CH), 128.25 (CH), 128.34 (CH), 128.53 (CH), 128.77 (CH), 128.94 (CH), 132.72 (CH), 138.28 (CH), 139.78 (C), IR (neat) 3028, 2838, 1636 cm⁻¹, MS (CI) *m*/*z* 302 (M⁺), 304 ([M+H]⁺); HRMS (CI) *m*/*z* [M+H]⁺ 302.0541, 304.0525 (calcd for C₁₆H₁₇BrN 302.0544, 304.0524)

Typical experimental procedure for preparation of 3, 4, 5, 7, and 8 (Scheme 6-2). To a solution of 1,1-diethyl 2-hydrogen ethenetricarboxylate (1) (prepared from 1,1-diethyl 2-tert-butyl ethenetricarboxylate (272 mg, 1 mmol) upon treatment with CF₃CO₂H (4 mL))⁷ in DMF (0.8 mL) were added (*E*)-3-aryl-2-buten-1-amine (2a) (237 mg, 1 mmol) in DMF (0.8 mL), Et₃N (0.14 mL, 102 mg, 1 mmol), HOBt (1-hydroxybenzotriazole) (270 mg, 2.2 mmol), and EDCI (1-[3-(dimethylamino)- propyl]-3-ethylcarbodiimide hydrochloride) (199 mg, 1 mmol) at 0 °C. The reaction mixture was stirred for 1 h at 0 °C, and was allowed to warm to room temperature and stirred for 20 h. The reaction mixture was diluted with CH₂Cl₂. The organic phase was washed with saturated aqueous NaHCO₃ solution, 2 M aqueous citric acid, saturated aqueous NaHCO3 and water, dried (Na2SO4), and evaporated in vacuo. The residue was purified by column chromatography on silica gel eluting with hexane-EtOAc to give **3a** (214 mg, 49%). **3a**: $R_f = 0.8$ (hexane-EtOAc = 1 : 4); pale yellow viscous oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.30 (t, J = 7.1 Hz, 3H), 1.33 (t, J = 7.0 Hz, 3H), 1.33 (s, 3H), 2.85-2.89 (m, 1H), 3.04 (dd, J = 10.5, 4.0 Hz, 1H), 3.31 (dd, J = 10.4, 7.4 Hz, 1H), 3.72 (d, J = 8.2 Hz, 1H), 4.13-4.33 (m, 4H), 4.37 (d, J = 20.4 Hz, 1H), 4.47 (d, J = 14.4 Hz, 1H), 7.25-7.38 (m, 10H), ¹³C NMR (100.6 MHz. CDCl₃) δ (ppm) 14.42 (CH₃), 15.02 (CH₃), 21.29 (CH₃), 39.44 (CH), 39.79 (CH), 45.38 (CH₂), 46.71 (CH₂), 59.81 (CH₂), 64.53 (CH₂), 79.56 (C), 82.58 (C), 125.04 (CH), 127.82 (CH), 128.22 (CH), 128.73 (CH), 136.04 (C), 142.66 (C), 161.30 (C), 166.73 (C), 172.59 (C), IR (neat) 3061, 2982, 1698, 1620, 1495, 1444, 1262 cm⁻¹, MS (EI) *m/z* 435 (M⁺), HRMS (EI) *m/z* (M⁺) 435.2047 (calcd for C₂₆H₂₉NO₅ 435.2046)

4c-ax: $R_f = 0.5$ (hexane-EtOAc = 1 : 1); colorless crystals; mp 156.0-157.0 °C; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.21 (t, J = 7.0 Hz, 3H), 1.25 (d, J = 4.5 Hz, 3H), 1.27 (t, J = 8.6 Hz, 3H), 2.70-2.79 (m, 1H), 3.17-3.32 (m, 4H), 4.08-4.16 (m, 1H), 4.23-4.48 (m, 5H), 4.77 (d, J = 15.0 Hz, 1H), 7.26-7.37 (m, 7H), 7.51 (d, J = 1.4, 1H), 7.66 (s, 1H), ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 13.79 (CH₃),14.07 (CH₃), 18.53 (CH₃), 34.69 (CH), 35.43 (CH) 43.90 (CH₂) 46.55 (CH₂), 46.60 (CH₂), 62.26 (CH₂), 62.81 (CH₂), 124.80 (CH), 127.61 (CH), 127.92 (CH), 128.21 (CH),

128.73 (CH), 130.62 (CH), 134.25 (C) 136.68 (C), 146.03 (C), 157.80 (C), 167.85 (C), 170.23 (C), 171.63 (C), IR (KBr) 2983, 1745, 1685, 1437, 1258, 1116 cm⁻¹, MS (EI) m/z 503 (M⁺), HRMS (EI) m/z (M⁺) 503.1923 (calcd for C₂₇H₂₈F₃N₅O₅ 503.1920)

4d-eq: (0.5 mmol scale, DMF, 110 °C, 147 mg, 60%) $R_f = 0.2$ (hexane-EtOAc = 1 : 1); colorless crystals; mp 165.0-166.0 °C; ¹H NMR (400 MHz, CDCl₃) 1.26 (t, J = 7.1 Hz, 1H), 1.33 (d, J = 6.6 Hz, 3H), 1.39 (t, J = 7.1 Hz, 3H), 2.14-2.25 (m, 1H), 2.93-3.01 (m, 1H), 3.04-3.11 (m, 3H), 3.52 (t, J = 8.4, 1H), 4.12-4.20 (m, 1H), 4.30-4.52 (m, 5H), 4.68 (d, J = 14.8 Hz, 1H), 7.27-7.51 (m, 5H), 7.50 (d, J = 8.8 Hz, 1H), 8.13 (dd, J = 8.7, 2.4), 8.27 (d, J = 2.5 Hz, 1H), ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 13.92 (CH₃), 14.13 (CH₃), 18.77 (CH₃), 38.94 (CH), 39.15 (CH), 46.54 (CH₂), 49.45 (CH), 49.63 (CH₂), 60.97 (CH₂), 62.55 (CH₂), 63.17 (CH₂), 122.95 (CH), 125.95 (CH), 127.72 (CH), 128.23 (CH), 128.56 (CH), 128.82 (CH), 136.01 (C), 136.49 (C), 146.35 (C), 148.25 (C), 167.71 (C), 167.00 (C), 171.31 (C), IR (KBr) 3110, 3002, 1737, 1677, 1609, 1517, 1517, 1433, 1253 cm⁻¹, MS (EI) *m/z* 480 (M⁺), HRMS (EI) *m/z* (M⁺) 480.1900 (calcd for C₂₆H₂₈N₂O₇ 480.1897)

5: $R_f = 0.4$ (hexane-EtOAc = 1 : 1); colorless crystals; mp 148.0-149.0 °C; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.17 (t, J = 5.4 Hz, 3H), 1.27 (t, J = 7.1 Hz, 3H), 1.49 (s, 3H), 3.18 (t, J = 4.5 Hz, 1H), 3.33-3.38 (m, 1H), 3.49-3.53 (m, 1H), 3.66-3.71 (m, 2H), 3.86-3.94 (m, 1H), 4.06-4.23 (m, 4H), 4.72 (d, J = 14.5 Hz, 1H), 7.15 (d, J = 8.4 Hz, 1H), 7.22-7.24 (m, 2H), 7.28-7.35 (m, 4H), 7.42 (t, J = 8.1 Hz, 1H), 7.61 (t, J = 8.0 Hz, 1H), 7.92 (d, J = 8.4 Hz, 1H), 7.97 (d, J = 8.6 Hz, 1H), 8.24 (d, J = 8.6 Hz, 1H), 8.33 (s, 1H), ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 13.89 (CH₃), 13.94 (CH₃), 18.79 (CH₃), 43.58 (CH), 43.90 (CH), 46.99 (CH₂), 47.03 (CH₂), 52.18 (CH), 61.84 (CH₂), 61.94 (CH₂), 93.87 (C), 108.77 (CH), 120.32 (CH), 122.28 (CH), 124.01 (CH), 124. 74 (CH), 127.93 (CH), 128.53 (CH), 128.62 (CH), 128.76 (CH), 129.61 (CH), 133.44 (CH), 135.48 (C), 141.44 (C) 142.55 (C), 167.56 (C), 167.80 (C), 171.68 (C), IR (KBr) 2991, 1736, 1685, 1530, 1489, 1442, 1269, 1023 cm⁻¹, MS (EI) *m/z* 615 (M⁺), HRMS (EI) *m/z* (M⁺) 615.2327 (calcd for C₃₂H₃₃N₅O₈ 615.2329)

7: $R_f = 0.6$ (hexane-EtOAc = 1 : 1); pale yellow oil; ¹H NMR (400 MHz, CDCl₃) (2 rotamers, ratio 1:1) δ (ppm) 1.29, (t, J = 7.1 Hz, $3H \times 0.5$), 1.30 (t, J = 7.1 Hz, $3H \times 0.5$), 1.32 (t, J = 7.1 Hz, $3H \times 0.5$), 1.37 (t, J = 7.1 Hz, $3H \times 0.5$), 3.80 (d, J = 7.0 Hz, $2H \times 0.5$), 3.93 (d, J = 7.0 Hz, $2H \times 0.5$), 4.24-4.41 (m, $4H + 2H \times 0.5$), 4.52 (s, $2H \times 0.5$), 6.16 (t, J = 7.0 Hz, $1H \times 0.5$), 6.22 (t, J = 7.0 Hz, $1H \times 0.5$), 6.91-6.94 (m, $2H \times 0.5$), 6.99 (m, $2H \times 0.5$), 7.19-7.35 (m, 9H), ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 13.97 (CH₃), 14.04 (CH₃), 14.06 (CH₃), 14.09 (CH₃), 44.20 (CH₂), 46.49 (CH₂), 47.96 (CH₂), 62.01 (CH₂), 62.07 (CH₂), 62.23 (CH₂), 62.29 (CH₂), 125.10 (CH), 125.63 (CH), 127.23 (CH), 127.23 (CH), 127.76 (CH), 128.10 (CH), 128.17 (CH), 128.29 (CH), 128.41(CH), 128.52 (CH), 128.63 (CH), 128.73 (CH), 128.83 (CH), 128.85 (CH), 128.89 (CH), 129.04 (CH), 129.40 (CH), 134.25 (CH), 134.66 (C), 134.90 (CH), 135.16 (C), 135.37 (C), 137.22 (C), 137.61 (C), 162.80 (C), 162.89 (C), 164.16(C), 164.23 (C), 164.293 (C), 164.33(C), IR (neat) 2982, 1731, 1650, 1443, 1255, 1204 cm⁻¹, MS (EI) *m*/*z* 499, 501 (M⁺), HRMS (EI) *m*/*z* (M⁺) 499.0994, 501.0973 (calcd for C₂₅H₂₆BrNO₅ 499.0994, 501.0974)

8: (1 mmol scale, benzene, 80 °C, 242 mg, 48%) $R_f = 0.4$ (hexane-EtOAc = 1 : 1); pale yellow viscous oil; ¹H NMR (400 MHz, CDCl₃) δ (ppm) 1.13 (t, J = 7.0 Hz, 3H), 1.38 (t, J = 7.1 Hz, 3H), 3.11-3.18 (m, 1H), 3.22 (t, J = 8.4 Hz, 1H), 3.61 (d, J = 11.9 Hz, 1H), 3.66 (t, J = 8.9 Hz, 1H), 3.98-4.06 (m, 1H), 4.17-4.23 (m, 1H), 4.26 (d, J = 14.7 Hz, 1H), 4.38-4.52 (m, 1H), 4.79 (d, J = 9.8 Hz, 1H), 4.82 (d, J = 14.7 Hz, 1H), 7.25-7.36 (m, 9H), 7.68-7.71 (m, 1H), ¹³C NMR (100.6 MHz, CDCl₃) δ (ppm) 13.70 (CH₃), 13.98 (CH₃), 40.28 (CH), 46.96 (CH₂), 48.29 (CH), 52.03 (CH₂), 52.57 (CH), 60.79 (C), 62.19 (CH₂), 62.68 (CH₂), 127.48 (CH), 127.56 (CH), 128.31 (CH), 128.36 (CH), 128.45 (CH), 128.57 (CH), 128.69 (CH), 134.36 (C), 134.54 (C), 136.02 (C), 168.66 (C), 169.44 (C), 171.83 (C), IR (neat) 2980, 1731, 1640, 1445, 1260 cm⁻¹, MS (EI) *m/z* 499, 501 (M⁺), HRMS (EI) *m/z* (M⁺) 499.0990, 501.978 (calcd for C₂₅H₂₆BrNO₅ 499.0994, 501.0974)

Transformation of 7 to 8 (Table 6-5, entry 5). To a solution of **7** (483 mg, 0.97 mmol) in benzene (2 mL) was added Et₃N (0.135 mL, 99 mg, 0.97 mmol). The mixture was stirred at 80 °C for 16 h.

The reaction mixture was concentrated under reduced pressure. The residue was purified by column chromatography on silica gel eluting with hexane-Et₂O to give **8** (324 mg, 67%)

References

¹ B. B. Snider, D. M. Roush, J. Org. Chem. 1979, 44, 4229.

² (a) S. Yamazaki, K. Fujinami, Y. Maitoko, K. Ueda, K. Kakiuchi, J. Org. Chem. 2013, 78, 8405.

(b) Y. Fukushima, S. Yamazaki, A. Ogawa, Org. Biomol. Chem. 2014, 12, 3964. (c) S. Yamazaki,

J. Wada, K. Kakiuchi, Can. J. Chem. 2015, 93, 1122.

³ S. Yamazaki, H. Sugiura, M. Niina, Y. Mikata, A. Ogawa, *Heterocycles* 2016, 92, 485.

⁴ S. Yamazaki, H. Sugiura, S. Ohashi, K. Ishizuka, R. Saimu, Y. Mikata, A. Ogawa, *J. Org. Chem.* **2016**, 81, 10863

⁵ W. Oppolzer, R. Achini, E. Pfenninger, H. P. Weber, Helv. Chim. Acta 1976, 59, 1186.

⁶ H. Sugiura, S. Yamazaki, K. Go, A. Ogawa., Eur. J. Org. Chem. doi/abs/10.1002/ejoc.201801508

⁷ S. Yamazaki, K. Ohmitsu, K. Ohi, T. Otsubo, K. Moriyama, Org. Lett. 2005, 7, 759.

⁸ G. A. Pinna, G. Cignarella, G. Loriga, G. Murineddu, J.-M. Mussinu, S. Ruiu, P. Faddad, W. Fratta, *Bioorg. Med. Chem.* **2002**, 10, 1929.

⁹ J. Limberger, T. S. Claudino, A. L. Monteiro, RSC Adv. 2014, 4, 45558.

Chapter 7

Conclusion

In this research, sequential amide formation/cyclization reaction of substituent arylpropenylamine and ethenetricarboxylate was investigated.

In chapter 2, the reaction of furylamines and ethenetricarboxylate with sequential amide formation/intramolecular Diels-Alder (IMDA) reaction at room temperature, was studied. Amide intermediates were not detected.

In chapter 3, intramolecular [2 + 2] and [4 + 2] cycloaddition (IMDA) reactions of cinnamylamides and ethenetricarboxylate in sequential processes were studied. Diversity of the reaction pattern depending on the substituents of the benzene ring was formed. Reaction of cinnamylamines without substituents on the benzene ring and with halogens and OMe on para positions at room temperature gave cyclobutane-fused pyrrolidines as the major products via [2 + 21 cycloaddition. Reaction of ethenetricarboxylate and cinnamylamines bearing electron-withdrawing groups such as NO₂, CN, CO₂Me, CO₂Et, or CF₃ on ortho and para positions in the presence of EDCI/HOBt/Et₃N at room temperature or at 60-80 °C gave tetrahydrobenz[*f*]isoindolines via IMDA reaction as the major products.

In chapter 4, IMDA reactions of various 3,3-diarylpropenylamides of electron-deficient alkenes to give hexahydrobenzo[*f*]isoindoles stereoselectively were studied. Reaction of ethenetricarboxylate with 3,3-diaryl-2-propen-1-amines under the amide formation conditions gave the tricyclic compounds in sequential processes involving IMDA reaction. The reaction gave *cis*- and *trans*-fused tricyclic compounds selectively, depending on the substituents on the benzene ring, reaction temperature, and solvent. In the reaction with dissymmetrically substituted 3,3-diaryl-2-propen-1-amines, *trans*-substituted aryl group reacted mainly as a styrene component. Amides of electron-deficient alkenic carboxylic acids such as fumarate do not

undergo cyclization at room temperature sequentially and the reaction on heating gave *trans*-fused hexahydrobenzo[f]isoindoles.

In chapter 5, the reaction of ethenetricarboxylate with (heteroaryl)propenylamine was studied. Reaction of ethenetricarboxylate with (E)-3-(2-furyl)-2-propen-1-amines in the presence of EDCl/HOBt/Et₃N at 80-110 °C gave *cis*-fused tricyclic compounds as major products. On the other hand, reaction with (E)-3-(3-furyl)-2-propen-1-amines at 80-110 °C gave *trans*-fused tricyclic compounds as major products. The reaction of 3-(3-pyridinyl)-2-propen-1-amine gave HOBt-incorporated pyrrolidine diastereoselectively.

In chapter 6, the reaction of 3-aryl-2-butenylamines with ettenetricarboxylate was developed stereoselectively. Reaction of (*E*)-3-aryl-2-buten-1-amine without substituents on the benzene ring at room temperature gave cyclobutane-fused pyrrolidine as the major product via [2 + 2] cycloaddition. Reaction of (*E*)-3-aryl-2-buten-1-amine bearing *p*-NO₂ and *p*-CF₃ on benzene ring in the presence of EDCI/HOBt/Et₃N at room temperature or at 110 °C gave tetrahydrobenz-[*f*]isoindolines via IMDA reaction as the major products stereoselectively. Reaction of (*E*)-3-phenyl-3-bromo-2-propen-1-amine under the amide formation conditions at room temperature gave non-cyclized amide. The amide was transformed to *cis*-fused tricyclic compound in the presence of acid or base on heating.

In this thesis, the efficient and chemoselective synthesis of cyclized products in the reaction of electron-deficient alkenyl carboxylates such ethenetricarboxyate with various arylpropenylamines was developed. The new synthetic methods of the multicyclic compounds are expected to be useful in the development of new highly functional materials.

List of Publications

- 1. Inter- and Intramolecular Diels-Alder Reaction of Ethenetricarboxylate Derivatives
- S. Yamazaki, H. Sugiura, M. Niina, Y. Mikata, A. Ogawa, Heterocycles. 2016, 92, 3, 485

(Chapter 2)

- Intramolecular [2 + 2] and [4 + 2] Cycloaddition Reactions of Cinnamylamides of Ethenetricarboxylate in Sequential Processes
- S. Yamazaki, H. Sugiura, S. Ohashi, K. Ishizuka, R. Saimu, Y. Mikata, A. Ogawa, J. Org. Chem. 2016, 81, 22, 10863

(Chapter 3, Chapter 6)

- 3. Intramolecular Cyclization of 3,3-Diarylpropenylamides of Electron-deficient Alkenes: Stereoselective Synthesis of Functionalized Hexahydrobenzo[*f*]isoindoles
- H. Sugiura, S. Yamazaki, K. Go, A. Ogawa, *Eur. J. Org. Chem.* doi/abs/10.1002/ejoc.201801508 (Chapter 4, Chapter 6)
- 4. Sequential Intramolecular Diels-Alder Reaction of 3-Heteroaryl-2-propenylamides of Ethenetricarboxylate
- H. Sugiura, S. Yamazaki, A. Ogawa, Submitted to J. Heterocycl. Chem.

(Chapter 5, Chapter 6)

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